**FOREWORD** 

**INTRODUCTION** 

<u>1,1'-DIFLUOROETHYLENE (VDF,VF2)</u> CAS N°: 75-38-7

# SIDS Initial Assessment Report for 13<sup>th</sup> SIAM (Bern, Switzerland, November 6-9, 2001)

Chemical Name: 1,1-difluoroethylene (VDF, VF2)

**CAS No. :** 75-38-7

Sponsor Country: United States of America

#### National SIDS Contact Point in Sponsor Country (or Lead Organisation which ever is

#### applicable):

United States Environmental Protection Agency Oscar Hernandez, Director Risk Assessment Division 1201 Constitution Ave, NW Washington, DC 20460 (202) 564-7641 hernandez.oscar@epa.gov

#### **Industry**:

American Chemistry Council Sherri Clark 1300 Wilson Blvd Arlington, VA 22209 (703) 741-5619

#### **History**:

The work and review process undertaken for this chemical was done by industry and the United States Environmental Protection Agency. Documents were prepared and reviewed by industry prior to submission to sponsor country. Sponsor country conducted reviews of submitted data and offered comments to industry. Industry prepared and resubmitted documents for consideration at SIAM 13. All data were obtained within the consortia's archives and libraries. In addition, databases that were searched may be found in section 7 of the SIAR.

Testing :	No testing (X)
	Testing ()

## **Comments** :

Deadline for Circulation : September 14, 2001

#### **Date of Circulation :**

## SIDS INITIAL ASSESSMENT PROFILE

CAS No.	75-38-7
Chemical Name	1,1-Difluoroethylene or vinylidene fluoride
Structural Formula	$F_2C=CH_2$

## RECOMMENDATIONS

The chemical is currently of low priority for further work.

## SUMMARY CONCLUSIONS OF THE SIAR

#### Human Health

Metabolic/kinetic studies with vinylidene fluoride (VF2) in rats and mice indicate that during inhalational exposure, VF2 reaches a maximum level in blood at 15 minutes. Some metabolism takes place in the liver. It has been hypothesized that some fluoroacetic acid may be formed during metabolisation of VF2, which in turn may interfere with the citric acid cycle. However, mammalian toxicity study results do not indicate that this leads to structural or functional pathological changes.

Acute toxicity of VF2 is low with 1 hr LC00 of 200,000 ppm (524,000 mg/m<sup>3</sup>) in rats. Only slight CNS effects were noted at very high concentrations (80% in air). Cardiac sensitization studies were negative. Several inhalation exposure studies have been conducted in rats and mice exposed 6h/d, 5d/week for 13 weeks. No target organs were consistently identified although effects on the kidney, spleen and testes were reported in various studies. In rats and mice the LOEC of 500 ppm (13,100 mg/m<sup>3</sup>) was identified, based on body, organ weight and clinical chemistry changes in the absence of histopathological changes. A NOEC of 250 ppm (6,550 mg/m3) was identified in rats. At 40,000 - 50,000 ppm (104,8000 - 131,000 mg/m<sup>3</sup>) effects on the nasal epithelium were noted in rats.

In chronic toxicity/carcinogenicity inhalation studies, rats and mice were exposed 6 h/d, 5d/week for 24 and 18 months, respectively at concentrations up to 10,000 ppm (26,200 mg/m<sup>3</sup>). Neoplastic findings were comparable in control and treated animals. An earlier 52 week study in rats exposed orally to VF2 indicated increased lipomas/liposarcomas. However, this study was performed according to a protocol with significant deviations from currently prescribed guidelines and was reported in insufficient detail for proper evaluation. In genotoxicity studies, VF2 has shown some activity in bacterial assays, but was negative in the *in vitro* chromosomal aberration and gene mutation study in mammalian cells. *In vivo*, VF2 was negative in a mouse micronucleus and Drosophila SLRL test. Thus there is no evidence of genotoxicity *in vivo* Overall, the results suggest that VF2 does not present a genotoxic hazard to man.

VF2 did not induce teratogenic or embryofetal toxicity effects in developmental toxicity studies in rats exposed up to 10,000 ppm during gestation days 6 –15. The NOEL for reproductive effects is  $\geq$  7000 ppm (18,340 mg/m<sup>3</sup>) in rat studies.

#### Environment

VF2 is a gas at ambient temperatures **a**d atmospheric pressure. Emissions will only occur during production and processing of VF2 and will partition nearly exclusively to air (>99%). Its low log  $P_{ow}$  does not indicate any significant bioaccumulative potential. In air, VF2 will be degraded by reaction with hydroxyl radicals. A half-life of 3.3 days has been calculated. Likely primary products resulting from the tropospheric degradation of VF2 are COF<sub>2</sub> and formaldehyde. Fluorogyloxal (CFOCHO) may also be a product of the degradation of VF2. The ultimate degradation products are formaldehyde, HF and CO2.

No biodegradation studies in water have been performed for VF2, however, related gaseous materials (tetrafluoroethane, pentafluoroethane, difluoromethane, 1-chloro-1,1-difluorethane, vinylidine chloride) generally

showed < 10% degradation indicating that transformation to metabolites in soil or the water compartment may be considered very low. Based on these analogous substances, VF2 is not expected to be readily biodegradable and testing **i** not recommended. Due to specific physico-chemical properties of VF2 its production and use pattern, and its nearly exclusive partitioning to air, no aquatic toxicity testing has been performed. Using QSAR, the LC50 (96 hr) for fish is 245 mg/L, the daphnia LC50 (48 hr) is 250 mg/L and the green algae EC50 (96 hr) is 149 mg/L.

#### Exposure

VF2 is almost exclusively used as a monomer for the production of fluoropolymers (polyvinylidene flouride) and as copolymer with hexafluoropropylene and chlorotrifluoroethylene. In the United States, production is performed in closed systems and is anticipated to be representative of global production methods based on the chemicals physical chemical properties. Global production in 1999 was approximately 33,000 tonnes (72,600,000 pounds). The emissions of VF2 come exclusively from production and processing installations of VF2. The segment of the population exposed directly or indirectly to VF2 is very limited: workers during production and processing. Consumers manipulating goods made of VF2 (polymers) are not exposed to VF2. Possible emissions during production and processing are low and lead to very low atmospheric concentrations. VF2 is not expected to be released to water systems but in case of emissions will totally partition to air. It will not partition to the water from the air. VF2 is a flammable gas at ambient temperature (limits 4.7 % to 25.1%). Its flammability constitutes its most important physical danger.

## NATURE OF FURTHER WORK RECOMMENDED

No recommendation.

## FULL SIDS SUMMARY

CAS N	CAS Nº 75-38-7 SPECIES		PROTOCOL	RESULTS
PHYS	ICO-CHEMICAL	•		
2.1	Melting point		handbook	-144°C
2.2	Boiling point		handbook	-83 to -85.7°C
2.3	Density			0.67 to 0.92 gm/cm <sup>3</sup> at 20°C
2.4	Vapour pressure		handbook	35700 hPa at 20°C
2.5	Partition coefficient		calculated	1.24
2.6	Water solubility		handbook	254 mg/l at 25°C
2.11	Oxidising properties	Not considered to have oxidisi	ing properties	
2.12	Additional remark Soil Koc	Henry's law constant: 0.226 7.12	atm.m3/mol	
ENVI PATH	RONMENTAL FATE AND WAY			
3.1.1	Photodegradation		calculated	Rate cst: $2*10^{12}$ to $2.04*10^{-13}$ cm <sup>3</sup> /molecule/sec T1/2: 3.3 days
3.1.2	Stability in water	No data. VF2 will quickly eva (see Mackay modeling)	aporate to the atmosph	eric compartment
3.2	Monitoring data	No data		
3.3	Transport and Distribution	According to the Mackay model (level 1) more than 99.6% of emitted VF2 will collect in the atmosphere with negligible amounts remaining in water, sediment and soil.		
3.5	Biodegradation	No data. VF2 will evaporate from soil and surface waters before any significant degradation can take place.		
ЕСОТ	OXICOLOGY	_		
4.1	Acute/prolonged toxicity to fish			No data
4.2	Acute toxicity to aquatic invertebrates			No data
4.3	Toxicity to aquatic plants e.g. algae			No data
4.4	Toxicity to micro- organisms e.g. bacteria			No data
4.5.1	Chronic toxicity to fish			No data
4.5.2	Chronic toxicity to aquatic invertebrates			No data
ΤΟΧΙ	COLOGY	_		
5.1.1	Acute Oral			Not applicable
5.1.2	Acute Inhalation		Lester at al., 1950	> 80% V/V
5.1.3	Acute Dermal			Not applicable
5.2.1	Skin irritation/corrosion			Not applicable

5.2.2	Eye irritation/Corrosion			No specific test results but no irritation observed during other tests.
5.4	Repeated dose	Rat, 2 weeks, inhalation	Dupont Cy. OTS0215306, 1977	NOEL < 25000 ppm
		Mouse, 2 weeks, inhalation	Newton, 1988	NOEL = 40000 ppm
		Rat, 13 weeks, inhalation	Reuzel, 1986	NOEL: 1000 ppm LOEL: 7000 ppm
		Mouse, 13 weeks, inhalation	Newton, 1989	NOEL: 7000 ppm LOEL: 40000 ppm
5.5	Genetic Toxicity In vitro			
	Bacterial test	S.typhimurium	Russell, 1979	Positive only for TA1535
		S.typhimurium TA 100	Bartch, 1979	strain in presence of S9
	Non-bacterial test	CHO cells/ HGPRT	Rickard, 1986	Negative with and without S9
		CHO cells/chromos.	Rickard, 1986	Negative with and without S9
		Aberration		Regative with and without 57
5.6	Genetic Toxicity In vivo	Mouse, micronucleus, inhalation	Hodson-Walker, 1988	Negative
		D.melanogaster, sex linked recessive lethal test	Sernau, 1988, 1989	Negative
5.7	Carcinogenicity	Rat, 52 weeks, oral	Maltoni, 1979	NOEC < 4.12 mg/kg
		Rat, 104 week, inhalation	Arts, 1991	NOEC: 10000ppm
		Mouse, 78 weeks, inhalation	Newton, 1991	NOEC: 10000 ppm
5.8	Reproduction Toxicity	Rat, 13 weeks, inhalation	Reuzel, 1986	NOEC: = or >7000 ppm
		Rat, 13 weeks, inhalation	Koeter,	NOEC: = or > 7000 ppm
5.9	Development / Teratogenicity	Rat, gestation day 5 to 15, inhalation	Mecler, 1978	NOEC: = or > 10 000 ppm
5.11	Human experience	No data.		

## SIDS INITIAL ASSESSMENT REPORT

## 1,1-Difluoroethylene

# 1. Identity

Purity	99.5 % (does not contain any additives for
	stabilization.)
Molecular weight	64
Melting point	-144°C
Boiling point	$-83 \text{ to} - 85^{\circ}\text{C}$
Vapor pressure	35700 hPa (26960 mmHg) at 20°C
Water solubility	254 mg/L at 25C
Partition Coefficient (Log Kow)	1.24
Henry's Law Constant	0.226 atm.m <sup>3</sup> /mole
Other:	Colorless, flammable gas at ambient temperature

## 2 General Information on Exposure

#### **Production volume**

Solvay, Atofina, and Ausimont are the major producers of VF2. The production sites are located in France and the USA. Global production in 1999 was approximately 33,000 metric tons (72,600,000 pounds).

## **Production method**

Several methods for manufacture of VF2 are based on pyrolysis including the dechlorination of 1,2dichloro-1,1 – difluoroethane in the presence of a catalyst and the dehydrofluorination of 1,1,1trifluoroethane. Based on physico-chemical properties VF2 is produced in closed systems.

## Uses and function of VF2

VF2 is used to manufacture polyvinylide ne fluoride and elastomeric copolymers with chlorotrifluoroethylene and hexafluoropropylene. The polymerisation process is performed in closed systems. VF2 is usually polymerised above its critical temperature of 30.1°C and at pressures above 3 Mbars (30 atm). VF2 is a closed system intermediate.

## Form of Marketed Product

Manufacturers of VF2 either polymerize the compound on-site or sell and transport the substance in bulk quantities as liquefied gas under pressure (100 gallon gas cylinders or tube trailers).

## Information on Safe Handling Procedures (Storage)

VF2 is stored in open air and protected from direct sunlight in steel containers (typical 50 to 100 m<sup>3</sup>) fitted with a safety valve or vent and pumps are monitored by explosimeters. These containers are stored away from ignition and heat sources. Equipment and piping containing VF2 is grounded to avoid static build up and electrostatic discharges.

## 2.1. Sources of Releases to the Environment

## **Production and processing releases**

VF2 is handled in closed systems under high pressure. Exposure during handling is considered to be negligible since it is carried out in sealed pipes and vessels. Emissions will therefore be incidental during equipment failures and maintenance operations. Monitoring systems are in place and capable of detection at levels of 0.1 ppm. Leaks are therefore quickly revealed and corrected.

## Release from polymer use

Polymers made from the monomer VF2 do not contain unreacted monomer above 1 ppm. The polymers, which are thermally, che mically, and UV-light resistant, are used for tank linings, pumps, valves and lined pipes in chemical or food processing equipment. When the polymer product is used with compatible materials, the decomposition of the polymer is minimal, leading to negligible concentrations in food, when used in food processing equipment, or to the environment.

## 2.2 Human exposure

VF2 is a gas at room temperature. Therefore, the main route of human exposure is via inhalation. The two main exposure groups assessed for possible VF2 inhalation are occupational groups and populations residing in locations in the vicinity of producing and processing facilities.

## 2.2.1. Occupational exposure

Potential exposure to VF2 is limited to its production and polymerization. The American Conference of Governmental Industrial Hygienists', Threshold Limit Value (TLV), Time Weighted Average (2001) for VF2 is 500 ppm.

