



### New Architecture for Space Solar Power Systems: Fabrication of Silicon Solar Cells Using In-Situ Resources

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#### **Lunar Power Requirements**

- Initial: 100kW to 1 MW
- Non-Nuclear/ Non-Mechanical
- Solar Cells
  - Current technology: 300 W to 500 W/kg
  - From 300 to 3000kg to transport lightweight cells
  - High costs
- ⇒ Manufacture Solar Cells Directly on the Moon by Utilizing
  In Situ Resources

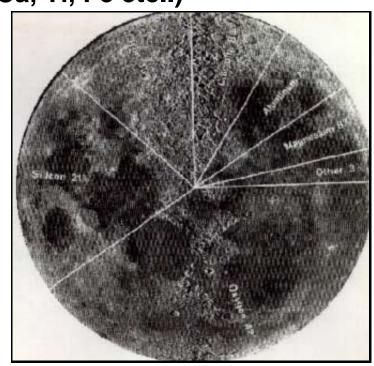


# Production of Solar Cells on the Surface of the Moon from Lunar Regolith



Past Interest in Lunar Resource Utilization: Extraction of Oxygen (waste by product Si, Al, Ca, Ti, Fe etc..)

- Elements Required for Si-based Solar Cells are Present on the Moon
  - Silicon
  - Iron
  - Titanium Oxide
  - Calcium
  - Aluminum

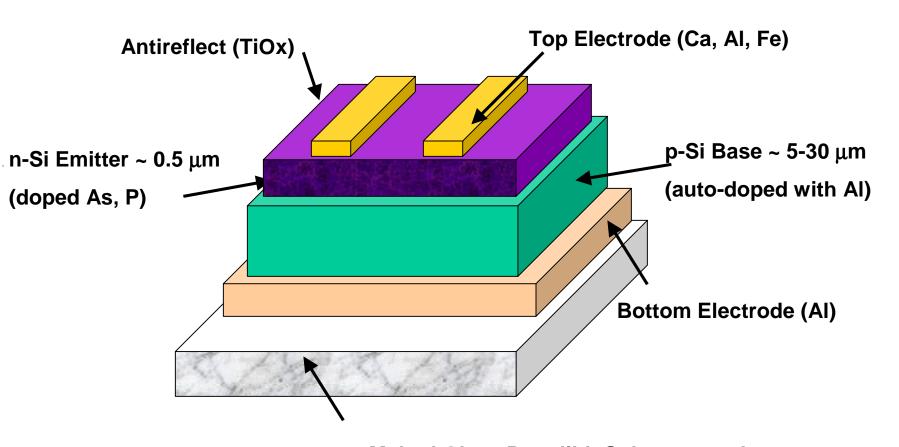


- . Moon's Surface is an Ultra-High Vacuum
  - $\sim 10^{-10} \, \text{Torr (day)}$
  - Use vacuum evaporation to make thin film solar cells



#### **Lunar Silicon Solar Cell**





Melted Glass Regolilth Substrate and Bottom Electrode



## Lunar vs Terrestrial Solar cell Materials



Terrestrial Material	Purpose	<b>Lunar Material</b>	
Glass/Silicon	Substrate	Lunar Melted Glass	
Silicon	Solar Cell Absorber	Silicon	
Aluminum	Back Contact	Aluminum	
Silver	Front Contact	Aluminum or Calcium	
TiO2	Anti-Reflection Coating	TiO <sub>2</sub> , Phosglass, or Geikeilites (TiMgO mineral)	
Boron	P-Type Dopant	Aluminum	
Phosphorous	N-Type Dopant	Phosphorous	
Copper	Cell Interconnects	Aluminum or Calcium	





