

# Why radium in shales and how to process

- Returned water may contain radium because organic rich shales tend to be rich in uranium
- ~20% of water will be returned
- Waste water treatment facility like the one at Niagra Falls can treat return flow
- Whole issue can be avoided by fracking with gas

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Total Organic Carbon

# Uranium increases with TOC richness of shale

Harper Pennsylv Geol 38(1) 2008 p5:  
Radioactivity = organic richness = gas

- K, Th, U adsorbed on clays (Wignall and Meyers (1988))
- Th/U~4 (Faure, 1977)
- U can be authigenic (Wignall and Meyers (1988))

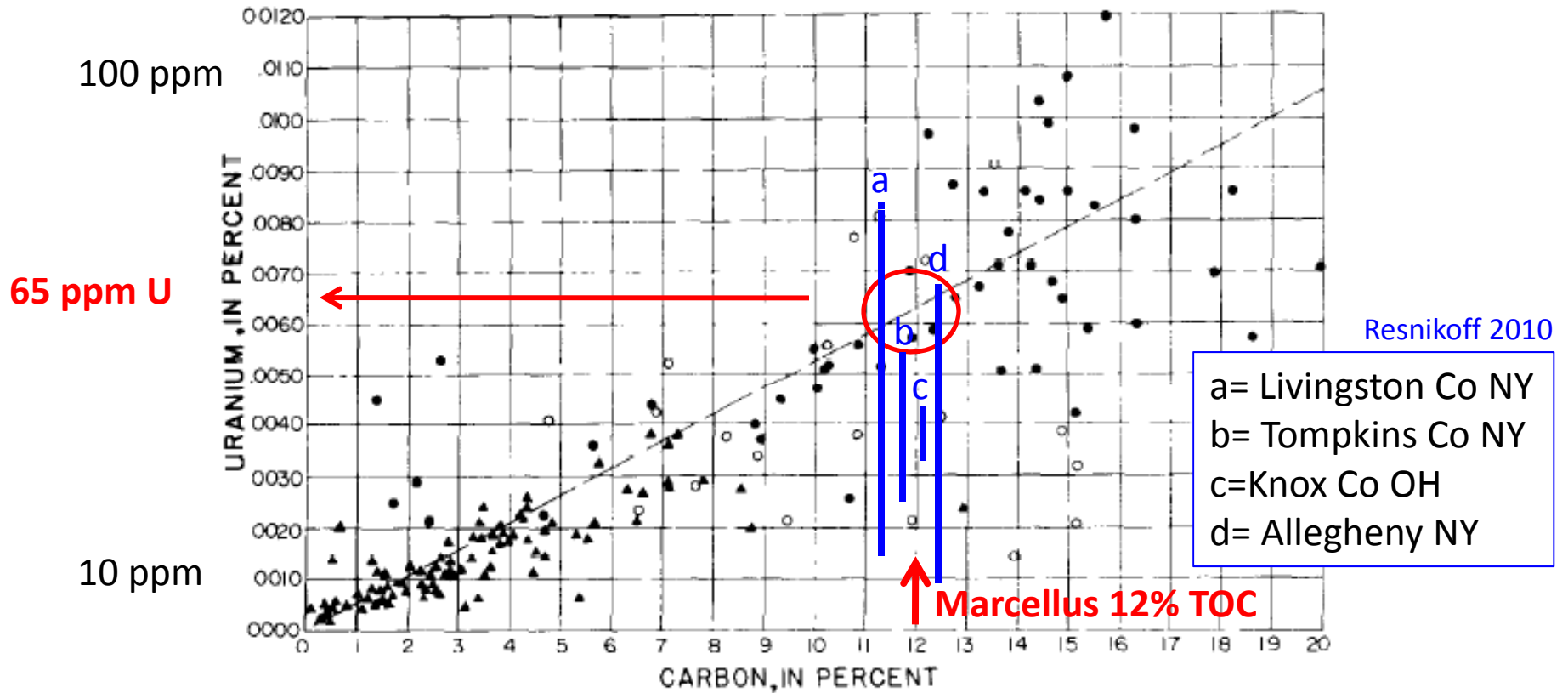
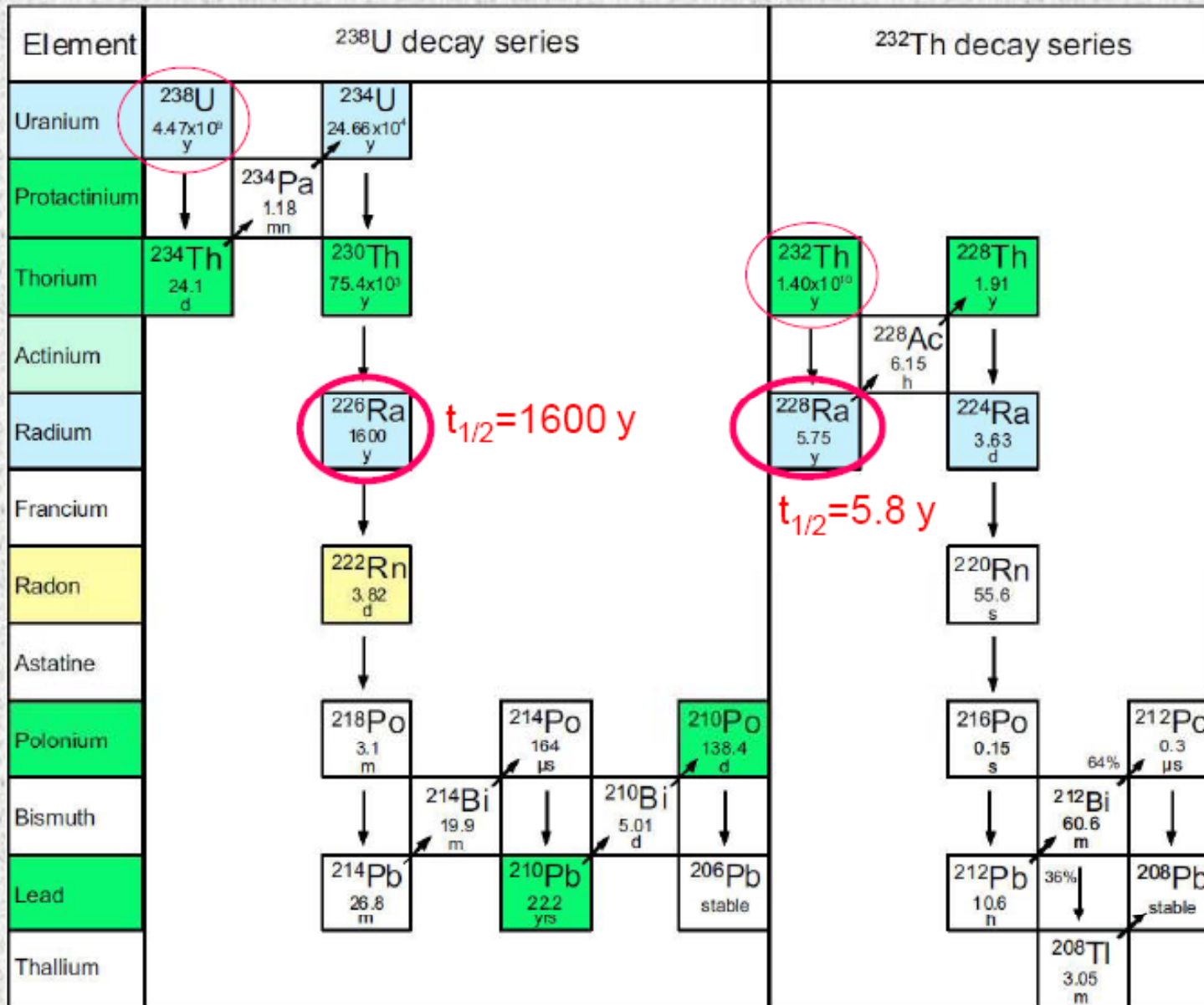


Fig. 1. Organic carbon vs uranium in Appalachian Devonian shale samples. Solid circles and line of correlation represent data from SWANSON (1960); triangles show data from LEVENTHAL and GOLDHABER (1978), and LEVENTHAL (1979). open circles show data from this study.