In a survey of a VF2 production plant in 1992-1993, a mean work place VF2 concentration of 0.9 ppm was determined. During 0.7% of the working time a level in the work atmosphere of 22.7 ppm was reached. (Solvay, unpublished data, ref 58) In the mid 1980's, two polymerization plants in the USA were monitored for exposures to VF2 during the polymerization process. In one plant, there were no exposures noted above 1.2 ppm VF2 (TWA). In the second plant (personal sampling), depending on the function of the person involved, VF2 atmospheric measurements ranged between < 0.1 ppm and 40 ppm. All measurements above 10 ppm were from incidental exposures (other than normal working conditions and maintenance) and of short duration. For example, the 40 ppm (personal sampling) value was obtained during non-routine work or maintenance. An emergency repair of a malfunctioning pump required that a vapor line be bleed resulting in a short-term elevated exposure to the maintenance worker. (US EPA, draft 1999, ref 57) Since the performance of these surveys, process controls and structural production conditions have been improved, so that possibility of exposure of workers to VF2 has further decreased.

No monitoring data are available for other VDF manufacturers/processors in other countries. Since manufacturing/processing is similar world-wide for VF2, it may be assumed that exposure values would be within the same range.

## 2.2.2. Consumer Exposure

Consumer exposure is not anticipated based on the physico-chemical properties and use of VF2. There are no direct consumer uses of fluoroalkenes. In addition, final products for which VF2 is used do not contain unreacted VF2 above 1 ppm. Thus, consumer exposure is not anticipated.

## 2.2.3. Indirect Exposure via the Environment.

VF2 is a gas, has a low calculated log Kow (1.24) and is thus considered to have no significant bioaccumulation potential. Records indicate that during manufacturing and processing of VF2, under normal operating conditions, releases to the environment are anticipated to be minimal. Therefore, indirect human exposure is not expected.

## 2.3 Environmental Fate

## 2.3.1. Distribution in Air, Water, Soil

VF2 is a gaseous substance at ambient temperature (BP: -83 to -85.7°C, VP: 35700 hPa at 20°C) with a limited solubility in water (254 mg/l at 20°C and 1 bar atmospheric pressure). Under natural conditions, this compound will evaporate from water and almost completely enter the atmospheric compartment. This is indicated by the results of the Level I and III Mackay modeling based on the physico-chemical data as mentioned in the SIDS dossier. Estimated values indicate that more than 99.94 % of emitted VF2 will partition to the air compartment.

The remaining amount (0.05%) will distribute to the aquatic compartment.

VF2 is unlikely to persist in water due to its high vapour pressure (35700 hPa). The EPIWIN model calculates the half life (T  $_{1/2}$ ) in river water as 0.8 hours and T  $_{1/2}$  in lake water as 76 hours (See IUCLID Dossier). Due to its density in the liquid phase (0.67 g/cm3 at 25°C), VF2 will float on water rather than dissolve in it if it is accidentally released under supercooled conditions. Its leaching into soil and ground water in such circumstances will also be negligible. Its (calculated) log Kow (1.24) indicates that the substance would not significantly partition to organic material in soil or water. The calculated soil Koc (partition coefficient organ carbon- water) is 7.12. Hence the presence of organic matter would not contribute significantly to increased concentrations in these compartments over those determined by its solubility and volatility.

## 2.3.2. Abiotic and Biotic Degradation in Air, Water and Soil

No experimental data are available on the degradation of VF2. The Biowin program in the EPIWIN model indicates that VF2 will degrade in water and soil (timeframe of days to weeks). However, under the natural conditions prevailing in soil, natural waters and under standard temperature and pressure, VF2 and similar compounds will partition completely to the atmospheric compartment before any significant degradation could take place. In the atmosphere, VF2 is considered to be degraded by reaction with OH radicals. Howard (1976) calculated a Rate constant for this reaction of  $2*10^{-12}$ . Life-time calculation by the Prahter and Spivakovsky procedure yields a half life for VF2 of 3.3 days. The AOP program (See IUCLID Dossier for details) in the EPIWIN model yields similar results: T  $\frac{1}{2}$  (OH radicals): 4.7 days and T  $\frac{1}{2}$  (O3): 40.9 days.

Likely primary products resulting from the tropospheric degradation of vinylidene fluoride are  $COF_2$  and formaldehyde. Within a few weeks,  $COF_2$  will be hydrolyzed to  $CO_2$  and HF after uptake into cloud droplets. Fluorogyloxal (CFOCHO) may also be a product of the degradation of VF2, but this remains uncertain. If fluorogyloxal is formed, it will disappear rapidly from the atmosphere, probably mainly by photolysis on a time scale of a few hours, ultimately giving HF and carbon oxides.

## 2.3.3. Bioaccumulation in Different Environmental Compartments

VF2 has a calculated log Kow of 1.24 (KOWWIN program in the EPI WINN model). This value indicates that possible bioaccumulation in the food chain is not significant. Moreover, the limited solubility of VF2 as well as its almost total distribution to air and its short half life in that compartment makes significant contact with the organisms in the food chain negligible. This is supported by the BCF program results in the EPIWIN model (BCF: 1.8). (See IUCLID Dossier for details.)

Therefore, it is considered that bioaccumulation of VF2 in the food chain is of low concern.

## 2.4 Predicted Environmental Concentration.

No environmental monitoring data are available.

In view of the similar production and processing conditions of VF2 worldwide (closed systems), its physico-chemical properties, and its predicted distribution and degradation in air and water, the environmental concentrations of VF2 in the different environmental compartments are considered to be negligible.

## 3.0 Human Health Hazards

## 3.1. Effects on Human Health

## 3.1.1. Toxicokinetics / Metabolism

#### **Cellular interactions**

Rats exposed to VF2 concentrations up to 25000 ppm for up to 6 hrs, did not show hepatotoxic signs as measured by liver weight, histology and serum sorbital dehydrogenase activity. (Conolly, et al, 1979, ref 43)

Fed and fasted rats exposed to VF2 concentrations of up to 82000 ppm for 4 hrs did not show a toxicologically significant increase in serum alanine alpha-ketoglutarate transaminase. (Jaeger et al, 1975, ref 44)

## Toxicokinetics/Metabolism

VF2 was found to inactivate P450 enzyme and heme only to 17% in primed and unprimed rat hepatic microsomes when exposed to  $60\mu$ M for 30 minutes. Fluor was only minimally released in this test. (Baker et al., 1987, ref 46)

When rats were exposed to VF2 concentrations of 2200 ppm for 30 minutes, the animals showed an increased Fluor urinary excretion for 7 days post exposure. Creatinine excretion and urinary volume were not increased. The medulla of the kidneys showed hyperemia and a pale whitish band in the cortex, but no histopathological changes were observed in that organ. (Dilley et al., 1974, ref 47) After 8 hrs exposure to VF2 concentrations of 500 ppm, rats exhaled acetone, which was considered by the authors as an indication that VF2 is partly metabolized to fluoroacetate and that VF2 interacts with hepatic cytochrome P450. (Filser et al., 1980, ref 50)

Metabolism seems saturated at the 100 ppm exposure concentration. Vmax was calculated to be 1.1  $\mu$ M/hr/kg. (Filser et al., 1978, ref 48) Filser suggested that an epoxide intermediate is formed with subsequent rearrangement to halogenated acetaldehyde or acyl halide and formation of fluoroacetic acid; the latter could interfere with the citric acid cycle.(Filser et al., 1980, ref 50) Zwart (1985, ref 52) observed a saturation of VF2 metabolism in rats at an exposure concentration of 400 ppm. Stoeckle et al., (1979, ref 51) found a similar metabolism rate for VF2 as that observed by Filser in rats and observed that exposure to 2000 ppm VF2 for 14 weeks induced only a minimal ATPase deficiency in rat hepatocytes. Bolt et al., (1979, ref 60) reported that interaction with P450 with slowly metabolized compounds like VF2 may lead to inhibition of metabolism of drugs and other xenobiotics. Inhibition of p-hydroxylation of aniline and demethylation of aminopyrine by VF2 served as indicator reactions.

Medinski et al., (1990, ref 54) calculated kinetic parameters for VF2 in rats. Steady state blood levels of VF2 in rats were 15 ng/ml at 30 ppm exposure and 2400 ng/ml at 16,000 ppm exposure. Tmax was 15 minutes for all concentrations tested. After exposure blood levels decreased to 10% of Cmax in 1 hr. VF2 tissue air partition coefficients were: 0.07, 0.18, 0.8, 1.0 and 0.29 for water blood, liver, fat and muscle respectively.

Kinetic values measured and calculated for mice seemed to be lower than those found for rats. (Bechtold et al.,1988, ref 55)

## Comments on the toxicokinetic and metabolism studies

The above -mentioned studies give a fair idea of the molecular behavior of VF2 in the mammalian organism, and indicate that some metabolism of the molecule takes place. The long-term toxicity/carcinogenocity studies mentioned below show that the metabolic fate of VF2 in animals

does not lead to structural nor functional adverse effects at the concentrations tested (up to 10000 ppm for 2 years in the rat).

## 3.1.2 Acute toxicity

Several acute inhalation toxicity tests have been performed, which are more or less comparable to LC50 studies.

In a series of acute inhalation experiments, the lethal concentration for rats and mice after 1-hour exposure was greater than 200,000 ppm. Animals were observed for 7 days post exposure. Only the mice showed some minor behavioral changes. (Pennwalt, 1982, ref 15) In another test, rats were exposed to levels up to 80% (+20% oxygen) for up to 19 hours. No mortality was observed. At exposures to a concentration of 80%, some minor CNS effects were noted. No pathological changes were reported. (Lester et al. 1950, Ref 17) Dogs and cats survived exposure to 50% in air without cardiac sensitizing effects. (Burgison, 1955, ref 18) Other test reports mentioned mortality of mice at 128000 ppm after a four-hour exposure but results are difficult to interpret because of variable exposure periods and the possibility of inadequate oxygen supply. (Carpenter et al, 1949, ref 16)

In an acute hepatotoxicity test with Holtzman rats, animals were exposed to concentrations of 0, 500, 15000 and 25000 ppm for 4 to 6 hrs. No mortality was observed. Only animals pretreated with polychlorinated biphenyls showed increased liver weights, increased serum sorbital dehydrogenase activity and hepatocellular damage at VF2 concentrations of 500 ppm VF2 and above. (Conolly et al, 1979, ref 43)

Although none of the mentioned studies were performed according to currently prescribed standard protocols, the Pennwalt data indicate clearly that VF2 has a low acute toxicity. The other acute studies corroborate the weight of evidence for low acute toxicity.

Since the substance is a gas at ambient temperature, no specific primary cutaneous or ocular irritation studies were performed, but other toxicity study results indicate that VF2 has no significant irritancy to skin and eyes. Accidental releases of liquefied VF2 gas may cause frostbite when in contact with skin.

## 3.1.3 Repeated Dose Toxicity

CD1 mice were exposed by inhalation to VF2 concentrations of 0, 1000, 5000, 15000 and 40000 ppm for 2 weeks (6hrs/d, 5d/wk). No mortality was noted. There were no treatment-related effects on body weight, organ weights. No pathological changes were observed. The NOEL in this study was considered to be 40000 ppm VF2. (Newton, 1988, ref 24)

In an older 2-week inhalation study in albino rats, in which animals were exposed to 25000 ppm VF2 (6hrs/d, 5d/wk), a slight increase in urinary fluoride excretion, changes in RBC count and respiratory structural changes were noted. Due to lack of details of the study procedure these findings cannot be properly evaluated. Impurities in the test substance and/or a preexisting infection may have contributed to the observed effects. (Dupont, 1977, ref 20)

Weanling and young adult Sprague -Dawley rats were exposed by inhalation to VF2 concentrations of 0, 250, 1000 and 7000 ppm for 13 weeks (6hrs/d, 5d/wk). This study was combined with a fertility/reproduction study. (see also Koeter, 1986, ref 39) An interim sacrifice and evaluation was made at 4 weeks. Satellite groups were subjected to a recovery period of 10 weeks after exposure. Compared to control no treatment related changes were observed regarding body weight, hematology, urinalysis, mortality, estrus cycle, and organ weights. Gross pathology revealed no changes; on histopathology, no treatment-related effects were noted on sperm concentration, number of sperm cells with deformed heads/tails or on numbers of isolated heads/tails in weanlings or young adults. A slight degeneration of the vomeronasal organ in animals exposed to 7000 ppm was observed. At four weeks, the change was noticed both in weanlings and in young adults. At 13 weeks, it occurred only in weanlings. After 10 weeks recovery, it was not observed in any animal.

(Reuzel, 1986, ref 19) A simple fertility study was also conducted as part of the study, in which male and female rats that had been exposed to VF2 in the 13-week subchronic portion of the study were mated with untreated partners. No effects of potential reproductive significance were observed. (Koeter et al., 1986, ref 39)

CD1 mice were exposed by inhalation to VF2 concentrations of 0, 1000, 7000 and 40000 ppm for 13 weeks (6hrs/d, 5d/wk). Treatment-related, but not dose-related, increase in locomotor activity was noted. The increase was observed in both sexes and was prominent in mid study. It declined afterwards. Increased sensitivity to touch was seen in the 40000 ppm males during the last week of exposure and in mid and high dose females in the last 2 weeks of exposure. Dose-related rough haircoat was observed in males after eight weeks of exposure.

An increase in mean corpuscular Hb was seen in the high dose males. There were no-treatment related effects on body weight, food consumption nor on organ weights. No treatment related macroscopic or microscopic pathological changes were observed. Since the (subjective) clinical symptoms are not related to pathological changes the NOAEL is considered to be 7000 ppm. (Newton, 1989, ref 25)

Fisher 344 rats were exposed by inhalation for 90 days (6hrs/d, 5d/wk) to 0, 500, 1500, 5000, 15000 and 50000 ppm VF2. Transient body weight changes were observed which had subsided at the end of the study. Unexplained body weight changes were also seen in the controls. RBC counts, Hb levels and hematocrit decreased in males of the 1500 and 50000 ppm group. No such changes were observed in the females. In this study, some changes in clinical chemistry (9 SGPT, 8SGOT, 8SDH, 8creatinine, 8 BUN) and organ weights (relative and/or absolute: 8liver, 8 kidney, 8thymus, 9brain, 9heart, 9testis) were observed. However, there was no dose relationship nor clear relation to gender. Histopathological examination only revealed one animal with serous rhinitis and erosion of the eptelium of the nasal cavity (50000 ppm). The authors concluded that the NOEC in this study was < 500 ppm, and although observable effects were noted at all exposure levels, dose related effects were only apparent at the highest concentration. (Manus et al., 1984, ref 21)

In another study, Manus et al., (1984, ref 22) exposed BCF mice by inhalation for 13 weeks (6hrs/d, 5d/wk) to similar VF2 concentrations as in the above mentioned rat study. Some mortality and transient bodyweight loss during the study was observed which was suspected to be related to a failing watering system. In this study body weight changes were seen in females. Liver weight changes occurred (1500 ppm males and 15000 ppm females) but were not dose related. On histopathology, only mild renal changes indicative of regeneration was observed in very low incidence in all dosed males and high dose females and was not evidently related to dose. The authors concluded that the NOEC in this study was < 500 ppm.

