

The potential of CO₂-based production cycles in biotechnology to fight the climate crisis

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Supplementary Table 1. Calvin-Benson-Bassham (CBB) cycle reactions used for the calculation of the minimal metabolic driving force by eQuilibrator.

Reaction	Relative flux
CO ₂ + ribulose 1,5-bisphosphate + H ₂ O \rightleftharpoons 2 3-phospho d-glycerate	3
ATP + 3-phospho d-glycerate \rightleftharpoons ADP + Glycerate-1,3-Bisphosphate	6
Glycerate-1,3-Bisphosphate + nadph \rightleftharpoons D-glyceraldehyde 3-phosphate + Pi + nadp	6
d-glyceraldehyde 3-phosphate \rightleftharpoons glycerone phosphate	2
Glycerone phosphate + D-Glyceraldehyde 3-phosphate \rightleftharpoons D-fructose 1,6-bisphosphate	1
D-fructose 1,6-bisphosphate + H ₂ O \rightleftharpoons D-fructose 6-phosphate + Pi	1
D-fructose 6-phosphate + D-glyceraldehyde 3-phosphate \rightleftharpoons xylulose 5-phosphate + erythrose 4-phosphate	1
Erythrose 4-phosphate + glycerone phosphate \rightleftharpoons Sedoheptulose 1,7-bisphosphate	1
Sedoheptulose 1,7-bisphosphate + H ₂ O \rightleftharpoons sedoheptulose 7-phosphate + Pi	1
sedoheptulose 7-phosphate + D-glyceraldehyde 3-phosphate \rightleftharpoons ribose 5-phosphate + xylulose 5-phosphate	1
D-ribose 5-phosphate \rightleftharpoons D-ribulose 5-phosphate	1
D-xylulose 5-phosphate \rightleftharpoons D-ribulose 5-phosphate	2
D-ribulose 5-phosphate + ATP \rightleftharpoons ribulose 1,5-bisphosphate + ADP	3

Supplementary Table 2. Reductive tricarboxylic acid cycle (rTCA) reactions used for the calculation of the minimal metabolic driving force by eQuilibrator.

Reaction	Relative flux
oxaloacetate + NADH <=> malate + NAD+	1
malate <=> fumarate + H ₂ O	1
fumarate + FADH ₂ <=> succinate + FAD	1
succinate + Coenzyme A + ATP <=> succinyl-CoA + ADP + Pi	1
succinyl-CoA + CO ₂ + NADH <=> alpha-ketoglutarate + Coenzyme A + NAD+	1
alpha-ketoglutarate + CO ₂ + NADH <=> isocitrate + NAD+	1
Isocitrate <=> cis-aconitate + H ₂ O	1
cis-aconitate + H ₂ O <=> Citrate	1
citrate + ATP + Coenzyme A <=> AcetylCoA + oxaloacetate + ADP + Pi	1

Supplementary Table 3. Wood Ljungdahl pathway reactions used for the calculation of the minimal metabolic driving force by eQuilibrator.

Reaction	Relative flux
$\text{CO}_2 + \text{NADPH} \rightleftharpoons \text{Formate} + \text{NADP}^+$	1
$\text{Tetrahydrofolate} + \text{Formate} + \text{ATP} \rightleftharpoons \text{ADP} + \text{Pi} + 10\text{-Formyltetrahydrofolate}$	1
$10\text{-Formyltetrahydrofolate} \rightleftharpoons 5,10\text{-Methenyltetrahydrofolate} + \text{H}_2\text{O}$	1
$5,10\text{-Methenyltetrahydrofolate} + \text{NADPH} \rightleftharpoons 5,10\text{-Methylenetetrahydrofolate} + \text{NADP}^+$	1
$5,10\text{-Methylenetetrahydrofolate} + \text{NADH} \rightleftharpoons 5\text{-Methyltetrahydrofolate} + \text{NAD}^+$	1
$\text{CO}_2 + \text{H}_2 \rightleftharpoons \text{Carbon monoxide} + \text{H}_2\text{O}$	1
$5\text{-methyltetrahydrofolate} + \text{Coenzyme A} + \text{Carbon monoxide} \rightleftharpoons \text{Acetyl-CoA} + \text{Tetrahydrofolate}$	1
$\text{H}_2 + \text{NAD}^+ \rightleftharpoons 2\text{H}^+ + \text{NADH}$	1
$\text{H}_2 + \text{NADP}^+ \rightleftharpoons 2\text{H}^+ + \text{NADPH}$	2

Supplementary Table 4. Reductive glycine (rGlycine) pathway reactions used for the calculation of the minimal metabolic driving force by eQuilibrator.