Leventhal JS (1981) ... organic matter and ... U..of Appalachian Devonian black shales, GCA 45 p883-889

# Uranium and Thorium Decay Chains



Th/U Ratios:

Sandstone: ~4

Black Shale: ~0.2

## World-Wide Average U and Th Content in Igneous, Metamorphic, and Sedimentary Rocks

<u>Rock Type</u>	<u>U (ppm)</u>	<u>Th (ppm)</u>	<u>Th/U</u>
Ultramafic	0.01	0.05	3.6
Basalt	0.4	1.6	4.0
Gabbro	0.8	3.8	4.7
Granite	4.8	21.5	4.5
Nepheline syenite	14	48	3.4
Granulite	1.6	7.2	4.5
Granitic gneiss	3.5	12.9	3.7
Sandstone	1.4	5.5	3.9
Shale (gray-green)	3.2	11.7	3.7
Carbonate	2.2	1.2	0.5
Shale (black carbonaceous)	8.0	1.7	0.2
Marine phosphorite	76		<1
<b>Crustal rocks (avg)</b>	<b>2.5</b>	<b>10</b>	<b>4</b>
<b>Sea water</b>	<b>0.003</b>	<b>10<sup>-5</sup></b>	<b>0.0002</b>

Th/U ratio  
low for  
black shales

Sources: Rogers and Adams (1969); Gabelman, 1977; Rose, et al (1979); Woodmansee (1975)

# $^{226}\text{Ra}$ can be calculated

$$\frac{{}^{226}\text{Ra } m}{{}^{238}\text{U } m} = \frac{{}^{226}\text{Ra } \tau_{1/2}}{{}^{238}\text{U } \tau_{1/2}} = \frac{1600 \text{ y}}{4.47 \times 10^9 \text{ y}} = 3.58 \times 10^{-7}$$