An inhalation study was performed with Sprague-Dawley rats, whereby animals were exposed for 13 weeks (6hrs/d, 5d/wk) to VF2 concentrations of 0, 1000, 7000 and 40000 ppm. Interim sacrifices and observations were made at 2 and 4 weeks. In this study changes in body weight (transient in males and females), hematological parameters (from 7000 ppm in males and in 40000 ppm in females), clinical chemistry (in mid and high dose males and high dose females), urinalysis (in high dose males and transient in mid and high dose females) and organ weight changes (heart, lungs, spleen,) were observed.

These occurred at different concentrations, were not evidently concentration related and were sometimes transient. Microscopic examination of testis and epidydimis showed treatment related changes in mid and high dose groups throughout the study; these were characteristic of impaired spermatogenesis. The spleen showed treatment related changes characterized by lymphocytic depletion of the marginal zone, in males and females of the mid and high dose groups and in females of the low dose group at week 2. Vacuolar degeneration of the vomeronasal organ in the nose was observed in all treatment groups. Mineralization of the kidneys was increased in males of the high dose group after week 4. The NOEL in this study was considered to be < 1000 ppm.

(Appelman et al., 1985, ref 23)

## 3.1.4 Genetic Toxicity

Several in-vitro and in-vivo genotoxicity tests were performed with VF2.

#### In vitro studies

In an Ames test, in which *S. typhimurium* strains TA 1535, TA 100, TA1537 and TA98 were exposed (with and without the S9 activation system) to concentrations up to 50% VF2 in the gas phase, only strain TA 1535 reverted significantly at concentrations of 10% VF2 and above and only in presence of the activation system (number of revertant colonies increased maximally 2.6 time those of controls). (Russell, 1979, ref 29)

Bartsch et al, (1979, ref 30) exposed *S. typhimurium* TA 100 24 hrs to VF2 concentrations ranging from 20 to 50% in the gas phase; a marginal non-significant increase in revertant colonies were noted at 50% and only in the presence of the S9 activation system.

In an HGPRT test, Chinese Hamster Ovary (CHO) cells were exposed to VF2 concentrations of 0, 20, 40, 60, 80 and 100% in the gas phase for 19 and 5 hours (without and with S9 activation system). No mutant frequency increase was observed at any concentration. (Rickard, 1986, ref 31) In a separate study, the same author (Rickard et al., 1986, ref 32) also exposed CHO cell to investigate the occurrence of chromosomal aberrations. VF2 concentrations were 0, 25, 50, 75 and 100% in the gas phase. Exposure period was 5 and 2 hours (without and with S9 activation system). No chromosomal aberrations were observed at any concentration in the presence or absence of the activation system. VF2 had no toxic properties to the cells at any concentration (as measured by replication delay in subcultures).

In some older in-vitro tests (an *E. coli* test (Landry et al., 1968, ref 26) an Ames test (Jagannath, 1977, ref 27) and a BALB/3T3 cell transformation test (Matheson, 1978, ref 28)) mutagenic effects were observed. Since the exposure concentrations were not described or unknown and in some cases the studies were not adequately detailed the results are not as reliable as the more recent studies mentioned above.

## In vivo studies

Male and female mice were exposed to 0, 5000, 15000 and 40000 ppm VF2 for 6 hrs. Their bone marrow was collected at 24, 48 and 72 hrs after initiation of the exposure for microscopic evaluation of incidence of micronuclei in erythrocytes. VF2 was found not to be toxic to bone marrow nor influence the incidence of micronuclei at any concentration tested. (Hodson-Walker, 1988, ref 33)

A Sex Linked Recessive Lethal test was performed in male *Drosophila melanogaster* (fruit flies). Animals were exposed for 24 hrs to 0, 4.95, 22.8 and 43% VF2. After exposure, males were mated with non-exposed females. There was no significant difference in lethality of the progeny between controls and exposed males. Therefore, VF2 was not considered to be mutagenic to the X chromosome of *D. melanogaster*. (Sernau, 1988, 1989, refs 34, 35)

From all the presented genotoxicity tests it can be concluded that VF2 does not significantly interfere with the genome of organisms. The marginal, but positive finding in *S. typhimurium* TA1535 ( $\geq$ 10% VF2 in presence of an activation system) cannot be explained: it may indicate that some metabolite of VF2 may interfere directly or indirectly with the genomic integrity of some selected protocaryotic organisms.

## 3.1.5. Carcinogenicity

Sprague -Dawley rats were exposed by inhalation to VF2 concentrations of 0, 150, 600, 2500, 10000ppm for 104 weeks (6hrs/d, 5d/wk). At 12 months of exposure an interim sacrifice and examination was performed. No treatment-related effects were noted on survival, clinical

symptoms, ophthalmology, body weight gain, hematology, urinalysis, macroscopic and microscopic pathology. Food consumption tended to be lower in treated animals than in controls. Relative organ weights (brain, heart, epididymis) of the 150 ppm males only were decreased at study termination. There were no treatment related shifts in benign or malignant tumor incidence, total number of tumors, or total number of tumor bearing animals. (Arts, et al., 1991, ref 36) Newton (1991, ref 37) exposed mice by inhalation for 18 months to VF2 concentrations of 0, 600, 2500 and 10000 ppm. (6hrs/d, 5d/wk). No treatment- related effects were observed in survival, clinical symptoms, ophthalmology, body weight food consumption, hematology, nor in macroscopic or microscopic pathology. Incidence of observed benign and malignant neoplasms in exposed animals was not different from controls and was not considered related to treatment. In a one year study, Sprague-Dawley rats were fed 0, 4.12 and 8.25 mg/kg BW by gavage for 52 weeks (4 to 5 d/wk) VF2 was dissolved in olive oil. Animals were allowed to die naturally. Lipoma and liposarcomas were observed at necropsy after ratural death. The number of lipomas was increased in the high dose group compared to controls. None were seen in the low dose group. There was a dose-related trend in the numbers of liposarcoma's. However, this study was performed according to a protocol with significant deviations from currently prescribed guidelines (number of animals, exposure period, route of administration, statistical evaluation). The study was not reported in great detail, which impedes proper interpretation of the results. Moreover liposarcomas were found in fat tissue of different embryonal origin but considered as one type of tumor. (Maltoni et al., 1979, ref 38)

## 3.1.6. Reproductive/developmental toxicity

As previously mentioned in the repeat dose section, studies on male fertility have been documented. In Appelman et al., 1985 (ref 23), upon microscopic examination of the testis and epidydimis, it was noted that treatment related changes in the mid and high dose groups were observed. Observations were characteristic of impaired spermatogenesis. However, these findings were not reproduced in a subsequent study by Reuzel et al (1986, ref 19). In this male fertility study, males were exposed to VF2 for 13 weeks then specially evaluated for effects on the gonads: (organ weights, weeks 5 and 13: coagulating glands, epididymis, prostate, seminal vesicles, testes, uterus and testes of animals in the 10 week recovery group. Microscopic evaluation: Sperm morphology at week 14, histological eximination of testis and epididymis (all animals), ovaries and uterus (control and high dose). The gross and microscopic pathology evaluations revealed no treatmentrelated effects on sperm concentration, on number of sperm cells with deformed heads/tails, or on numbers of isolated heads/tails in weanlings or young adults. Reuzel established a NOEL of 7000 ppm for those fertility effects.

Koeter at al (1986, ref 39) exposed male and female rats for 15 weeks prior to mating and 2 weeks during mating to 0, 250, 1000 and 7000 ppm VF2. After mating females were exposed until day 15 of gestation to the same concentrations. No effects were observed at any concentration on mortality, parental body weight, fertility indices, reproductive performance, male reproductive organ weights, or on histopathology. There were no treatment-related effects on ovary weights, or on litter data. The NOEL was considered to be greater than or equal to 7000 ppm.

In a teratology study female pregnant rats were exposed to 0, 2000 and 10000 ppm VF2 during gestation days 6 to 15. Mothers were sacrificed on day 20 of pregnancy. No effects were observed at any concentration on maternal body weight, food consumption, nor on number of implantation sites, corpora lutea live and dead fetuses resorption sites. No treatment-related effects were seen at any concentration tested on fetal soft tissues of head, thoracic and visceral organs, nor on skeletal structures. The NOEC for teratogenicity was considered to be greater than or equal to 10000 ppm. (Meckler et al., 1978, ref 40)

## 3.1.7 Physical hazard

VF2 is a flammable gas. This property constitutes the substance's most important physical hazard.

## 3.1.8 Initial assessment for human health.

Due to the conditions of production and use of VF2 (closed system intermediate), resulting occupational concentrations are considered to be low. This is corroborated by, albeit limited, occupational exposure surveys (exposure normally below 10 ppm).

In this hazard evaluation, the primary studies to be used to determine the chemical's hazard are the chronic toxicity/carcinogenicity studies. It is believed that the overall quality, detailed description and length of the studies, provides a more sound basis for determining the hazard of VF2. Many of the subacute/subchronic studies do not have the level of detail provided in the more recent GLP chronic studies. In addition, many of the effects observed in the subchronic studies were not considered to be dose related or transient and in many cases not related to organ pathology. The overall weight of evidence regarding the studies and the inability to reproduce the same effects in the chronic GLP studies should be taken into consideration. The Arts and Newton studies are considered to be the key studies for the evaluation of VDF. These two studies were performed according to currently accepted guidelines and GLP standards. Since they are lifetime studies which are reported in great detail, they are considered as the most reliable representation of the toxicological effects of VF2 in animals. It can be concluded from the results that lifetime exposure to 10000 ppm does not induce effects of any toxicological significance for rats and mice, and does not increase incidence of any type of cancer in these species over their incidence in non-exposed animals.

No published data were retrieved on possible adverse health effects resulting from human exposure to VF2 in the more than 40 years of industrial use of the substance.

## 4. Hazards to the Environment

Data are not available concerning aquatic effects. However, exposure models determined that >99% of VF2 would be present in the air compartment. The Log Kow indicates that bioaccumulation is not likely. Thus, because of VF2's physical chemical properties, its nearly exclusive partitioning to air, and its use pattern, the conduct of aquatic toxicity testing was not necessary.

## 4.1. Aquatic effects

As explained above, no aquatic toxicity testing has been performed. However, a possible toxicity of VF2 was calculated with the ECOSAR program in the EPIWIN model and the details of this model are included in the dossier. The LC50, 96 hrs for fish is estimated to be 245 mg/l; the Daphnia LC50, 48 hrs is estimated to be 250 mg/l; .the green algae EC50, 96 hrs is estimated to be 149 mg/l.

Using the Mackay fugacity Model (level I and III)(See IUCLID dossier), it was determined that about 0.05% of discharged VF2 would be partitioned to the water compartment and less than 0.0002% would be partitioned to the sediment compartment.

In addition, QSAR modeling was performed with the structural analogues vinylidene chloride and 1,2 difluoroethylene. It should be noted that the approach to compare halogenated alkanes and alkenes is of limited reliability. Vinylidene chloride is a water-soluble liquid at ambient temperature (BP: 32°C, Water solubility: 2.5 gm/l) and vinylidene fluoride has a (BP: -83°C and water solubility: 250 mg/l). No ecotoxicological data were retrieved on 1,2 difluoroethylene. 1,1 Dichloroethylene was found to be only harmful to Crustaceans. The 48 hrL C50 for Daphnia magna was 11.6 mg/l. For both fish and algae the predicted aquatic toxicity values (LC/EC 50 > 100 mg/L) indicate low acute aquatic toxicity concern in these species.

## 4.2 Terrestrial effects

A possible toxicity of VF2 was calculated with the ECOSAR program in the EPI WIN model. The LC50 for the earthworm, 14 days, is calculated to be 675 mg/kg.

The Mackay Level I and III models calculated that >99% of the emitted VF2 would partition to the air compartment and negligible amounts to soil. The calculated half-life of VF2 in the atmosphere is 3.3 days. Therefore, VF2 is not expected to deposit onto soil and crops to any significant degree. Consequently, concentrations of VF2 to which certain mammals, birds and other terrestrial organisms could be exposed are considered to be negligible.

Effects on plants via atmospheric contact are not anticipated. The half-life of VDF in atmosphere is 3.3 to 4.7 days, which results in a complete transformation of VDF in atmosphere in a few weeks time.

Supposed ultimate degradation products of VF2 are CO2 and HF. The expected low VDF concentrations in the atmosphere in the vicinity of an emission source and the relatively short half life of the substance will render the representative testing of the effect of VDF on plants technically difficult and its results at least questionable when extrapolated to the natural environment.

## 4.3 Initial Assessment for the environment

Due to the physical-chemical properties of VF2, the conditions of its production and use (closed system intermediate), its predicted partitioning to air with negligible partitioning to other compartments and its predicted fast degradation, resulting environmental concentrations are considered to be very low.

No toxicity data for aquatic and terrestrial organisms have been generated, but are considered to be of low concern.

#### 5. Conclusions and Recommendations

## 5.1. Conclusions

#### **Conclusion for human health**

VF2 has been tested for acute toxicity, genotoxicity and for toxicity after repeated exposure, including its toxic potential on fertility, prenatal development reproduction as well as for its chronic toxicity and carcinogenic potential.

Although in earlier repeated exposure tests, some effects were observed on the hematopoietic system, the liver, the kidneys and the testes, these effects were not reproduced in more recent lifetime exposure tests in mice and rats. General signs of systemic toxicity were observed in rats and mice at 500 ppm in several 13-week inhalation studies. Overall, a LOEC of 500 ppm based on body weight, organ weight and clinical chemistry changes in the absence of histopathological changes can be identified for rats and mice. A NOEC of 250 ppm can be identified for rats. At 40,000 - 50,000 ppm effects on the nasal epithelium were noted in rats.