Reaction	Relative flux
Tetrahydrofolate + Formate + ATP \rightleftharpoons ADP + Pi + 10-Formyltetrahydrofolate	2
10-Formyltetrahydrofolate \rightleftharpoons 5,10-Methenyltetrahydrofolate + H ₂ O	2
5,10-Methenyltetrahydrofolate + NADPH \rightleftharpoons 5,10-Methylenetetrahydrofolate + NADP ⁺	2
5,10-Methylenetetrahydrofolate + NH ₃ + CO ₂ + NADH \rightleftharpoons Glycine + Tetrahydrofolate + NAD ⁺	1
Glycine + 5,10-Methylenetetrahydrofolate + H ₂ O \rightleftharpoons serine + tetrahydrofolate	1
serine \rightleftharpoons pyruvate + NH ₃	1

Supplementary Table 5. Serine cycle reactions used for the calculation of the minimal metabolic driving force by eQuilibrator.

Reaction	Relative flux
Methanol + Ubiquinone \rightleftharpoons Formaldehyde + Ubiquinol	1
Formaldehyde + H ₂ O + NAD ⁺ \rightleftharpoons Formate + NADH	1
Tetrahydrofolate + Formate + ATP \rightleftharpoons ADP + Pi + 10-Formyltetrahydrofolate	1
10-Formyltetrahydrofolate \rightleftharpoons 5,10-Methenyltetrahydrofolate + H ₂ O	1
5,10-Methenyltetrahydrofolate + NADPH \rightleftharpoons 5,10-Methylenetetrahydrofolate + NADP ⁺	1
5,10-Methylenetetrahydrofolate + Glycine + H ₂ O \rightleftharpoons Tetrahydrofolate + L-Serine	1
L-Serine + Glyoxylate \rightleftharpoons Hydroxypyruvate + Glycine	1
Hydroxypyruvate + NADH \rightleftharpoons Glycerate + NAD ⁺	1
Glycerate + ATP \rightleftharpoons 3-Phosphoglycerate + ADP	1
3-Phosphoglycerate \rightleftharpoons Phosphoenolpyruvate + H ₂ O	1
Phosphoenolpyruvate + Carbon dioxide + H ₂ O \rightleftharpoons Oxaloacetate + Pi	1
Oxaloacetate + NADH \rightleftharpoons Malate + NAD ⁺	1
ATP + Malate + Coenzyme A \rightleftharpoons ADP + Pi + Malyl-CoA	1
Malyl-CoA \rightleftharpoons Acetyl-CoA + Glyoxylate	1

Supplementary Table 6. Xylulose monophosphate (XuMP) cycle reactions used for the calculation of the minimal metabolic driving force by eQuilibrator.

Reaction	Relative flux
methanol + O2 <=> formaldehyde + H2O2	3
formaldehyde + D-xylulose 5-phosphate <=> Glycerone + D-Glyceraldehyde 3-phosphate	3
glycerone + ATP <=> glycerone phosphate + ADP	3
Glycerone phosphate + D-Glyceraldehyde 3-phosphate <=> d-fructose 1,6-bisphosphate	1
D-fructose 1,6-bisphosphate + H2O <=> D-fructose 6-phosphate + Pi	1
D-fructose 6-phosphate + D-glyceraldehyde 3-phosphate <=> xylulose 5-phosphate + erythrose 4-phosphate	1
Erythrose 4-phosphate + glycerone phosphate <=> Sedoheptulose 1,7-bisphosphate	1
Sedoheptulose 1,7-bisphosphate + H2O <=> sedoheptulose 7-phosphate + Pi	1
sedoheptulose 7-phosphate + D-glyceraldehyde 3-phosphate <=> ribose 5-phosphate + xylulose 5-phosphate	1
ribose 5-phosphate <=> xylulose 5-phosphate	1
Glycerone phosphate <=> D-Glyceraldehyde 3-phosphate	1

Supplementary Table 7. Xylulose monophosphate (XuMP) cycle reactions without sedoheptulose-1,7-bisphosphatase (SBP) to calculate the minimal metabolic driving force by eQuilibrator.