for 60 ppm U get  $2.15 \times 10^{-5}$  ppm  $^{226}\text{Ra}$

$$10^{-6} \text{ ppm } ^{226}\text{Ra} = 1 \text{ pCi} / g_{\text{rock}}$$

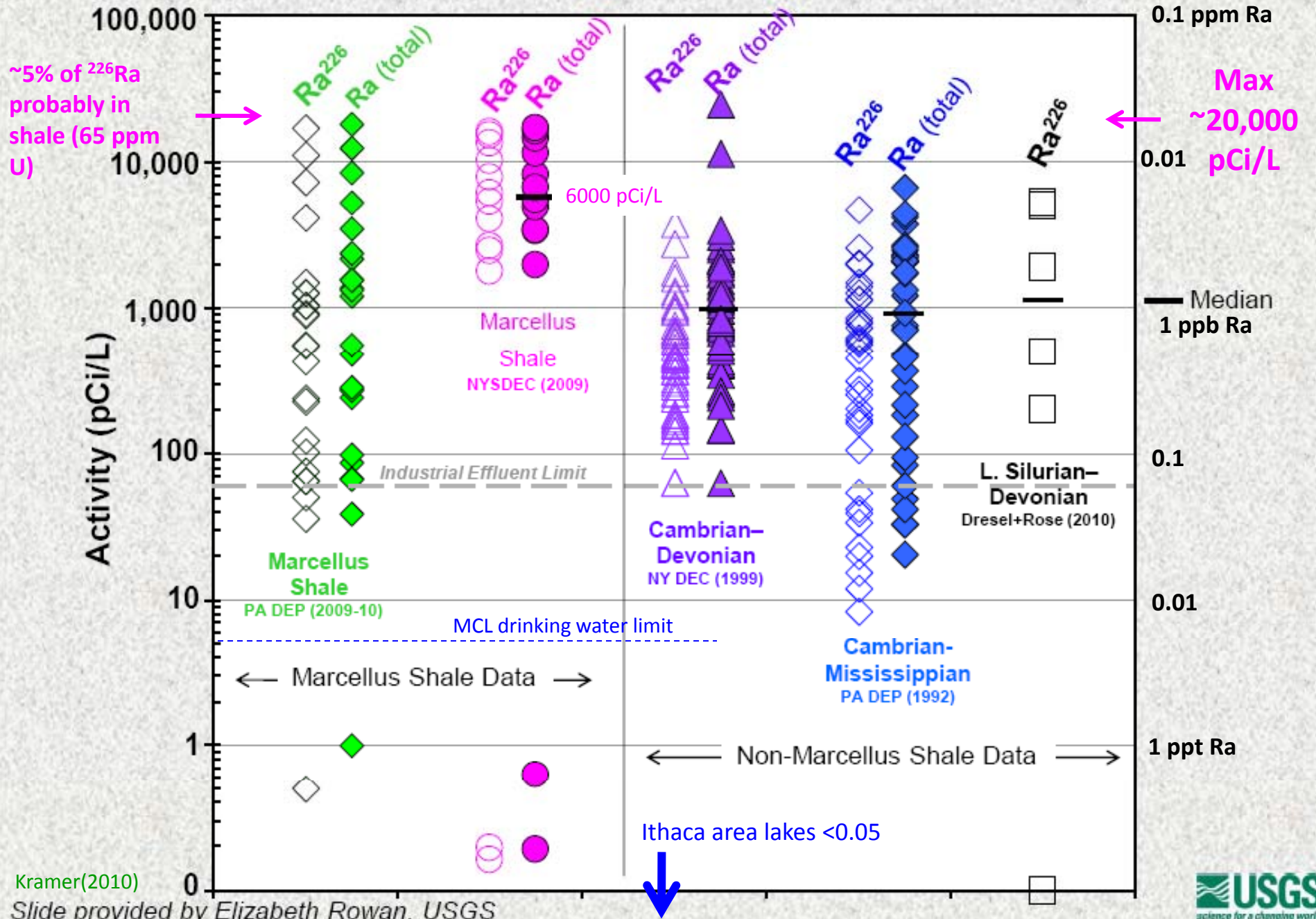
Thus 60 ppm U = 21.5 pCi/g<sub>rock</sub>

At  $2 \text{ g}_{\text{rock}}/\text{cm}^3$ , 10% porosity, 60 ppm U could put 420,000 pCi/L into pore water