In genotoxicity studies, VF2 has shown some activity in bacterial assays, but was negative in the *in vitro* chromosomal aberration and gene mutation study in mammalian cells. *In vivo*, VF2 was negative in a mouse micronucleus and Drosophila SLRL test. Thus there is no evidence of genotoxicity *in vivo*.

The chronic toxicity/carcinogenicity tests and the tests on fertility and reproduction are the key studies for evaluation of the SIDS endpoints. It can be concluded from the results that for rats and mice, the lifetime exposure to 10000 ppm does not induce effects of any toxicological significance, and does not increase incidence of any type of cancer in these species over their incidence in non-exposed animals.

Since VF2 is produced and processed in closed systems, occupational exposure is low. Monitoring of occupational exposure to VF2 during production and processing indicates that typical average exposure concentrations are below10 ppm with incidental exposure above these values. Residents in the vicinity of production and processing plants are expected be exposed to much lower concentrations than those found on the work floor. In view of the use of VF2, production of various polymers with VF2 residual <1 ppm, consumer exposure is not expected.

## **Conclusion for the Environment**

Based on its physical-chemical properties and its predicted fate in the environment, VF2 is not expected to have significant toxic properties for aquatic organisms. Modeling data indicate a possible degradability in the aquatic compartment. However, its short residence time (few hours to few days) in the aquatic compartment will not lead to the formation of any significant metabolite concentration in that compartment. VF2 is only used for production of polymers in closed systems whereby emissions to the aquatic environment are not a concern.

Emissions of VF2 will partition almost completely to air and will be degraded swiftly by reaction with OH radicals (calculated half-life: 3.3 days). Therefore, atmospheric VF2 will not contribute significantly to global warming and does not constitute a significant concern for animals and man exposed via that compartment.

## 5.2. Recommendations

## Human health

The performed toxicity test programs on VF2 cover adequately all endpoints to be considered in this SIAR. Therefore no additional toxicity tests are recommended.

## Environment

No further environmental testing is deemed necessary for following reasons :

VF2 is a gas with limited water solubility and a low Log Kow.

Mackay level I and III modelling indicate that VF2 will partition nearly completely to air. Where aquatic and terrestrial toxicity testing and biodegradation testing is concerned following considerations are made. In order to obtain and maintain adequate concentrations in such test systems, specific containment techniques have to be used (semi static, flow through, closed vessels without head space) which may lead to artificial changes in itself (e.g. lack of CO2 in algal media, increased oxygen depletion in closed systems...), which in turn may lead to artifacts in the effects observed.

Moreover, such artificial conditions may not be at all representative of those in the natural environment.

Another important consideration is the pattern of use. The material is used only as a chemical intermediate in the production of polymers, in closed systems. VF2 is not expected to be released to water systems, but in any case emissions will partition to air >99%.

#### 6. References (1) http://toxnet.nml.nih.gov (2) Solvay (1993). (3)Dohany, J.E., Robb, L.E. (1980), Fluorine compounds, organic, Kirk-Othmer Encycl. Chem. Tech. 3rd ed., NY, Wiley INTRSCI Publ., 11, 65. Yaws, C.L., Yang, H.C., Hopper, J.R., Hansen, K.C. (July 1990), Organic (4) chemicals: water solubility data, Chemical Engineering, 115-118. Matheson Gas Data Book, 1966 (5) (6) Elf Atochem, Safety Data Sheet, 15.02.1995. Reliability criteria: not assignable: documentation insufficient for assessment (4e) (7) Elf Atochem, Safety Data Sheet, 15.02.1995. Mears, W.H., et al., Ind. Eng. Chem. vol 47 p 1449-1454, 1955 (8) (9) Solvav SDS. (10)Chou, J.T., Jurs, P.C. (1979), "Computer Assisted Computation of Partition Coefficients from Molecular Structures Using Fragment Constants", J. Chem. Inf. Comput. Sci., 19, 172-78. Solvay MSDS (11)Syracuse Research Corporation Calculated Values (1988) (12)Internal data Solvay, J. Franklin, 2001 (13)EPI WINN, (SRC model), EPA, 2001 (14)Pennwalt corporation, Unpublished data, Acute inhalation toxicity of vinylidene (15)fluoride, 1982 Carpenter, C.P., Smyth, H.F., Pozzani, U.C. (1949), The assay of acute vapor (16)toxicity, and the grading and interpretation of results on 96 chemical compounds, J. Ind. Hyg. Toxicol., 31(6), 343-346. Lester, D., Greenberg, L.A. (1950), A.M.A. Arch. Ind. Hyg. Occupational Med., 2, (17)335. Burgison, R.M. et al J. Pharmacol.Exp..Ther., vol 114, p 470-472, 1955 (18)(19) Reuzel, P.G., Beems, R.B., Dreef-van der Meulen, H.C., Willems, M.I., (1986), Sub-chronic (13-week) inhalation toxicity study of vinylidene fluoride in weanling and young adult rats, CIVO/TNO report no V86.321/250956 (20)Dupont (1977). US EPA Doc. no 878220582. Fiche no OTS0215306. Manus, A.G., Maloney, B.A., Craig, D.K., Keller, J.G (1984), Thirteen-week (21)subchronic study in F344 rats -Vinylidene fluoride - Final report, LBI Project no 12199-02 (NTP program). (22)Manus, A.G., Maloney, B.A., Craig, D.K., Keller, J.G. (1984), Thirteen-week subchronic study in B6C3F1 mice - Vinylidene fluoride - Final report, LBI Project no 12199-03 (NTP program). Appelman, L.M., Beems, R.B., Falke, H.E., Dreef-van der Meulen, H.C., Reuzel, (23)P.G. (1985), Sub-chronic (13-week) inhalation toxicity study of vinylidene fluoride in rats, CIVO/TNO report no V85.449/241407. (24)Newton, P.E. (1988), A two week inhalation toxicity study of vinylidene fluoride in the mouse, Bio/dynamics report project no 87-8035. (25)Newton, P.E. (1989), A thirteen week inhalation toxicity study of vinylidene fluoride in the mouse, Bio/dynamics report project 87-8021. Landry, M.M., Fuerst, R. (1968), Gas ecology of bacteria, Chapter 34, 370-80. (26)(27)Jagannath, D.R. (1977), Mutagenicity evaluation of Isotron 1132a, Litton Bionetics, LBI proje ct no 20838. (28)Matheson, D.W. (1978), Mutagenicity evaluation of Isotron 1132a in the in vitro

OECD SIDS

(29)	Russell, J.F. (1979), Mutagenic activity of ethylene, 1,1-difluoro- in the Salmonella/microsome assay. Haskell Lab. report no 729-78
(30)	Bartsch, H., Malaveille, C., Barbin, A., Planche, G. (1979), Mutagenic and alkylating metabolites of halo-ethylenes, chlorobutadienes and dichlorobutenes
(31)	Rickard, L.B. (1986), Mutagenicity evaluation of vinylidene fluoride in the
(22)	CHO/HPRT assay, Haskell Lab. report no 601-80.
(32)	vitro assay for chromosome aberrations in chinese hamster ovary (CHO) cells, Haskell Lab. report no 606-86.
(33)	Hodson-Walker, G., Mackay, J.M., Cracknell, S., Cowlyn, T., (1988), Vinylidene fluoride: assessment of clastogenic action on bone marrow erythrocytes in the micronucleus test. LSR report 88/0655.
(34)	Sernau, R.C. (1988), Mutagenicity test on vinylidene fluoride (1,1-difluoroethylene) Drosophila Melanogaster sex-linked recessive lethal test, Hazleton HLA study no
(35)	Sernau, R.C. (1989), Mutagenicity test on vinylidene fluoride (1,1-difluoroethylene)
	Drosophila Melanogaster sex-linked recessive lethal test, Hazleton HLA study no 10214-0-461. (Second Revision).
(36)	Arts, J.H., Bos-Kuijpers, M.H., Woutersen, R.A. (1991), Chronic
	toxicity/carcinogenicity inhalation study of vinylidene fluoride vapour in rats, CIVO/TNO report V91.039.
(37)	Newton, P.E. (1991), An inhalation oncogenicity study of vinylidene fluoride in the mouse, Bio/dynamics report project no 87-8022.
(38)	Maltoni, C., Tovoli, D. (1979), First experimental evidence of the carcinogenic effects of vinylidene fluoride. Med. Layoro, 5, 363-68
(39)	Koeter, H.B., van Marwijk, M.W., Reuzel, P.G. (1986), Fertility inhalation toxicity study with vinvlidene fluoride (VE2) in rats. CIVO/TNO report no 86 422/250957
(40)	Mecler, F.J., Beliles, R.P. (1978), Teratology study in rats Isotron 1132a - 1,1- difluoroethylene, Litton Bionetics, LBI project no 20891.
(41)	Solvay data.
(42)	Fluoroalkene Industry Group (1986).
(43)	Conolly, R., Szabo, S., Jaeger, R. (1979), Proc. Soc. Exp. Biol. Med., 162, 162-169.
(44)	Jaeger, RJ, et al., Arch.Environ.Health, vol 30, p 26-31, 1975
(45)	Bolt, H.M., Filser, J.G., et al., Arh.hig.rada.toksicol. vol 30, suppl. p 369-377, 1979
(46)	Baker, M.T., Bates, J.N., Leff, S.V. (1987), Comparative defluorination and cytochrome P-450 loss by the microsomal metabolism of fluoro- and fluorochloroethenes. Drug Metab Disp. 15(4), 499–503
(47)	Dilley, J.V., Carter, V.L., Harris, E.S. (1974), Fluoride ion excretion by male rats after inhalation of one of several fluoroethylenes or hexafluoropropene, Toxicol.
(48)	Filser I.G. Bolt H.M. et al. Toxicology letters vol 2 n 247-252 1978
(49)	Filser, J.G., et al., Arch. Toxicol., vol 49, n 107-116, 1982
(50)	Filser, J.G., Bolt, H.M., Arch, Toxicol., vol 45, p 109-116, 1980.
(51)	Stoeckle, G. et al., Toxicology letters, p 337-342, 1979
(52)	Zwart, A. (1985), Metabolic elimination of vinylidene fluoride vapour in rats, CIVO/TNO report no V85.082/241406
(53)	Medinsky, M.A., Bechtold, W.E., Birnbaum, L.S., Chico, D.M., Gerlach, R.F.,

Henderson, R.F. (1988), Uptake of vinylidene fluoride in rats simulated by a physiological model, Fund. Appl. Toxicol., 11(2), 250-260.

OECD SIDS	1,1-DIFLUOROETHYLENE
(54)	Medinsky, M.A., Bechtold, W.E., Birnbaum, L.S., Henderson, R.F. (1990), Measurement of steady-state blood concentrations in B6C3F1 mice exposed by
	inhalation to vinylidene fluoride, Toxicology, 64(2), 255-63.
(55)	Bechtold, W.E., Medinsky, M.A., Gerlach, R.F. (1988), The determination of a
	volatile gas, vinylidene fluoride, in blood during a nose-only exposure, J. Anal.
	Toxicology, 12, 62-66.
(56)	IARC monograph on the evaluation of carcinogenic risk of chemicals to humans,
	volume 39, 1986, p 227 to 235
(57)	US EPA RM-1 Risk Assessment Draft: Vinyl Fluoride (CAS No.75-02-5)
	Vinylidene Fluoride (CAS No. 75-38-7), Tetrafluoroethene (CAS No. 116-14-3),
	Hexafluoropropene (CAS No. 116-15-4) Trifluoroethene (CAS No. 359-11-5),
	3,3,3-Trifluoro-1-Propene (CAS No. 677-21-4), Leslie Scott, Branch/Division:
	High Production Volume Chemicals Branch/RAD, March 19, 1999
(58)	Solvay S.A. Bruxelles, unpublished data.
(59)	Howard, CJ, 1976 As cited in VDF results of EPIWIN (v3.05), US EPA Version
	July 12, 2000
(60)	Bolt, H.M., Filser, J.G., et al., Studies On Liver Microsomal Metabolism and
	Interaction of Vinyl Chloride and Related Compounds in Relation to Possible
	Carcinogenicity Arh.Hig.Rada.Toksicol. Vol 30, Suppl. P 369-377, 1979

## 7. Databases

Following databases were consulted to update and verify the available physico-chemical, toxicological and environmental information on VF2:

TOXLINE TOXLIT GENETOX CCRIS DART/ETIC EMIC HSDB RTECS AQUIRE GIABS MEDLINE TSCATS CHRIS NIOSHTIC DATALOG ENVIROFATE BIODEG BIOLOG PHYTOTOX TERETOX CHAPMAN & HALL MERCK INDEX

# IUCLIDData Set

Existing Chemical CAS No. EINECS Name EC No. Molecular Weight Molecular Formula	: ID: 75-38-7 : 75-38-7 : 1,1-difluoroethylene : 200-867-7 : 64.04 : CH2=CF2
Producer related part Company Creation date	: Solvay S.A. : 06.12.1994
Substance related part Company Creation date	: Solvay S.A. : 06.12.1994
Status Memo	: : JPE
Printing date	: 12.03.2002
Date of last update	12.03.2002
Number of pages	: 297
Chapter (profile) Reliability (profile) Flags (profile)	:

I GENERAL	INFORMATION	ld 75-38-7
I. ULIYLINAL		<b>Date</b> 12.03.2002
1.0.1 APPLIC	ANT AND COMPANY INFORMATION	
1.0.2 LOCATI	ON OF PRODUCTION SITE, IMPORTER OR FORMUL	ATOR
1.0.3 IDENTIT	Y OF RECIPIENTS	
1.0.4 DETAILS	S ON CATEGORY/TEMPLATE	
1.1.0 SUBS T	ANCE IDENTIFICATION	
1.1 GENER	AL SUBSTANCE INFORMATION	
Purity type Substance typ Physical statu Purity Colour Odour Remark	i constant of the second seco	ration with the following behalf: edex
04.05.1995	24	
I.I.Z SPECII		
Type of specti Remark	a : other: : Mass : 11 Reliability criteria: Not assignable: se (4b)	condary literature
14.03.2001	(12)	(1)
1.2 SYNON Difluoro 1.1 et 09.01,1995	YMS AND TRADENAMES	
<b>Difluoro 1.1 et</b> 09.01.1995	nylene	
Difluoroethene 22.03.1995		
<b>VF2</b> 09.01.1995		