## Typical compositions of lunar mare regolith, anorthite, ilmenite and pyroxene (weight %)

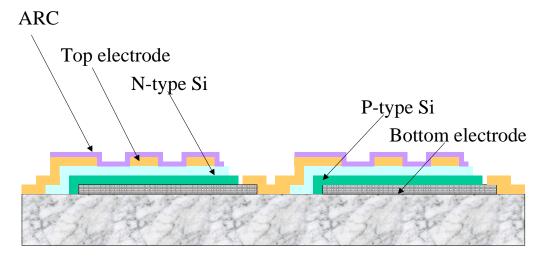
	Apollo 15	Ilmenite	Anorthite	Low-Calcium
	Regolith	$(FeTiO_3)$	$(CaAl_2Si_2O_8)$	pyroxene
				(Ca,Mg,Fe)SiO <sub>3</sub>
$SiO_2$	46.7	0.1	44.2	52.4
TiO <sub>2</sub>	1.7	52.2		0.4
$Al_2O_3$	13.2		35.8	1.89
$Cr_2O_3$	0.4	0.5		1.0
FeO	16.3	44.4	0.2	16.9
MnO	0.2	0.2		
MgO	10.9	1.4		23.9
CaO	10.4	0.2	19.7	2.6
Na <sub>2</sub> O	0.4		0.2	
K <sub>2</sub> O	0.2			
$P_2O_5$	0.2			
S	0.1			
Total	100.7	99.0	100.0	99.14



#### **Fabrication of Silicon Solar Cells**



- Use lunar materials (Si, Fe, TiO<sub>2</sub>, etc.)
- Lunar 'glass' substrate melt regolith by solar heat
- Deposit polycrystalline silicon solar cells by solar evaporation
- Interconnect solar cells serially for ~50-100V
- Do cell fabrication robotically

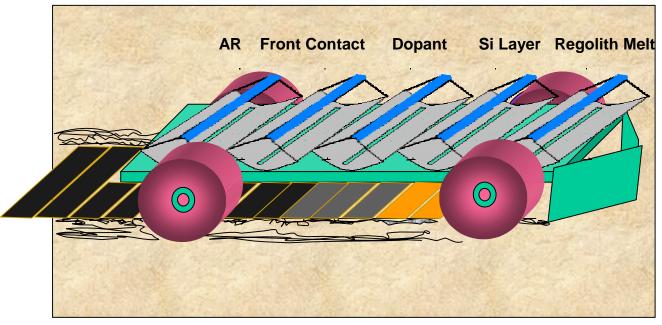


Lunar Solar Cell Monolithic Interconnection



## Mechanized Solar Cell Growth Facility - Rover





- Solar powered

**Solar Electric Motion** 

**Solar Thermal Evaporation** 

- Continuous lay-out of cells on lunar surface
- Remotely controlled



## Mechanized Regolith Processing Rover/Crawler



- Regolith scoop
- Solar thermal/ electric heat
- Regolith processing flow
- Closed-cycle processing
- Recycle volatiles
- Feed solar cell production rover(s)