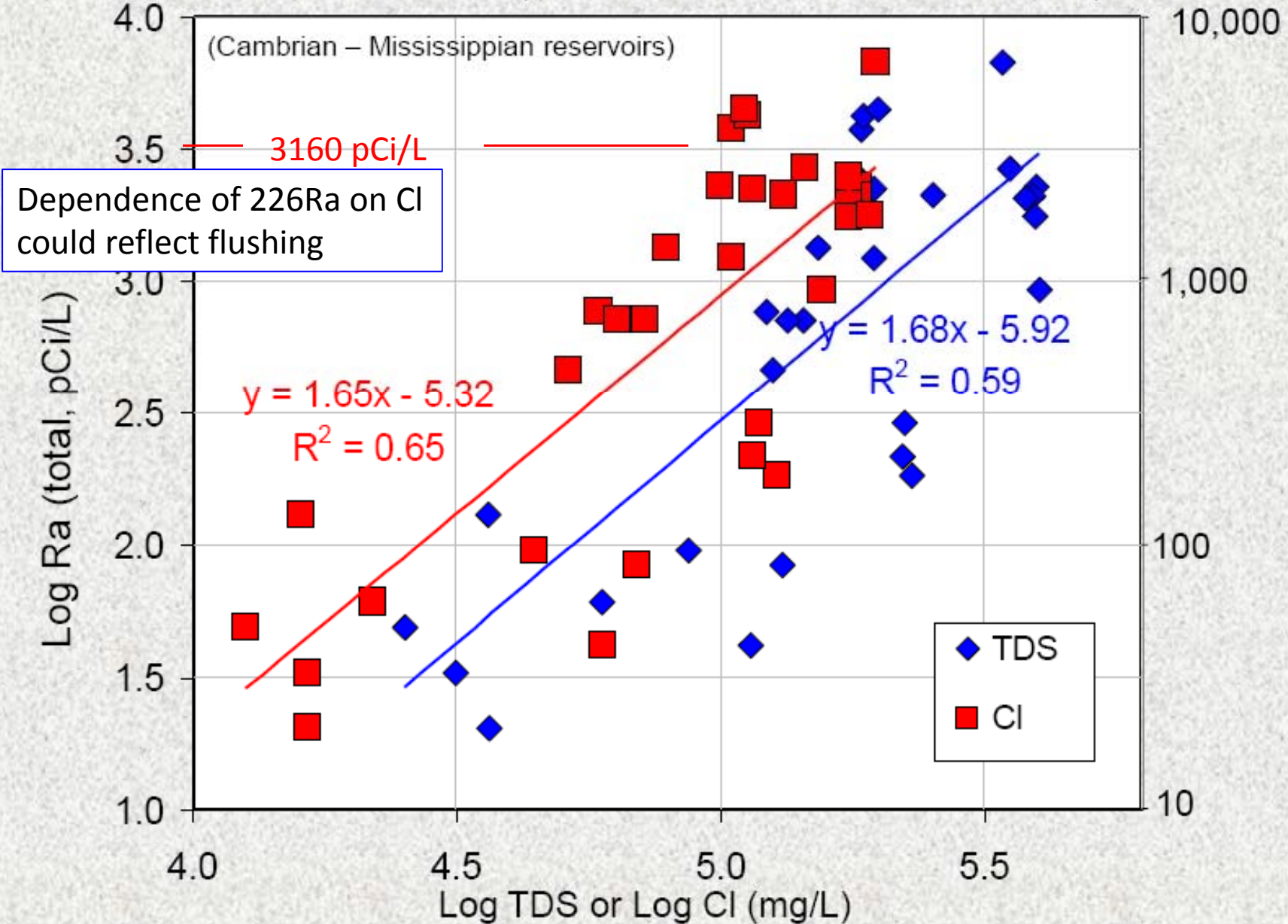
Actually see up to 5% of this loading, or 20,000 pCi/L in return fluids



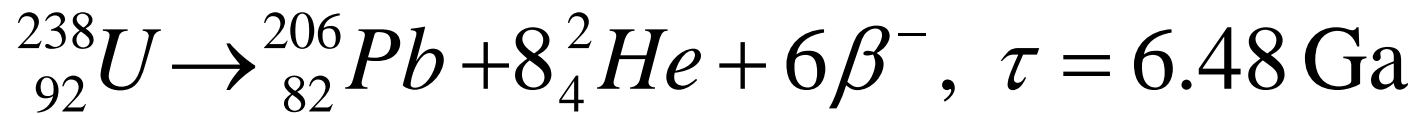
# Radium Activity Ranges in Produced Water, NY and PA



# Ra vs. Cl and TDS, Oil and Gas Well Brines, PA



# Radio decay adds $^4\text{He}$ to Marcellus gas



$$\frac{m_{\text{He}}}{m_{\text{U}}} = 6R_{\text{Th}/\text{U}} \left( \frac{t}{\tau_{\text{Th}}} \right) + 8 \left( \frac{t}{\tau_{\text{U}}} \right)$$