OLCD SIDS
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1. GENERAL INFORMATION

#### 1.3 IMPURITIES

#### 1.4 ADDITIVES

#### 1.5 TOTAL QUANTITY

#### 1.6.1 LABELLING

Labelling :	provisionally by manufacturer/importer
Specific limits	
Nota :	, ,
R-Phrases	(10) Flammable
S-Phrases	(21) When using do not smoke
09.01.1995	

#### 1.6.2 CLASSIFICATION

Class	sified	:	provisionally by manufacturer/importer
Class of danger		:	flammable
R-Ph	rases	:	(10) Flammable
Speci	ific limits	:	
1 <sup>st</sup>	Concentration	:	
2 <sup>nd</sup>	Concentration	:	
3 <sup>rd</sup>	Concentration	:	
4 <sup>th</sup>	Concentration	:	
5 <sup>th</sup>	Concentration	:	
6 <sup>th</sup>	Concentration	:	
7 <sup>th</sup>	Concentration	:	
8 <sup>th</sup>	Concentration	:	
1 <sup>st</sup> .	Classification	:	
2 <sup>nd</sup>	Classification	:	
3 <sup>rd</sup>	Classification	:	
4 <sup>th</sup>	Classification	:	
<b>5</b> <sup>th</sup>	Classification	:	
6 <sup>th</sup>	Classification	:	
7 <sup>th</sup>	Classification	:	
8 <sup>m</sup>	Classification	:	
09.01	.1995		

#### 1.6.3 PACKAGING

#### 1.7 USE PATTERN

Type of use	: Type
Category	: Use in closed system
21.02.1995	
Type of use	: Industrial
Category	: Polymers industry
21.02.1995	
Type of use	: Use
Category	: Anti-static agents
21.02.1995	Ũ

## OECD SIDS

Date 12.03.2002

(2)

#### 1.7.1 DETAILED USE PATTERN

#### 1.7.2 METHODS OF MANUFACTURE

#### 1.8 REGULATORY MEASURES

1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

Type of limit	: other: SAEL
Limit value	: 1000 other: ppm
Remark	: SAEL = Solvay acceptable exposure limit in 1995.
08.12.2000	

#### 1.8.2 ACCEPTABLE RES IDUES LEVELS

#### 1.8.3 WATER POLLUTION

Remark	:	Because vinylidene fluoride is a gas under all environmental
		conditions, it is expected to be approx.100 % in air. See
		also derogation statement.

#### 08.12.2000

- 1.8.4 MAJOR ACCIDENT HAZARDS
- 1.8.5 AIR POLLUTION
- 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES
- 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS
- 1.9.2 COMPONENTS
- 1.10 SOURCE OF EXPOSURE

Remark : Country where Solvay production plant is located in EU: - France

14.03.1995

#### 1.11 ADDITIONAL REMARKS

Remark : MAK classification: IIIB (1983). 18.01.1995

#### 1.12 LAST LITERATURE SEARCH

1.13 REVIEWS

2 PHYSICO CHEMICA	ΑΙ ΠΑΤΑ	ld 75-38-7
		Date 12.03.2002
2.1 MELTING POINT		
Value	· = -144 ℃	
Remark	: Reliability criteria: reliable with restriction: har	ndbook
	(2g)	
18.06.2001		(3) (1)
2.2 BOILING POINT		
Value	· - 857 °C at 1013 bPa	
Reliability	: (4) not assignable	
13.03.2001	· / · · · · · · · · · · · · · · · · · ·	(3) (4)
	: =-85.7 °C at	
Kellability	: (4) not assignable	(2)
13.03.2001		(3)
Value	: =-83 °C at	
Remark	: Reliability criteria: reliable with restriction: hand	dbook
	(2g)	
18.06.2001		(1) (5)
Value	: -83 °C at	
Remark	: Reliability criteria: Reliable with restrictions: D	ata from
25.04.2004	handbook or collection of data (2g)	
25.04.2001		(5)
2.3 DENSITY		
2.3 DENSITY		
2.3 DENSITY Type	: Density	
2.3 DENSITY Type Value	: Density : = .92 g/cm <sup>3</sup> at -20 °C	
2.3 DENSITY Type Value Source 24.04.1995	: Density : = .92 g/cm³ at -20 °C : Elf Atochem	(6)
2.3 DENSITY Type Value Source 24.04.1995	: Density : = .92 g/cm³ at -20 °C : Elf Atochem	(6)
2.3 DENSITY Type Value Source 24.04.1995 Type	: Density : = .92 g/cm³ at -20 °C : Elf Atochem : Density	(6)
2.3 DENSITY Type Value Source 24.04.1995 Type Value	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> </ul>	(6)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1005	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> </ul>	(6)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> </ul>	(6) (7)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> </ul>	(6) (7)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Source 24.04.1995	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> </ul>	(6) (7)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Method	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> <li>other: calculated</li> </ul>	(6) (7)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> <li>other: calculated</li> <li>1955</li> </ul>	(6) (7)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Method Year GLP Tot substance	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> <li>other: calculated</li> <li>1955</li> </ul>	(6) (7)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Method Year GLP Test substance Remark	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> <li>other: calculated</li> <li>1955</li> <li>Liquid density</li> </ul>	(6) (7)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Method Year GLP Test substance Remark	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> <li>other: calculated</li> <li>1955</li> <li>Liquid density. Reliability criteria: reliable with restrictions:Acconstant</li> </ul>	(6) (7)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Method Year GLP Test substance Remark 20.08.2001	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> <li>other: calculated</li> <li>1955</li> <li>Liquid density. Reliability criteria: reliable with restrictions:Accordation method (2f)</li> </ul>	(6) (7) epted (8)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Method Year GLP Test substance Remark 20.08.2001 Type	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> <li>other: calculated</li> <li>1955</li> <li>Liquid density. Reliability criteria: reliable with restrictions:Accorcalculation method (2f)</li> </ul>	(6) (7) epted (8)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Method Year GLP Test substance Remark 20.08.2001 Type Value	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> <li>other: calculated</li> <li>1955</li> <li>Liquid density. Reliability criteria: reliable with restrictions: Acc calculation method (2f)</li> </ul>	(6) (7) epted (8)
2.3 DENSITY Type Value Source 24.04.1995 Type Value Source 24.04.1995 Type Value Method Year GLP Test substance Remark 20.08.2001 Type Value Remark	<ul> <li>Density</li> <li>= .92 g/cm<sup>3</sup> at -20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= .67 g/cm<sup>3</sup> at 20 °C</li> <li>Elf Atochem</li> <li>Density</li> <li>= 617 kg/m3 at 23.6 °C</li> <li>other: calculated</li> <li>1955</li> <li>Liquid density. Reliability criteria: reliable with restrictions:Accorcalculation method (2f)</li> <li>617 kg/m3 at 24 °C</li> <li>as liquified gas</li> </ul>	(6) (7) epted (8)

OECD SIDS	I,I-DIFLUOKOEIHYLE.
2.PHYSICO CHEMIC	AL DATA Id 75-38-7 Date 12 03 2002
	reliability criteria: reliable witrh restrictions: collection of data (2g)
14.03.2001	(1)
Туре	: relative density
Value	$= 2.2 \text{ at } 25 ^{\circ}\text{C}$
Remark	: Vapour density (air=1).
	reliability criteria: not assignable: documentation insufficient for assessment
	(4e)
10.05.1995	(9)
2.3.1 GRANULOMETRY	
2.4 VAPOUR PRESSU	URE
Malua	
value	= 33/00 NPa at 20 °C
Remark	: Reliability criteria: reliable with restrictions: data from
25.04.2001	Tandbook of collection of data.(29)
20.04.2001	(5)
2.5 PARTITION COEFF	FICIENT
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance	FICIENT : = 1.24 at °C : other (calculated) : 1979 :
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark	FICIENT : = 1.24 at °C : other (calculated) : 1979 : Reliability criteria: reliable with restrictions: acceptable
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark	FICIENT : = 1.24 at °C : other (calculated) : 1979 : : Reliability criteria: reliable with restrictions: acceptable calculation method (2f)
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark 18.06.2001	FICIENT 
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark 18.06.2001 2.6.1 SOLUBILITY IN DIF	FICIENT 
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark 18.06.2001 2.6.1 SOLUBILITY IN DIF	FICIENT
<ul> <li>2.5 PARTITION COEFF</li> <li>Partition coefficient Log pow pH value Method Year GLP Test substance Remark</li> <li>18.06.2001</li> <li>2.6.1 SOLUBILITY IN DIF</li> <li>Solubility in</li> </ul>	FICIENT
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark 18.06.2001 2.6.1 SOLUBILITY IN DIF Solubility in Value	FICIENT : = 1.24 at °C : other (calculated) : 1979 : : : Reliability criteria: reliable with restrictions: acceptable calculation method (2f) (10) FFERENT MEDIA
<ul> <li>2.5 PARTITION COEFF</li> <li>Partition coefficient Log pow pH value Method Year GLP Test substance Remark</li> <li>18.06.2001</li> <li>2.6.1 SOLUBILITY IN DIF</li> <li>Solubility in Value pH value</li> </ul>	FICIENT 
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark 18.06.2001 2.6.1 SOLUBILITY IN DIF Solubility in Value pH value concentration	FICIENT for the second secon
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark 18.06.2001 2.6.1 SOLUBILITY IN DIF Solubility in Value pH value concentration Temperature effects	FICIENT for the second secon
<ul> <li>2.5 PARTITION COEFF</li> <li>Partition coefficient Log pow pH value Method Year GLP Test substance Remark</li> <li>18.06.2001</li> <li>2.6.1 SOLUBILITY IN DIF</li> <li>Solubility in Value pH value concentration Temperature effects Examine different pol.</li> </ul>	FICIENT 
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark 18.06.2001 2.6.1 SOLUBILITY IN DIF Solubility in Value pH value concentration Temperature effects Examine different pol. pKa	FICIENT : = 1.24 at °C : other (calculated) : 1979 : Reliability criteria: reliable with restrictions: acceptable calculation method (2f) (10) FFERENT MEDIA : = 165 mg/l at 25 °C : at °C : at 25 °C
<ul> <li>2.5 PARTITION COEFF</li> <li>Partition coefficient Log pow pH value Method Year GLP Test substance Remark</li> <li>18.06.2001</li> <li>2.6.1 SOLUBILITY IN DIF</li> <li>Solubility in Value pH value concentration Temperature effects Examine different pol. pKa Description</li> </ul>	FICIENT = 1.24 at °C other (calculated) 1979 Reliability criteria: reliable with restrictions: acceptable calculation method (2f) (10) FFERENT MEDIA = 165 mg/l at 25 °C at °C at 25 °C
2.5 PARTITION COEFF Partition coefficient Log pow pH value Method Year GLP Test substance Remark 18.06.2001 2.6.1 SOLUBILITY IN DIF Solubility in Value pH value concentration Temperature effects Examine different pol. pKa Description Stable Examine state	FICIENT : = 1.24 at °C : other (calculated) : 1979 : Reliability criteria: reliable with restrictions: acceptable calculation method (2f) (10) FFERENT MEDIA : = 165 mg/l at 25 °C : at °C : at 25 °C

Deg. produci	•	
Method	: other: no data	
Year	: 1990	
GLP	: no data	
Test substance	:	
Remark	: Reliability criteria: reliable with restrictions: handbook	
	(2g)	
	Solubility at 1 atm.	
25.06.2001	-	(4)

UNEP PUBLICATIONS

OECD SIDS		1,1-DIFLUOROETHYLENE
2.PHYSICO CHEMICA	AL DATA	<b>ld</b> 75-38-7
		Date 12.03.2002
Solubility in Value pH value concentration Temperature effects Examine different pol.	: 180 mg/l at 25 °C at °C	

Reliability criteria: Not assignable: secondary literature

(1)

(9)

2.6.2	SURFACE TENSION

pKa Description

Stable

Method

Remark

18.06.2001

Year

GLP

Deg. product

Test substance

Test type	:	Other
Value	:	at °C
Concentration	:	
Remark	:	Is a gas at ambient temperature and atmospheric pressure.
13.03.2001		

#### 2.7 FLASH POINT

Remark	: None
	Reliability criteria: Not assignable: documentation
	insufficient for assessment (4e)
14.03.2001	

at 25 °C

other: no data

no data

(4b)

:

1

:

:

:

:

:

:

1

#### 2.8 AUTO FLAMMABILITY

Value : Method :	= 390 °C at other: no data
Year :	
GLP :	no data
Test substance :	
Remark	Reliability criteria: Not assignable: documentation insufficient for assessment (4e)
Reliability : 14.03.2001	(4) not assignable