### **Phase I objectives**

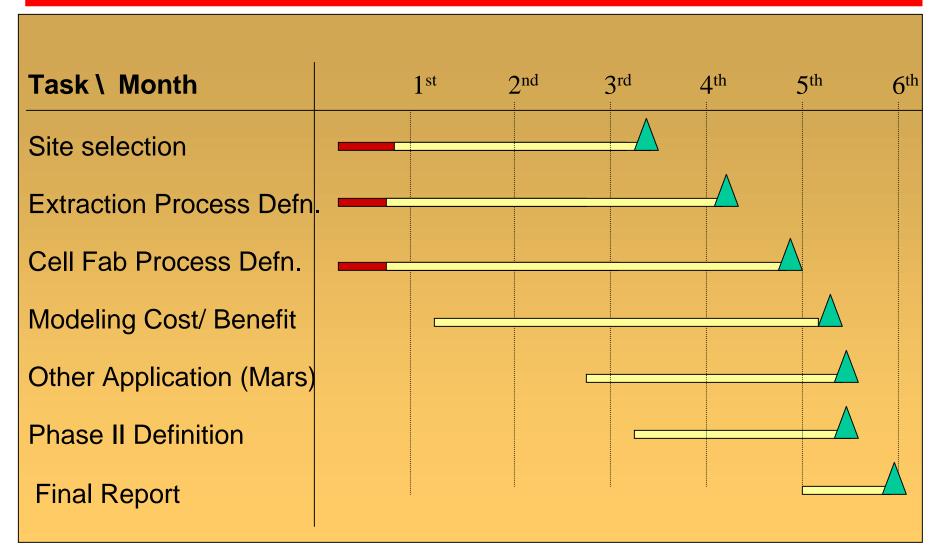


- 1-Selection of typical locations for solar cell production on the Moon
- 2-. Definition of available raw materials on the Moon and definition of processes for extraction of solar cell materials
- **3-Definition** of production process for solar cell arrays on the surface of the Moon
- 4-Performance modeling for assessing cost/benefit ratio for the power system concept
- 5-Evaluate extension of the concept beyond the Moon (e.g. Mars)
- 6-Identification of key Tasks for Phase II of the program



#### Phase I tasks/objectives







#### **Lunar Site selection**



#### Direct Fabrication on the Surface in Equatorial Regions

- Front side of Moon
  - . Lunar Base
  - . He³ mining
  - . Tourism
- Back Side
  - . Radio astronomy

#### **South Pole - Circumferentially Covered Mountain Peak**

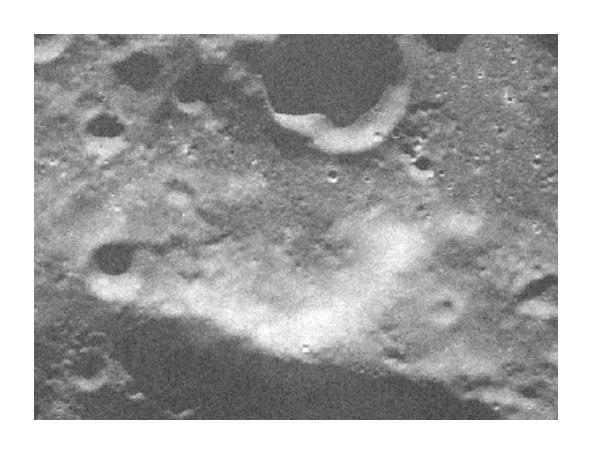
- Continual sunlight
  - . Earth view (power beaming)
  - . Possible water mining

Site Selection: Mineral composition and surface topography will impose a Solar cell design /Solar Cell Layout



### Solar Cells on the Surface of the Moon **SV** from Lunar Regolith





This image was prepared by J-L Margot, Cornell University, from Earth-based radar measurements of the Moon. The illuminated mountain at the bottom of the picture is located at 0°, 85°S. This is artificially illuminated, but approximates the illumination at noon on the Moon. The dark side never views the Earth and much of it is in permanent shadow. The scale of the picture is 50x70km. The illuminated peak is about 150 km from the South Pole and is ~6km above the plains to the north.



#### **Solar Cell Material Extraction Process (1)**



#### e.g. Carbothermal Reduction of Anorthite

Step 1.  $4 CH_4 -----> 4 C + 8 H_2$ 

$$1650^{\circ}$$
C Step 2. CaAl<sub>2</sub>Si<sub>2</sub>O<sub>8</sub> + 4 C -----> CaO + Al<sub>2</sub>O<sub>3</sub> + 2 Si + 4 CO

(anorthite) m.p. 1521°C

Ni Catalyst 250°C

Step 3.  $4 \text{ CO} + 12 \text{ H}_2$  ----->  $4 \text{ CH}_4 + 4 \text{ H}_2\text{O}$ 

 $75^{\circ}$ C Step 4. 4 H<sub>2</sub>O + electrolysis ----> 4 H<sub>2</sub> + 2 O<sub>2</sub>