Powerful cross-checks are potentially available



# ppm U indicated by $^4\text{He}$ in Marcellus gas

			% gas-saturated					
RTh/U	ppm U	Time[Ma]	porosity		grain density [kg/m <sup>3</sup> ]	rock density[kg/m <sup>3</sup> ]		
0.2	60	350	10		2750	2475		
			mol wt CH <sub>4</sub>		Tsufr	grad T [K/km]	grad P bar/km	depth [km]
T1/e U [Ga]	T1/e Th [Ga]		16		20	20	100	3
6.5	20							
	mHe/mU		0.451769		moles of He generated per mole of U			
	mU/m <sup>3</sup>		0.62395		moles uranium per m <sup>3</sup> of rock			
	mHe/m <sup>3</sup>		0.281881		moles He per m <sup>3</sup> rock			
	kg He /m <sup>3</sup>		0.001128		kg He per m <sup>3</sup> rock			
	pore space		0.1		m <sup>3</sup> pore space per m <sup>3</sup> rock			
	T source [C]		80		temperature in the source shale			
	P source [bars]		301		Pressure in the source shale			
	density CH <sub>4</sub> kg/m <sup>3</sup>		178.4716		gas law density of methane in the source shale			
	kg CH <sub>4</sub> /m <sup>3</sup>		17.84716		kg CH <sub>4</sub> pre m <sup>3</sup> of source rock			
	ppm He in CH <sub>4</sub>		63.17671		ppm $^4\text{He}$ in CH <sub>4</sub> pore space gas			

200 ppm U = 210 ppm  $^4\text{He}$ , etc.

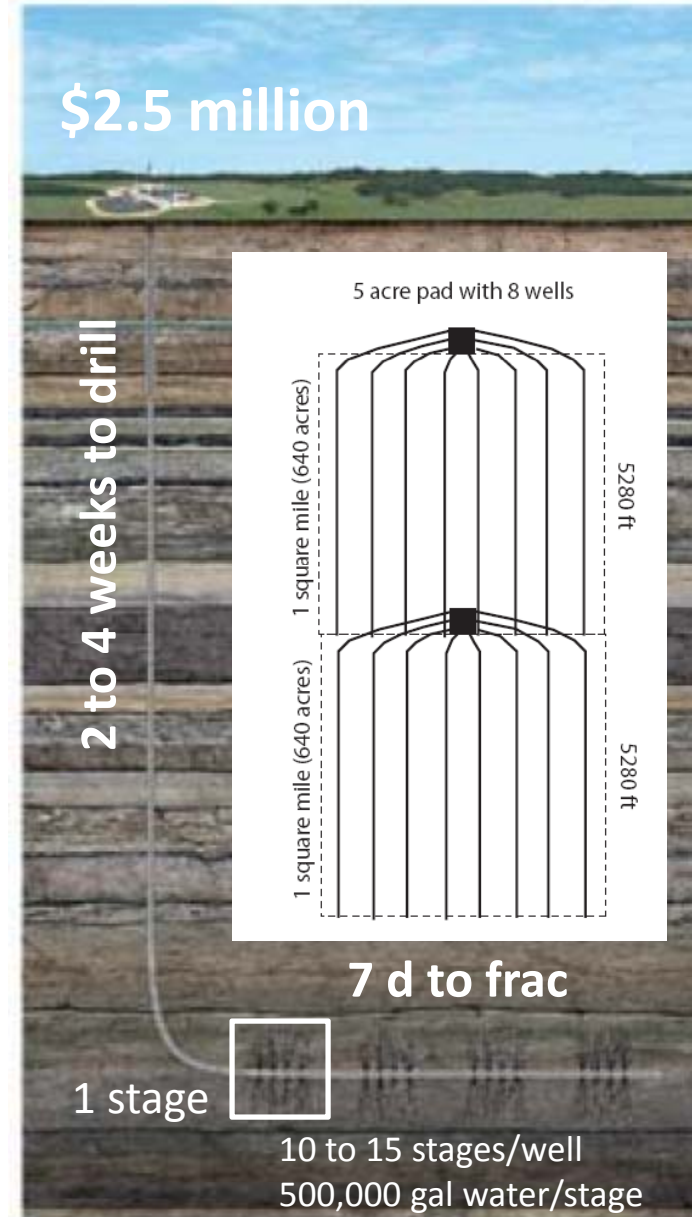
For Tompkins Co will need to process ~1 cfs return flow