UNEP PUBLICATIONS

#### 2.9 FLAMMABILITY

Result	:	flammable
Method	:	other: no data
Year	:	
GLP	:	no data

2.PHYSICO CHEM	ICAL DATA Id 75-38-7
	Date 12.03.2002
Test substance	
Test substance	
Remark	: Flammability limits in air : lower limit: 4.7 %
	upper limit : 25.1 %
	Reliability criteria: Not assignable: documentation
	insufficient for assessment (4e)
20.08.2001	(9)
2.10 EXPLOSIVE PF	ROPERTIES
Method	: other: no data
Year	
GLP	: no data
	. 110 Uala
rest substance	
Remark	: Explosive limits: 5.8-20.3 vol % in air.
	Reliability criteria: reliable with restrictions: handbook
	(2g)
Reliabilitv	: (4) not assignable
18 06 2001	(3)
10.00.2001	
2.11 OXIDIZING PRO Result Remark	PERTIES     no oxidizing properties     Reliability criteria: Not assignable: documentation
2.11 OXIDIZING PRO Result Remark	<ul> <li><b>DPERTIES</b></li> <li>: no oxidizing properties</li> <li>: Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> </ul>
2.11 OXIDIZING PRO Result Remark 19.06.2001	<ul> <li><b>DPERTIES</b> <ul> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> </ul> </li> <li>(11)</li> </ul>
2.11 OXIDIZING PRO Result Remark 19.06.2001 2.12 DISSOCIATION	<ul> <li>DPERTIES</li> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> <li>(11)</li> <li>N CONSTANT</li> </ul>
<ul> <li>2.11 OXIDIZING PRO</li> <li>Result Remark</li> <li>19.06.2001</li> <li>2.12 DISSOCIATION</li> <li>2.13 VISCOSITY</li> </ul>	<ul> <li>SPERTIES</li> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> <li>(11)</li> <li>N CONSTANT</li> </ul>
<ul> <li>2.11 OXIDIZING PRO Result Remark</li> <li>19.06.2001</li> <li>2.12 DISSOCIATION</li> <li>2.13 VISCOSITY</li> <li>2.14 ADDITIONAL F</li> </ul>	Substrain Service Servi
<ul> <li>2.11 OXIDIZING PRO Result Remark</li> <li>19.06.2001</li> <li>2.12 DISSOCIATION</li> <li>2.13 VISCOSITY</li> <li>2.14 ADDITIONAL F Remark</li> </ul>	<ul> <li>SPERTIES</li> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> <li>(11)</li> <li>N CONSTANT</li> <li>REMARKS</li> <li>Henry's Law Constant: 0.226 atm.m3/mole Based on the EPIWIN model (SBC)</li> </ul>
<ul> <li>2.11 OXIDIZING PRO Result Remark</li> <li>19.06.2001</li> <li>2.12 DISSOCIATION</li> <li>2.13 VISCOSITY</li> <li>2.14 ADDITIONAL FOR Remark</li> </ul>	<ul> <li>SPERTIES</li> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e) (11)</li> <li>N CONSTANT</li> <li>REMARKS</li> <li>Henry's Law Constant: 0.226 atm.m3/mole Based on the EPIWIN model (SRC). Henry's law constant: 5.09E-3 atm Cu M/mole.</li> </ul>
<ul> <li>2.11 OXIDIZING PRO Result Remark</li> <li>19.06.2001</li> <li>2.12 DISSOCIATION</li> <li>2.13 VISCOSITY</li> <li>2.14 ADDITIONAL FOR Remark</li> </ul>	<ul> <li>PERTIES</li> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e) (11)</li> <li>A CONSTANT</li> <li>REMARKS</li> <li>Henry's Law Constant: 0.226 atm.m3/mole Based on the EPIWIN model (SRC). Henry's law constant: 5.09E-3 atm Cu M/mole. Reliability criteria: reliable with restrictions: accepted calculation method (2f)</li> </ul>
<ul> <li>2.11 OXIDIZING PRO Result Remark <ol> <li>19.06.2001</li> </ol> </li> <li>2.12 DISSOCIATION</li> <li>2.13 VISCOSITY</li> <li>2.14 ADDITIONAL FOR Remark <ol> <li>Remark</li> </ol> </li> </ul>	<ul> <li>SPERTIES</li> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> <li>(11)</li> <li>A CONSTANT</li> <li>REMARKS</li> <li>Henry's Law Constant: 0.226 atm.m3/mole Based on the EPIWIN model (SRC). Henry's law constant: 5.09E-3 atm Cu M/mole. Reliability criteria: reliable with restrictions: accepted calculation method (2f) 12)</li> </ul>
2.11 OXIDIZING PRO Result Remark 19.06.2001 2.12 DISSOCIATION 2.13 VISCOSITY 2.14 ADDITIONAL F Remark 10.03.1995 Remark	<ul> <li>SPERTIES</li> <li>I no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> <li>(11)</li> <li>A CONSTANT</li> <li>REMARKS</li> <li>Henry's Law Constant: 0.226 atm.m3/mole Based on the EPIWIN model (SRC). Henry's law constant: 5.09E-3 atm Cu M/mole. Reliability criteria: reliable with restrictions: accepted calculation method (2f) 12)</li> <li>Critical temperature: Tc = 30 degC</li> </ul>
2.11 OXIDIZING PRO Result Remark 19.06.2001 2.12 DISSOCIATION 2.13 VISCOSITY 2.14 ADDITIONAL R Remark 10.03.1995 Remark	Section 2         Image: section 2
2.11 OXIDIZING PRO Result Remark 19.06.2001 2.12 DISSOCIATION 2.13 VISCOSITY 2.14 ADDITIONAL F Remark 10.03.1995 Remark	<ul> <li>SPERTIES</li> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> <li>(11)</li> <li>N CONSTANT</li> <li>REMARKS</li> <li>Henry's Law Constant: 0.226 atm.m3/mole Based on the EPIWIN model (SRC). Henry's law constant: 5.09E-3 atm Cu M/mole. Reliability criteria: reliable with restrictions: accepted calculation method (2f) 12)</li> <li>Critical temperature: Tc = 30 degC Critical pressure : Pc = 4.43 MPa Solution is obtained actional attention</li> </ul>
2.11 OXIDIZING PRO Result Remark 19.06.2001 2.12 DISSOCIATION 2.13 VISCOSITY 2.14 ADDITIONAL R Remark 10.03.1995 Remark	<ul> <li>SPERTIES</li> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> <li>(11)</li> <li>A CONSTANT</li> <li>REMARKS</li> <li>Henry's Law Constant: 0.226 atm.m3/mole Based on the EPIWIN model (SRC). Henry's law constant: 5.09E-3 atm Cu M/mole. Reliability criteria: reliable with restrictions: accepted calculation method (2f) 12)</li> <li>Critical temperature: Tc = 30 degC Critical pressure : Pc = 4.43 MPa Soluble in chlorinated solvents, ethanol, ether.</li> </ul>
2.11 OXIDIZING PRO Result Remark 19.06.2001 2.12 DISSOCIATION 2.13 VISCOSITY 2.14 ADDITIONAL R Remark 10.03.1995 Remark Source	Section 2       Image: constant of the section of the se
2.11 OXIDIZING PRO Result Remark 19.06.2001 2.12 DISSOCIATION 2.13 VISCOSITY 2.14 ADDITIONAL R Remark 10.03.1995 Remark Source 24.04.1995	<ul> <li>PERTIES</li> <li>no oxidizing properties</li> <li>Reliability criteria: Not assignable: documentation insufficient for assessment (4e)</li> <li>(11)</li> <li>N CONSTANT</li> <li>REMARKS</li> <li>Henry's Law Constant: 0.226 atm.m3/mole Based on the EPIWIN model (SRC). Henry's law constant: 5.09E-3 atm Cu M/mole. Reliability criteria: reliable with restrictions: accepted calculation method (2f) 12)</li> <li>Critical temperature: Tc = 30 degC Critical pressure : Pc = 4.43 MPa Soluble in chlorinated solvents, ethanol, ether.</li> <li>Elf Atochem</li> </ul>

#### 3.1.1 PHOTODEGRADATION

INDIRECT PHOTOLYSIS Sensitizer Conc. of sensitizer Rate constant Degradation Remark		O3 70000000000 molecule/cm <sup>3</sup> .0000000000000000028 cm <sup>3</sup> /(molecule*sec) % after T 1/2: 40.9 days Calculated with SRC calculation method, P/C data as mentioned in the
		IUCLID data set.
<b>Source</b> 21.08.2001	:	Reliability criteria: reliable with restrictions: accepted calculation method (2f) EPI WINN, (SRC method), EPA
INDIRECT PHOTOLYSIS Sensitizer Conc. of sensitizer Rate constant Degradation	: :	OH 1500000 molecule/cm <sup>3</sup> .000000000022667 cm <sup>3</sup> /(molecule*sec) % after
Remark	:	T 1/2: 4.7 days
		Calculated with the SRC model. P/C data as mentioned in the IUCLID data base
<b>Source</b> 21.08.2001	:	Reliability criteria:reliable with restrictions: accepted calculation method (2f) EPI WINN, (SRC model), EPA, 2001
Conc. of substance DIRECT PHOTOLYSIS Halflife t1/2 Degradation Quantum yield INDIRECT PHOTOLYSIS Sensitivar	: : : : : : : : : : : : : : : : : : : :	at 25 °C = 3.3 day(s) % after
Conc. of sensitizer Rate constant Degradation Deg. product Method Year	• • • • • • • • • • • • • • • • • • • •	cm³/(molecule*sec) % after other (calculated) 2001
GLP Test substance Remark	:	as prescribed by 1.1 - 1.4 The atmospheric lifetime of VF2 has been calculated by the Prather and Spivakovsky procedure for deriving the tropospheric OH lifetime of a gven compound by scaling to that of CCI3CH3 at the temp of 277°K
		For VF2 this atmospheric T1/2 is calculated to be $3.3$ days.
Reliability 20.08.2001	:	Reliability criteria: reliable with restriction: acceptable calculation method (2f) (3) invalid (13)

1,1-DIFLUOROETHYLENE Id 75-38-7

#### 3.1.2 STABILITY IN WATER

Deg. product Method Year	: other: :
GLP Test substance Remark	: : : No experimental data are available.
	Half-life in surface waters was calculated by the appropriate model provided by the Syracuse Research Corporation. Half-life (by evaporation) in river water is 0.9 hrs; from lake water is 77 hours
	P/C data as mentioned in the IUCLID data set.
	River Lake
	Depth in m 1 1 Wind veloc. m/sec 5 0.5 Current veloc. m/sec 1 0.05
21.08.2001	
3.1.3 STABILITY IN SOIL	
Type Radiolabel Concentration Soil temperature Soil humidity Soil classification Year Remark 14.03.2001	<ul> <li>other:</li> <li>°C</li> <li>No experimental data are available, but due to its low boiling point and its low logP value, the compound will evaporate from soil and collect in the atmosphere. (see also biodegradation and Mackay modeling.)</li> </ul>
3.2.1 MONITORING DATA	
3.2.2 FIELD STUDIES	
3.3.1 TRANSPORT BETWE	EN ENVIRONM ENTAL COMPARTMENTS
Type Media Air Water Soil Biota Soil Method	<ul> <li>Volatility</li> <li>other: air</li> <li>% (Fugacity Model Level I)</li> <li>% (Fugacity Model Level I)</li> <li>% (Fugacity Model Level I)</li> <li>% (Fugacity Model Level II/III)</li> <li>% (Fugacity Model Level II/III)</li> </ul>
	LINED DUDI ICATIONS

LIN VIROLAUTEIN I AL I	ΓΑΤΕ ΑΝΟ ΡΑΤΗΨΑνς	ld 75-38-7
	SATE AND FAITWATS	Date 12.03.2002
Year	:	
Remark	: Henry's law constant: 1270 kPa m3/mol (20 d	legC calculated).
10.05.1995		(9)
Туре	: Volatility	
Media	: other: water	
Air	: % (Fugacity Model Level I)	
Water	: % (Fugacity Model Level I)	
Soil	: % (Fugacity Model Level I)	
Biota	: % (Fugacity Model Level II/III)	
Soil	: % (Fugacity Model Level II/III)	
Method	:	
Year	:	
Remark	: Water evaporation: t 1/2, 23 hours (20 degC c	alculated).
10.05.1995		(9)
Remark	: Because vinylidene fluoride is a gas under all	environmental
	conditions, it is expected to be 100 % in air.(s	ee also
	Mackay modeling)	
14.03.2001		
Media Method	: other: soil adsorption : other (calculation)	
Media Method Year	: other: soil adsorption : other (calculation) :	
Media Method Year Remark	<ul> <li>other: soil adsorption</li> <li>other (calculation)</li> <li>Reliability criteria: Reliable with restrictions: a (2f)</li> </ul>	accepted calculation method
Media Method Year Remark Result	<ul> <li>other: soil adsorption</li> <li>other (calculation)</li> <li>Reliability criteria: Reliable with restrictions: a (2f)</li> <li>Koc = 7.12 (calculated on the basis of MW: 64)</li> </ul>	accepted calculation method 4.O3, Water solubility of 250
Media Method Year Remark Result	<ul> <li>other: soil adsorption</li> <li>other (calculation)</li> <li>Reliability criteria: Reliable with restrictions: a (2f)</li> <li>Koc = 7.12 (calculated on the basis of MW: 64 mg/l, BP: -83°C, MP: -144°C)</li> </ul>	accepted calculation method 4.O3, Water solubility of 250
Media Method Year Remark Result 12.03.2002	<ul> <li>other: soil adsorption</li> <li>other (calculation)</li> <li>Reliability criteria: Reliable with restrictions: a (2f)</li> <li>Koc = 7.12 (calculated on the basis of MW: 64 mg/l, BP: -83°C, MP: -144°C)</li> </ul>	accepted calculation method 4.O3, Water solubility of 250 (14)
Media Method Year Remark Result 12.03.2002 MODE OF DEGRADA	<ul> <li>other: soil adsorption</li> <li>other (calculation)</li> <li>Reliability criteria: Reliable with restrictions: a (2f)</li> <li>Koc = 7.12 (calculated on the basis of MW: 64 mg/l, BP: -83°C, MP: -144°C)</li> </ul>	accepted calculation method 4.O3, Water solubility of 250 (14)
Media Method Year Remark Result 12.03.2002 MODE OF DEGRADA	<ul> <li>other: soil adsorption</li> <li>other (calculation)</li> <li>Reliability criteria: Reliable with restrictions: a (2f)</li> <li>Koc = 7.12 (calculated on the basis of MW: 64 mg/l, BP: -83°C, MP: -144°C)</li> </ul>	accepted calculation method 4.O3, Water solubility of 250 (14)
Media Method Year Remark Result 12.03.2002 MODE OF DEGRADA BIODEGRADATION	<ul> <li>other: soil adsorption</li> <li>other (calculation)</li> <li>Reliability criteria: Reliable with restrictions: a (2f)</li> <li>Koc = 7.12 (calculated on the basis of MW: 64 mg/l, BP: -83°C, MP: -144°C)</li> </ul>	accepted calculation method 4.O3, Water solubility of 250 (14)
Media Method Year Remark Result 12.03.2002 MODE OF DEGRADA 5 BIODEGRADATION Remark	<ul> <li>other: soil adsorption</li> <li>other (calculation)</li> <li>Reliability criteria: Reliable with restrictions: a (2f)</li> <li>Koc = 7.12 (calculated on the basis of MW: 64 mg/l, BP: -83°C, MP: -144°C)</li> </ul> <b>ATION IN ACTUALUSE IN O experimental data are available.</b>	accepted calculation method 4.O3, Water solubility of 250 (14)

Primary and ultimate biodegradation is 3.7 to 3 and indicates biodegradation in a period of weeks.