⇒ Closed cyclic process yielding both OXYGEN and SILICON:

$$CaAl_2Si_2O_8$$
 ----->  $CaO + Al_2O_3 + 2 Si + 2 O_2$ 

## Solar Cell Material Extraction Process (2) SV



#### **Ilmentite Reduction (Hydrogen or Carbon)**

⇒ Yields iron for interconnect and TiO<sub>2</sub> for antireflect



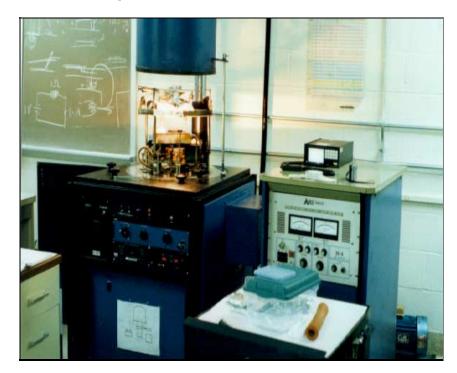
#### **Lunar Si Thin Film Evaporation**



- Extracted Silicon is much poorer than electronics-grade (doubtful for solar cell fabrication)

Vacuum evaporation of lunar simulant-extracted silicon\*

- Lab test
- 10<sup>-7</sup> Torr
- E-beam evaporation



Find Vacuum Purification

Silicon obtained by electrolysis and solidification from a hyperutectic Si- Al alloy (R. Keller)



### **Vacuum Evaporation/ Purification**



## Impurity Levels Measured in Starting Lunar-Si and Si Films Deposited on Al Foil

Impurities	Impurity extracted regolith*	in Si lunar	Impurity measured by SIMS on regolith films deposited on 1 mil-thick Al-foil
Aluminum (Al)	240 PPM		Non conclusive due to (substrate)
Calcium (Ca)	175 PPM		< 1 PPM
Lithium (Li)	31 PPM		< 1 PPM
Copper (Cu)	20 PPM		< 1 PPM
Iron (Fe)	18 PPM		< 1 PPM

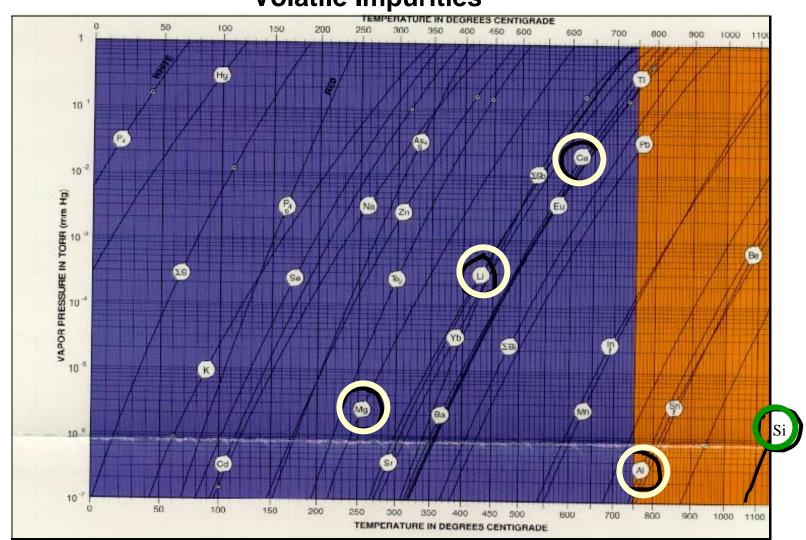
<sup>(\*)</sup> As measured by R. Keller from EMC Consultants.