Reaction	Relative flux
methanol + O2 <=> formaldehyde + H2O2	3
formaldehyde + D-xylulose 5-phosphate <=> Glycerone + D-Glyceraldehyde 3-phosphate	3
glycerone + ATP <=> glycerone phosphate + ADP	3
Glycerone phosphate + D-Glyceraldehyde 3-phosphate <=> d-fructose 1,6-bisphosphate	2
D-fructose 1,6-bisphosphate + H2O <=> D-fructose 6-phosphate + Pi	2
D-fructose 6-phosphate + D-glyceraldehyde 3-phosphate <=> xylulose 5-phosphate + erythrose 4-phosphate	1
D-fructose 6-phosphate + erythrose 4-phosphate <=> sedoheptulose 7-phosphate + D-glyceraldehyde 3-phosphate	1
sedoheptulose 7-phosphate + D-glyceraldehyde 3-phosphate <=> ribose 5-phosphate + xylulose 5-phosphate	1
ribose 5-phosphate <=> xylulose 5-phosphate	1
Glycerone phosphate <=> D-Glyceraldehyde 3-phosphate	1

Supplementary Table 8. Ribulose monophosphate (RuMP) cycle (transaldolase variant, TA) reactions to calculate the minimal metabolic driving force by eQuilibrator.

Reaction	Relative flux
methanol + O2 <=> formaldehyde + H2O2	3
D-ribulose 5-phosphate + formaldehyde <=> hexulose 6-phosphate	3
hexulose 6-phosphate <=> D-fructose 6-phosphate	3
ATP + D-fructose 6-phosphate <=> ADP + D-fructose 1,6-bisphosphate	1
D-fructose 6-phosphate + D-glyceraldehyde 3-phosphate <=> D-xylulose 5-phosphate + D-erythrose 4-phosphate	1
D-fructose 1,6-bisphosphate <=> glycerone phosphate + D-glyceraldehyde 3-phosphate	1
glycerone phosphate <=> D-glyceraldehyde 3-phosphate	1
D-erythrose 4-phosphate + D-fructose 6-phosphate <=> D-sedoheptulose 7-phosphate + D-glyceraldehyde 3-phosphate	1
D-sedoheptulose 7-phosphate + D-glyceraldehyde 3-phosphate <=> D-ribose 5-phosphate + D-xylulose 5-phosphate	1
D-ribose 5-phosphate <=> D-ribulose 5-phosphate	1
D-xylulose 5-phosphate <=> D-ribulose 5-phosphate	2

Supplementary Table 9. Ribulose monophosphate (RuMP) cycle (sedoheptulose-1,7-bisphosphatase variant, SBP) reactions to calculate the minimal metabolic driving force by eQuilibrator.

Reaction	Relative flux
methanol + O2 <=> formaldehyde + H2O2	3
D-ribulose 5-phosphate + formaldehyde <=> hexulose 6-phosphate	3
hexulose 6-phosphate <=> D-fructose 6-phosphate	3
ATP + D-fructose 6-phosphate <=> ADP + D-fructose 1,6-bisphosphate	2
D-fructose 6-phosphate + D-glyceraldehyde 3-phosphate <=> D-xylulose 5-phosphate + D-erythrose 4-phosphate	1
D-fructose 1,6-bisphosphate <=> glycerone phosphate + D-glyceraldehyde 3-phosphate	2
glycerone phosphate <=> D-glyceraldehyde 3-phosphate	1
glycerone phosphate + D-erythrose 4-phosphate <=> D-sedoheptulose 1,7-bisphosphate	1
D-sedoheptulose 1,7-bisphosphate + H2O <=> D-sedoheptulose 7-phosphate + Pi	1
D-sedoheptulose 7-phosphate + D-glyceraldehyde 3-phosphate <=> D-ribose 5-phosphate + D-xylulose 5-phosphate	1
D-ribose 5-phosphate <=> D-ribulose 5-phosphate	1
D-xylulose 5-phosphate <=> D-ribulose 5-phosphate	2

Supplementary Table 10. Carbon fixing pathways and their energy demand when pyruvate is the output molecule for each pathway. For the conversion of Acetyl-CoA to pyruvate, a reaction catalyzed by pyruvate ferredoxin oxidoreductase with NADH as reducing equivalent was added.