In an STP, the biodegradation accounts for 0.04 % of the removal of the original imput.(63 % evaporates)

 Reliability criteria: reliable with restrictions: accepted calculation model (2f)

 Source
 :
 EPI WINN (SRC model) , EPA, 2001

21.08.2001

OECD SIDS	
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#### 3. ENVIRONMENTAL FATE AND PATHWAYS

1,1-DIFLUOROETHYLENE d 75-38-7

## 3.6 BOD5, COD OR BOD5/COD RATIO

Remark	:	No experimental data are available.
21.08.2001		-

#### 3.7 BIOACCUMULATION

BCF	:	= 1.8
Elimination	:	
Method	:	other: calculated with SRC model
Year	:	
GLP	:	
Test substance	:	
Remark	:	No experimental data are available.
		Reliability criteria: Reliable with restrictions: accepted calculation method

21.08.2001

(15) (15)

#### 3.8 ADDITIONAL REMARKS

21.08.2001

Date 12.03.2002

#### 4.1 ACUTE/PROLONGED TOXICITY TO FISH

Туре	: other: calculated
Species	: other: fish
Exposure period	: 96 hour(s)
Unit	: mg/l
LC50	: ca. 246
Remark	: no experimental data available.
Source	: EPI WINN, (SRC model), EPA, 200
21.08.2001	

#### 4.2 ACUTE TOXICITY TO AQUATIC INVERTEB RATES

Type Species Exposure period Unit EC50 Remark	:	other: calculated Daphnia sp. (Crustacea) 48 hour(s) mg/l ca. 250 no experimental data are available.
<b>Source</b> 21.08.2001	:	Reliability criteria; reliable with restrictions: accepted calculation method (2f) EPI WINN, (SRC model), EPA, 2001
Type Species Exposure period Unit EC50 Remark	:	other: calculated other: mysid shrimp 96 hour(s) mg/l ca. 122 No experimental data are available.
<b>Source</b> 21.08.2001	:	Reliability criteria: Reliable with restrictions: accepted calculation method (2f) EPI WINN, (SRC model), EPA, 2001

#### 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species	
Endpoint	: other: calculated
Exposure period	: 96 hour(s)
Unit	: mg/l
EC50	: ca. 149
Method	: other:
Year	:
GLP	:
Test substance	:
Remark	: No experimental data are available.
<b>Source</b> 21.08.2001	Reliability criteria: reliable with restrictions: accepted calculation method (2f) : EPI WINN, (SRC model), EPA, 2001
OECD SIDS	
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4. ECOTOXICITY	

### 4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

: no data.

Remark	
21.08.2001	

# 4.5.1 CHRONIC TOXICITY TO FISH

Method	:	other
Year	:	
GLP	:	
Test substance	:	
Remark	:	no experimental data.
21.08.2001		

#### 4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

Method	:	other
Year	:	
GLP	:	
Test substance	:	
Remark	:	no experimental data.
21.08.2001		-

#### 4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

## 4.6.2 TOXICITY TO TERRESTRIAL PLANTS

Method	:	other
Year	:	
GLP	:	
Test substance	:	
Remark 21.08.2001	:	no data.

## 4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

Type Species Endpoint Exposure period Unit LC50 Method Year GLP Test substance		other: calculated other: earthworm 14 day(s) Mg/kg soil dw ca. 675 other
Remark	:	no experimental data.
<b>Source</b> 21.08.2001	:	Reliability criteria: reliable with restrictions: accepted calculation method (2f) EPI WINN, (SRC model), EPA, 2001

### 4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

Method	:	other
Year	:	
GLP	:	
Test substance	:	
Remark	:	no data
21.08.2001		

# 4.7 BIOLOGICAL EFFECTS MONITORING

Remark	:	No data.
22.03.1995		

#### 4.8 BIOTRANSFORMATION AND KINETICS

Remark	:	No data.
22.03.1995		

## 4.9 ADDITIONAL REMARKS

Remark	: Due to specific physico chemical properties of VF2, no aquatic nor terrestrial toxicity testing has been performed. In order to obtain and maintain adequate concentrations in such test systems, specific containment techniques have to be used (semi static, flow through, closed vessels without head space) which may lead to artefactal changes in itself (e.g. lack of CO2 in algal media, increased oxygen depletion in closed systems), which in turn may lead to artefacts in the effects observed.
21.08.2001	Moreover, such artificial conditions may not be at all representative of those in the natural environment. (e.g. maintained test concentrations during fixed exposure times versus naturally decreasing concentrations by more than half in less than 24 hrs) Another important consideration is the pattern of use. The material is used only as in manufacturing as a chemical intermediate in the production of polymers, in closed systems. It is not expected to be released to water systems, and it will not partition to the water from the air.
Z1.U8.ZUU1	

OECD SIDS
5. TOXICITY

Date 12.03.2002

## 5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

In Vitro/in vivo Type Species Number of animals Males Females	: In vivo : Metabolism : Rat
Doses Males Females	
Vehicle Remark	<ul> <li>Male Holtzman rats were exposed to VF2 concentrations ranging from 0 to 25000 ppm for 4 to 6 hrs. Animals were either not pretreated, or pretreated with phenobarbital or with PCB.</li> <li>The goal was to evaluate the hepatotoxicity of VF2 by measurement of liver weight, serum sorbital dehydrogenase (SDH) activity and by histology of the liver.</li> <li>No effects were seen in non-pretreated animals nor in animals pretreated with phenobarbital.</li> <li>4-Hrs exposure of animals pretreated with PCB caused increased liver weight at all concentrations. (statistically significant at 15000 and 25000 ppm , but not at 500 ppm). The increase was not really dose-dependent.</li> <li>SDH values were significantly elevated in a dose-dependent fashion at all test concentrations.</li> <li>Histologically, the 15000 + PCB and 25000 ppm + PCB group showed balloonig of hepatocytes, centrilobular and mide-zonal necrosis surrounded by hemorrhagic and inflammatory zones.</li> <li>Lesions were considered to be distinctly different from those seen in similar tests with VCM.</li> <li>This is not considered to be a key study for SIDS evaluation and is therefore not further detailed in this document.</li> </ul>
21.11.2001	assessment.(2e)
In Vitro/in vivo Type Species Number of animals Males Females Doses Males	: In vivo : Metabolism : rat :
Females Vehicle Remark	: : Fed and fasted male Holzman rats were exposed to 26000 and

TOVICITY		₩ 75_99_7
IUXICITY		Date 12.03.2002
		<b>Dute</b> 12.00.2002
		82000 ppm of VE2 for 4 hrs (5 animals per feeding status and
		per test concentration. Controls were exposed to air.)
		Actual exposure concentrations were monitored with gas
		chrom atography
		The henatic enzyme alanine, alpha-ketoolutarate transaminase
		(AKT) in all animals wore measured after exposure
		(ART) in all animals were measured after exposure.
		Exposure to 82000 ppm had no appreciable effect at 24 hrs
		on serum AKT activity either in led or lasted rats. Althuogh
		there was a small statistically significant rise in serum
		AK I among fed rats after the 82000 ppm exposure, it was not
		considered to be a meaningfull elevation.
		I his concentration was without apparent central nervous
		system depressant effect.
		No mortality occurred during exposure.
		Reliability criteria: reliable with restrictions: xell
		documented, acceptable scientific principles, acceptable for
		assessment.
21.11.2001		
he \ //4 // /		
	:	IN VILTO
iype	:	
Species		rat
Number of ani	mals	
M	ales :	
_ F	emales :	
Doses		
N	ales :	
F.	emales :	
Vehicle	:	
Remark	:	Rat liver microsomes exposed to concentrations up to 7 kPa
		(partial pressure) VF2 in the gas phase.
		VF2 showed a progressive inhibition of aminopyrine
		demethylation with increasing VF2 concentrations.
		The authors conclude that VF2 may persist at the enzymic
		site and thus exhibit inhibition of oxygenation of drugs and
		xenobiotics.
		This is not a key study for SIDS evaluation and is therefore
		not further detailed in this document.
		Reliability criteria: reliable with restrictions: well
		documented, acceptable scientific principles, acceptable for
		assessment (2e)
21.11.2001		
In \/itera/ins - in	_	la vitra
	-	III viuo Metabolism
i ype Species		metabolion rat
Number of and	male	ιαι
	aies :	
Desca	entales :	
DOSES		
	aits :	
Mahiata Mahiata	emales :	
venicie	:	
<b>D</b> ·	•	Rat penatic microsomes(upprimed or primed with beta
Remark	•	Nat repaid microsofted (unprinted of printed with beta
Remark	•	naphtoflavone or phenobarbital) were exposed to variuos
Remark	·	naphtoflavone or phenobarbital) were exposed to variuos gaseous VF2 concentrations up to 60 $\mu$ M for up to 30 minutes.

ECD SIDS	1,1-DIFLUOROETHYLENE
TOXICITY	ld 75-38-7 Date 12.03.2002
	The capacity of VF2 to destroy P 450 and heme were measured.
	The release of Fluor ion was minimal (to approx. 3 nM F- /
	nM P 450) in uprimed as well as in primed microsomes at
	different times and at different concentrations of VF2.
	VF2 inactivates P 450 and heme noly tyo 17% under the
	conditions of these test.
	This is not considered a key study in the SIDS evaluation
	and is therefore not further detailed in this document.
	Reliabilty criteria: reliable with restrictions: well
	documented, acceptable scientific principles, acceptable for
21 11 2001	assessment.(2e)
21.11.2001	
In Vitro/in vivo -	: In vivo
Type Species	: Metabolism
Number of animals	. 101
Males	:
Females	:
Doses	
Females	
Vehicle	
Remark	: Male sprague-Dawley rats were exposed for 30 minutes to
	2200 ppm VF2.
	during days 1 to 7 post exposure.
	Urinary potassium excretion was only increase on day 6 after
	exposure.
	Creatine excretion and urinary volume were not significantly
	Kidneys showed macroscopic changes: marked hyperemia of the
	medulla, pazle whitish band in the cortex near the
	corticomedullary junction.
	No histological changes.
	Since this is not a key study no more details of the s tudy
	have been described in this document.
	Reliability criteria: Reliable with restrictions: well
21.11.2001	documented, meets scietific principles (2e)
In Vitro/in vivo	
Type	: Metabolism
Species	: rat
Number of animals	
Males	
Females Doses	:
Males	:
Females	:
Vehicle	
Remark	: Male Sprague-Dawley rats were whole body exposed to
	DUU PPIN VEZ TOR & NRS. Actual test concentrations were monitored by GC -MS
	UNEP Publications 41

OECD SIDS	1,1-DIFLUOROETHYLENE
5. TOXICITY	ld 75-38-7
	<b>Date</b> 12.03.2002
	Animals exhaled acetone at amounts of between 1.2 and 2.5 $\mu mol/hr/Kg$ body weight. (controls: 0.05 $\mu M/hr/kg)$
	Subsequently similar experiments were conducted by the same group that corroborated and refined the previous findings.
	Metabolism of VF2 after inhalation seems to be saturated at 100 ppm concentration
	The Vmax for metabolism of VF2 was calculated to be 1.1 $\mu$ M/hr/kg BW.
	The authors conclude that VF2 may interact with hepatic microsomal cytochrome P 450.
	The acetone formation may partly be explained by the
	Filser et al., (1980) put forward the hypothesis that all halogenated
	ethylenes share following metabolic steps:
	-rearrangement to the halogenated acetaldehyde or halogenated acyl halide
	- partial transformation of the latter to halogenated acetic acid
	formation is that the halogenated acetic acid formed(fluoroacetic acid)
	inhibits the enzymes of the citric acid cycle as suggested by others.
	These studiesare not considered key studies for SIDS evaluation .
	Reliability criteria: reliable with restriction: well
	documented, acceptable scientific principles, acceptable
21.11.2001	for assessment (2e)
in vitro/in vivo Tvpe	: In vivo : Metabolism
Species	: rat
Number of animals	
Males	
Doses	
Males	:
Females	:
venicie Remark	: Female newborn Wistar rats were whole-body exposed to 2000
	ppm for 4,6,8 and 10 weeks (8hr/d, 5d/wk).
	A replicate was done with same strain at the same dosage and
	at the same regimention o, 10, 12 and 14 weeks. Recovery, period for both replicates was 2 weeks
	Negative controls were run in a similar way.
	Test concentration was monitored with GC.
	ATPase content in liver cells was evaluated.
	After 14 weeks of exposure the number of A I Pase deficient
	the number of foci observed after 4 weeks exposure to VCM.
	In an experiment with Wistar rats exposed to VF2 at
	different concentrations, it was concluded that the metabolisation rate of VE2 was approx 1.1 µM /br/kg BW
	This is not a keystudy for SIDS evaluation and is therefore
10	
-2	UNEP Publications

ECD SID	S	1,1-DIFLUC	DROETHYLENE
TOXICI	ГҮ	ld 7 Date 1	75-38-7 12 03 2002
		Date	12.03.2002
		not further detailed in this document.	
		Reliability criteria: reliable with restrictions: well	
		assessment (2e)	
21.11.2001	1		
In Vitro/in 1			
in vitro/in \ Type	VIVO	: In vivo · Metabolism	
Species		: rat	
Number of	f animals		
	Males		
Doses	Females	:	
	Males	:	
	Females	:	
Vehicle		:	
Remark		: Male Sprague-Dawlev rats were exposed for 6 hrs to VF2	
		concentrations ranging from 20 to 1800 ppm.	
		It appeared that the maximum metabolism of $VE2$ vancur in	
		these rats is reached at a level of approx. 400 ppm using	
		the calculation method of Filser and Bolt.	
		This is not a kny at the for SIDS and with a state for	
		not further detailed in this document.	
		Reliability criteria: reliable without restriction: in	
		accordance with generally accepted scientific standards and described in sufficient detail (1d)	
21.11.2001	1		
In Vitro/in v	vivo	: In vivo	
Туре		: Toxicokinetics	
Species		: rat	
Number of	t animals		
	iviales Females		
Doses			
	Males	:	
Vehicle	Females		
Remark		: Male Fisher344/N rats were exposed nose only for 6 hrs to	
		concentrations of VF2 ranging from 30 to 16000 ppm.	
		Actual test concentrations were monitored by GC every 10	
		minutes.	
		Tidal volume and breathing frequency were not affected by	
		any concentration.	
		Experimentally determined steady-state blood levels of VF2	
		(GC) increased linearly with increasing exposure	
		concentration from 15 ng VF2/ ml (30 ppm) to 2400 ng VF2/ m	l
		at 16000 ppm.	
		1.0, and 0.29 for water, blood, liver, fat and muscle	
		respectively.	
		These values and published values for metabolism of VE2 Kr	n
			42
		UNEP Publications	43