# Space Vacuum Epitaxy Center University of Houston NASA Commercial Space Center



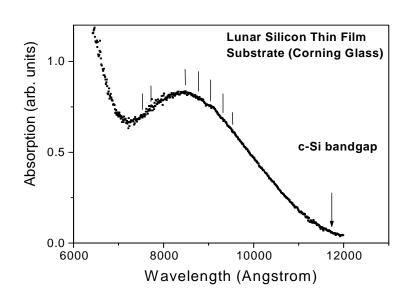
#### Volatile Impurities

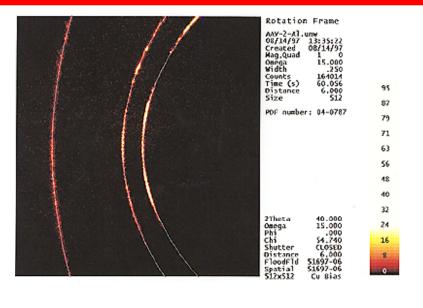




# Lunar Simulant Extracted Silicon Film Optical/structural Properties







- Bandgap ~ 1.1 eV
- Index of Refraction ~ 3.5
- •Conductivity p-type (10<sup>18</sup> cm<sup>-3</sup>)

- X-ray Diffraction
  - Films on Glass are nano-crystalline

Preliminary film properties suggests Si Solar cell efficiency ~6-9%



### Projected Cell production rate /unit



- ~ 10-100  $\mu$ m/hr evaporation rate (30- 300 m² in 1 lunar day)
- For 5-10% Efficiency Cells
  - ~ 5-50 kW/lunar day
  - ~ 50-500 kW/earth year
- Continuous Cell Replacement
  - Assume limited cell lifetime
    - . Radiation damage
    - . Particle damage





#### **Summary**

- Investigation of a new architecture for the development of solar cells using planetary (lunar) resources is proposed
- Ultra-high Vacuum on Lunar Surface Allows for Direct Thin Film Solar Cell Production
  - Less Mass to the Moon
  - Lunar Resources can be Utilized for Cell Production
  - Trade-off Cell Efficiency with Quantity
  - Multiple Facilities can be Utilized
  - Move to Industrial Scale Power Generation and Power Grid on the Moon





## Production of Solar Cells on the Surface of the Moon from Lunar Regolith

- Solar Concentrator to Melted Regolith for 'Glass' Substrate
  - Low thermal conductivity
  - 1 m<sup>2</sup> collector should yield ~ 1 2 mm thickness of melted regolith 'glass'
  - Dope ilmenite w/Ca, Mg, Al for higher conductivity - bottom electrode?
  - Continuously form glass substrate/bottom electrode as robotic rover moves





### **Lunar Solar Cell Components**

Layer	Thickness	Type	Source	<b>Indigenous Minerals</b>	Fabrication
	Range		Materials		Technique
Top electrode	0.1-1	Metallic Ca,	Lunar Ca,	Yes: Anorthite,	Vacuum thermal
	micron	Ca/Fe, or Al	CaAl, Fe	Ilmenite	evaporation
Antireflection	0.1-0.2	TiO <sub>2</sub> or SiO <sub>x</sub> or	Regolith	Yes :from Ilmenite	Vacuum Thermal
Coating	micron	$AlO_x$		(FeTiO <sub>3</sub> ) Anorthite	Evaporation
				$(CaAl_2Si_2O_8)$	
N-type Si	0.1-0.3	Si doped with	Lunar Si	Yes: (Anorthite)	Co-evaporation of
	micron	As, P, or S	N-dopant	PO <sub>x</sub> and S are present	Si and dopant
		(doping about		in low quantities in	
		100-200PPM)		minerals	
P- type Si	1-10	Si doped with Al	Lunar Si	Anorthite	Vacuum thermal
	micron	(20-50 PPM)			evaporation
Bottom	1 -2 micron	Al, Ca/Fe	Lunar Al	Anorthite	Vacuum thermal
Electrode					evaporation
Substrate	2-5 mm	Glass mineral	Lunar soil	Yes	Solar thermal
					melting





- Silicon p/n juncton ( solar evaporation)
  - ~10 to 40μm thick
  - p-doping with Al
  - n-doping with As or P (supplied terrestrially)
- Metallization
  - Bottom Electrode (AI)
  - Top Electrode (Ca /Al or Fe)
  - Evaporation through contact mask(grid pattern)
- Anti-reflection Coating
  - TiO<sub>2</sub> or SiO<sub>2</sub>