## Hydrofracking

- 1 horizontal well can tap 80 acres
- 8 wells per ~5 acre pad
- $5 \times 10^6$  gal/well;  **$40 \times 10^6$  gal/pad**
- water return ~20% =  **$8 \times 10^6$  gal/pad**
- 1 pad per square mile

## Tompkins County (pop 100,153)

- 421 mi<sup>2</sup> could be drilled
- if 50% developed over 10 years with 1 pad/mi<sup>2</sup>
  - 21 pads/yr
  - 210 wells /yr
  - 2,500 jobs (10 p/well) ~4% TC workforce
  - $10^9$  gallons of water /yr = 5 cfs
  - **1 cfs return water/yr**

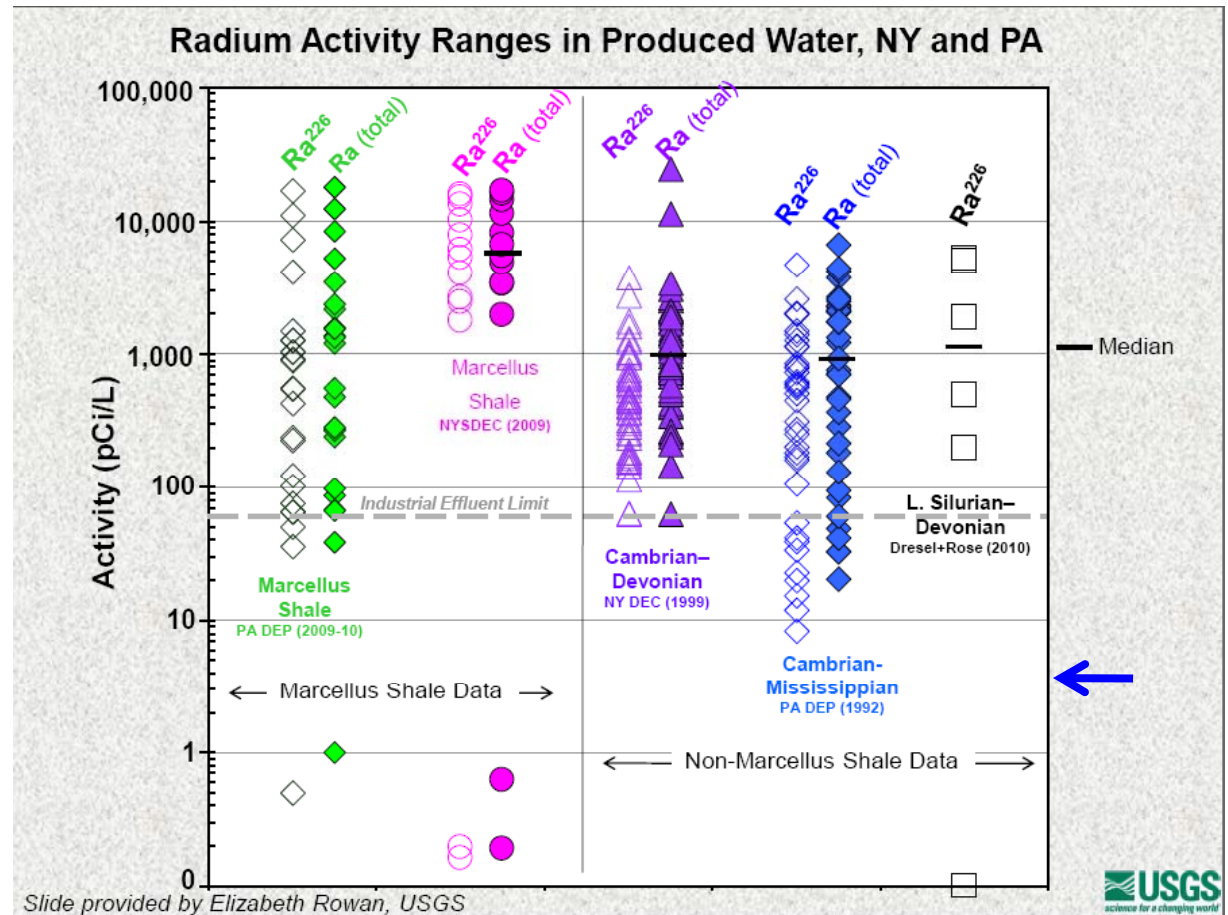


1 cubic ft = 7.48 gallons  
1cfs =  $0.24 \times 10^9$  gallons per year

Pad development scenario is from a presentation by Art Pierce in Lansing  
June 29, 2011

# Fracking chemicals less serious than dissolved Minerals like Ra

Radium is an issue that must be managed



- Managed at existing wells
- Can be detected and precipitated at well head
- Can be removed in drinking water treatment facilities

5 pCi/L = USEPA drinking water standard

Environmental Levels of Radium in Water of Central New York, Thomas F. Kraemer U.S. Geological Survey, Reston, VA, Finger Lakes Research Conference, December 4, 2010

# Return water can be treated

## Niagra Falls Wastewater Treatment

- Screen for large solids
- Remove grit in settling ponds
- flocculate with  $\text{FeCl}_3$  and polymer -> thickeners , belt dewater, disposal
- Flow through carbon beds
- Treat with peroxide and Na hyperchlorite
- dispose Niagra River
- capacity 136 ML/d = 55 cfs

Not same as drinking water treatment

- 21 pads generate 1 cfs return water