Pathway	Net reaction	Input	Output	ATP/ C-mol	NAD(P)H/ C-mol
CBB cycle	$4 \text{ H}_2\text{O} + 7 \text{ ATP} + 5 \text{ NADPH} + 3 \text{ CO}_2 \rightleftharpoons 5 \text{ NADP} + 7 \text{ ADP} + 7 \text{ phosphate} + \text{pyruvate}$	CO_2	Pyruvate	2.33	1.67
rTCA cycle	$2 \text{ ATP} + 4 \text{ NADH} + 3 \text{ CO}_2 + \text{FADH}_2 \rightleftharpoons \text{H}_2\text{O} + 2 \text{ ADP} + 4 \text{ NAD} + 2 \text{ phosphate} + \text{pyruvate} + \text{FAD}$	CO_2	Pyruvate	0.67	1.33 (+0.33 FADH)
Wood Ljungdahl pathway	$\text{ATP} + \text{NADH} + 3 \text{ CO}_2 + 4 \text{ H}_2 \rightleftharpoons 2 \text{ H}_2\text{O} + \text{ADP} + \text{NAD} + \text{phosphate} + \text{pyruvate}$	CO_2	Pyruvate	0.33	1.67
rGlycine pathway	$2 \text{ ATP} + 2 \text{ NADPH} + \text{NADH} + \text{CO}_2 + 2 \text{ formate} \rightleftharpoons \text{H}_2\text{O} + 2 \text{ NADP} + 2 \text{ ADP} + \text{NAD} + 2 \text{ phosphate} + \text{pyruvate}$	Formate, CO_2	Pyruvate	0.67	1
Serine cycle	$\text{H}_2\text{O} + 3 \text{ ATP} + \text{NADPH} + 2 \text{ NADH} + 2 \text{ CO}_2 + \text{methanol} + \text{ubiquinone} \rightleftharpoons \text{NADP} + 3 \text{ ADP} + 2 \text{ NAD} + 3 \text{ phosphate} + \text{pyruvate} + \text{ubiquinol}$	Methanol, CO_2	Pyruvate	1	1
XuMP cycle	$\text{H}_2\text{O} + \text{ATP} + 3 \text{ O}_2 + \text{NAD} + 3 \text{ methanol} \rightleftharpoons \text{ADP} + \text{phosphate} + \text{NADH} + 3 \text{ H}_2\text{O}_2 + \text{pyruvate}$	Methanol	Pyruvate	0.33	- 0.33
XuMP cycle w/o SBP	$\text{H}_2\text{O} + \text{ATP} + 3 \text{ O}_2 + \text{NAD} + 3 \text{ methanol} \rightleftharpoons \text{ADP} + \text{phosphate} + \text{NADH} + 3 \text{ H}_2\text{O}_2 + \text{pyruvate}$	Methanol	Pyruvate	0.33	- 0.33
RuMP cycle TA version	$3 \text{ O}_2 + \text{ADP} + \text{NAD} + \text{phosphate} + 3 \text{ methanol} \rightleftharpoons \text{H}_2\text{O} + \text{ATP} + \text{NADH} + 3 \text{ H}_2\text{O}_2 + \text{pyruvate}$	Methanol	Pyruvate	- 0.33	- 0.33
RuMP cycle SBP version	$3 \text{ O}_2 + \text{NAD} + 3 \text{ methanol} \rightleftharpoons \text{NADH} + 3 \text{ H}_2\text{O}_2 + \text{pyruvate}$	Methanol	Pyruvate	0	- 0.33