- Niagra plant could treat 1100 pads per year



Niagara Falls Wastewater Treatment Plant High Rate Treatment, Canada



<http://nfwb.org/customer/faq.php>

<http://www.water.mottmac.com/waterprojects/?mode=type&id=327912>



# Propane fracking eliminates water problems, cuts trucks by 5x, and does not impair resource

- no water injected
- no capillary seals
- no contaminants returned (Ra, metals)
- recycle 80% propane
- propane uniform chemistry → gelation chemistry simple (P-ester, Fe+++ SO<sub>4</sub> linker, MnO breaker)
- no flaring at startup
- Fewer trucks (30 vs 947)
- Fewer frac jobs (because all fractures good)
- Lower cost because more effective



Technological innovation can address issues

Robert Lestz, Gasfrac Energy Services Inc  
Cornell Lectures March 1 and 2, 2010

# References

Faure G (1977) Principles of Isotope Geochemistry, John Wiley, New York, 464 p.

Kramer TA (2010) Environmental levels of radium in water of central New York, Lecture to Finger Lakes Research Conference, Dec. 4.

Leventhal JS (1981) Pyrolysis gas chromatography- mass spectrometry to characterize organic matter and its relationship to uranium content of Appalachian Devonian black shales, Geochim Cosmochim Acta 45 p 883-889.

Resnikoff M (2010) Radioactivity in Marcellus Shale, Report prepared for Residents for the Preservation of Lowman and Chemung (RFPLC), <http://www.rwma.com/Marcellus%20Shale%20Report%205-18-2010.pdf>

Wignall PB and Myers KJ (1988) Interpreting benthic oxygen levels in mudrocks: a new approach, Geology 16 p 452-455.