OECD SIDS		1,1-DIFLUOROETHYLENE	
5. TOXICI	ГҮ	ld 75-38-7	
		Date 12.03.2002	
		<ul> <li>and Vmax, were incorporated in a physiological model.</li> <li>Model predictions agreed with experimentally determined data. Time to Cmax was less than 15 minutes for all concentrations.</li> <li>After end of exposure, VF2 blood levels decreased to 10% of Cmax in 1 hr.</li> <li>Simulation of the metabolism of VF2 indicated that the amount of VF2 metabolised per 6 hrs exposure period increased with increasing concentration of VF2 approaching a maximum at about 2000 ppm VF2.</li> <li>This air concentration results in a blood level of 260 ng VF2/ml at steady state.</li> <li>This is not considered a key study in the SIDS evaluation and is therefore not further detailed in this document.</li> <li>Reliability criteria: reliable with restrictions: well documented, acceptable scientific principles, acceptable for assessment (2e)</li> </ul>	
21.11.2001			
In Vitro/in v Type Species Number of Doses	rivo f animals Males Females	: In vivo : Toxicokinetics : mouse :	
D0363	Males	:	
	Females	:	
Vehicle Remark		: Male B6C3F1 mice were exposed to VF2 concentrations of 250, 3750 and 15000 ppm for 6 hrs (nose only). Exposure concentrations were monitored by GC at 10 minutes intervals. (No control was exposed).	
		Concentrations of VDF were measured in the blood. A physiological model to simulate blood levels of VF2 in	
		rats was adapted to mice by incormorating physiologcally realistic parameters for mice where appropriate. and by assuming that chemical specific parameters such as blood tissue partition coefficients determined for rats could also be applied to mice. Measured steady state levels of VF2 in blood of mice increased with increasing exposure concentration.	
		For both the 3750 and 15000 ppm VF2 exposures, the experimentally determined fell within the 95% confidence interval predicted by the physiological model. For the 250 ppm VF2 exposure, the experimentally determined values for VF2 in blood were lower than predicted by the model. At the lowest concentration it was not possible to detect	
		VF2 in blood taken 15 minutes after cessation of exposure, (suggesting that VF2 concentrations were below the LOD of 4 ng VF2/ml). For 15000 ppm exposure, VF2 could be detected in blood at 15 minutes post exposure.	
		This is not considered at key study for SIDS evaluation and	

OFCD SIDS	1 1 DIEL LIODOETLIVI ENE
5 TOVICITY	I,I-DIFLUOROETHYLENE
J. TUAICITI	
	is therefore not further detailed in this document.
	Reliability criteria: reliable with restrictions: well documented, acceptable scientific principles, acceptable for assessment.(2e)
21.11.2001	
In Vitro/in vivo	: In vivo
Туре	: Toxicokinetics
Species	: rat
Number of animals	
Males	:
Females	:
Doses	
Wales Econolog	
remaies Vehicle	
Remark	E344 rats were exposed to VE2 concentrations of 3750 and
	5000 ppm for 6 hrs inorder to develop a new method for measurnig VF2 in blood.
	At 3750 ppm and 5000 ppm the steady-state blood concentrations of VF2 were 240 and 626 ng/ml.
	This is not considered to be a key study for SIDS evaluation and is therefore not further detailed in this document.
	Reliability criteria: reliable with restrictions: well documented, acceptable scientific principles, acceptable for assessment (2e)
21.11.2001	233535ment.(26)
5.1.1 ACUTE ORAL TOXIC	CITY
Method	: other: not applicable : the compound is a gas at ambient temperature
rear CLP	
Test substance 16.03.2001	
5.1.2 ACUTE INHALATION	ΙΤΟΧΙΟΙΤΥ
Туре	: LCLo
Value	: > 200000 ppm
Species	: other: wistar rat
Strain	:
Sex	: male/female
Number of animals Vehicle	: 10 :
Doses	:
Exposure time Method	: 1 hour(s) :
Year	: 1982
GLP	:
Test substance	: no data
Remark	: 5 male and 5 female rats were exposed to 200000 ppm VF2 for 1 hr.

OECD SIDS	1,1-DIFLUOROETHYLENE
. TOXICITY	ld 75-38-7
	Date 12.03.2002
	Animals were observed for 7 days after exposure
	Animais were observed for 7 days after exposure.
	No mortality was observed; no respiratory, skeletal reflex,
	or gross pathological changes were observed in these rats.
	Poliobility aritaria: reliable with restriction: comparable
	to quideline study with acceptable restrictions (2c)
28.08.2001	(16)
Туре	: LCLo
Value	: > 200000 ppm
Species	: other: CF1 mouse
Strain	: . Mala
Sex Number of onimals	
Number of animals	. 0
Doses	
Exposure time	: 1 hour(s)
Method	
Year	: 1982
GLP	
Test substance	: no data
Remark	: 5 males per concentration were exposed to 4000, 8000,
	20000,40000, 100000 and 200000 ppm for 1 hour
	Animals were observed for 7 days after exposure
	Exposures were under static conditions.
	All mice survived the highest concentration (200000 ppm).
	It produced excessive grooming benaviour. There were no loss
	in body weight, no respiratory changes, abnormal reliexes,
	or gross patriological changes.
	Reliability criteria: reliable with restrictions:
	comparable to guideline study with acceptable restrictions
	(2c)
28.08.2001	(16)
Туре	: LCLo
Value	: ca. 128000 ppm
Species	: Kat
Strain	: Mala/famala
Jex Number of animals	
Vahiela	· Other: air
Doses	· • • • • • • • • • • • • • • • • • • •
Exposure time	
Method	:
Year	: 1949
GLP	: No
Test substance	:
Remark	: Animals were exposed to undescribed concentrations of VF2 in
	a dessicator (flow through) with variable exposure times,
	during which onset of first symptoms was noted.
	I he actual concentration was not monitored but calculated.
	According to the authors, a 4 hrs exposure period to a
	concentration of approx. $126000$ ppm of VF2 killed 2/6, 3/6, or $4/6$ animals
	Ut the authors do not evolude that animals died of inadequate
	oxvaen supply.
	2 2
6	UNEP Publications

DECD SIDS	1,1-DIFLUOROETHYLEN	NE
TOXICITY	ld 75-38-7	
	Date 12.03.2002	
	A necropsy was performed but not reported.	
	Reliability criteria: not reliable: significant	
	methodological deficiencies (3b)	
28.08.2001	(17)	
Туре	: Other: comparable to current acute inhalation toxicity test guidelines	
Value	:	
Species	: Rat	
Strain	:	
Sex	: no data	
Number of animals		
Vehicle	: Other: oxygen	
Doses		
Exposure time	: 19 hour(s)	
Method		
rear CIP	. IYUU : no data	
	. no data	
Test substance	To data	
Remark	Exposure time: from 30 minutes up to 19 h (some animals)	
	The carrier das was oxygen	
	No deaths were noted:	
	no loss of righting reflex, postural reflex or corneal	
	reflex was observed at any tested concentration.	
	slight intoxication at concentration 40 % and	
	above. At 80 % by volume in air, unsteady gait but no loss	
	of postural reflex. No progressive signs of intoxication	
	were noted at prolonged exposures to 19 hrs.	
	At necropsy, no signs of pulmonary irritation were observed.	
	Since animals were not observed for a 14 day period after	
	exposure, no reliable LC50 value can be determined.	
	Reliability criteria: not reliable: insufficient	
	documentation (3a)	
28.08.2001	(18)	
Туре	:	
Value	:	
Species	: dog	
Strain		
Sex	:	
Number of animals	:	
Vehicle	:	
Doses	:	
Exposure time	:	
Method	:	
Year	: 1990	
GLP	:	
Test substance	:	
Remark	: Inhalation, 25 to 50 % for 5 to 15 minutes (primed with	
	כאווכאוווווס, ווס סמוטומט פרופונגמנוטוו.	
	Reliability criteria: reliable with restrictions: well	
	documented, acceptable scientific principles, acceptable for	
	assessment (2e)	
16.05.2001	(19)	
	()	
Туре	:	
	IINFP Publications	 47
		• /

5 TOVICITY	
5. TOXICITY	la 73-38-7
	Date 12.03.2002
\/alue	
Species	: cat
Strain	
Sex	
Number of animals	
Vehicle	:
Doses	:
Exposure time	:
Method	:
Year	: 1990
GLP	:
Test substance	: Inholation OF to FO 0/ for F to 15 minutes (minutes) with
Remark	<ul> <li>Innalation, 25 to 50 % for 5 to 15 minutes (primed with opinenbring): no pordice consistization</li> </ul>
	epinepinne). No cardiac sensilization.
	Reliability criteria: reliable with restrictions: well
	documented, acceptable scientific principles, acceptable for
	assessment (2e)
16.05.2001	(19)
5.1.3 ACUTE DERMAL	TOXICITY
	: other: not applicable : the compound is a gas.
rear	
GLF Test substance	
16.03.2001	
5.1.4 ACUTE TOXICITY	(, OTHER ROUTES
5.1.4 ACUTE TOXICITY	(, OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001	(, OTHER ROUTES
<ul> <li>5.1.4 ACUTE TOXICITY</li> <li>16.03.2001</li> <li>5.2.1 SKIN IRRITATION</li> </ul>	Y, OTHER ROUTES
<ul> <li>5.1.4 ACUTE TOXICITY</li> <li>16.03.2001</li> <li>5.2.1 SKIN IRRITATION</li> </ul>	Y, OTHER ROUTES
<ul> <li>5.1.4 ACUTE TOXICITY</li> <li>16.03.2001</li> <li>5.2.1 SKIN IRRITATION</li> <li>Species</li> </ul>	r, other routes N
<ul> <li>5.1.4 ACUTE TOXICITY 16.03.2001</li> <li>5.2.1 SKIN IRRITATION Species Concentration</li> </ul>	r, other routes
<ul> <li>5.1.4 ACUTE TOXICITY 16.03.2001</li> <li>5.2.1 SKIN IRRITATION</li> <li>Species Concentration Exposure </li> </ul>	Y, OTHER ROUTES
<ul> <li>5.1.4 ACUTE TOXICITY 16.03.2001</li> <li>5.2.1 SKIN IRRITATION</li> <li>Species Concentration Exposure Exposure time</li> </ul>	Y, OTHER ROUTES
<ul> <li>5.1.4 ACUTE TOXICITY 16.03.2001</li> <li>5.2.1 SKIN IRRITATION</li> <li>Species Concentration Exposure Exposure time Number of animals Vabias</li> </ul>	Y, OTHER ROUTES
<ul> <li>5.1.4 ACUTE TOXICITY 16.03.2001</li> <li>5.2.1 SKIN IRRITATION</li> <li>Species Concentration Exposure Exposure time Number of animals Vehicle PDII</li> </ul>	Y, OTHER ROUTES
<ul> <li>5.1.4 ACUTE TOXICITY 16.03.2001</li> <li>5.2.1 SKIN IRRITATION</li> <li>Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result</li> </ul>	Y, OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification	Y, OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method	Y, OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year	A, OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year GLP	A, OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year GLP Test substance	<b>Y</b> , OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year GLP Test substance 10.03.1995	<b>N</b>
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year GLP Test substance 10.03.1995 5.2.2 EYE IRRITATION	<b>Y</b> , <b>OTHER ROUTES</b>
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year GLP Test substance 10.03.1995 5.2.2 EYE IRRITATION	Y, OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year GLP Test substance 10.03.1995 5.2.2 EYE IRRITATION	r, OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year GLP Test substance 10.03.1995 5.2.2 EYE IRRITATION	Y, OTHER ROUTES
5.1.4 ACUTE TOXICITY 16.03.2001 5.2.1 SKIN IRRITATION Species Concentration Exposure Exposure time Number of animals Vehicle PDII Result Classification Method Year GLP Test substance 10.03.1995 5.2.2 EYE IRRITATION Species Concentration Dose	Y         Y <td< td=""></td<>

OECD SIDS	1,1-DIFLUOROETHYLENE
5. TOXICITY	ld 75-38-7
	Date 12.03.2002
Exposure time	:
Comment	
Number of animals	:
Vehicle	:
Result	:
Classification	
Method	: other:
Year	:
GLP	
Test substance	
Remark	: No experimental data are available but data from acute, repeated and prolonged exposure by the inhalation route did not reveal any sign of eve irritation
16.03.2001	initiation route did not reveal any sign of eye initiation.

## 5.3 SENSITIZATION

:
:
:
:
: other: not applicable (gas)
:
:
:

#### 5.4 REPEATED DOSE TOXICITY

Type Species Sex Strain Route of admin. Exposure period Frequency of treatm. Post exposure period Doses Control group NOAEL Method Year GLP Test substance Remark	rat male/female Sprague-Dawley inhalation 13 weeks 6 h/d, 5 d/wk 250, 1000, 7000 ppm = 1000 ppm other: comparable to m odified OECD guideline 422 1985 yes The subchronic toxicity part of a combined subchronic toxicity/reproduction/fertility study is described here. weanling rats (180 per sex)and young adult rats (180 per sex) were whole-body exposed. An interim kill was performed at week 4.
	An interim kill was performed at week 4. A group of animals was retained for a recovery period of 10 weeks. Parameters evaluated:

OECD SIDS	1,1-DIFLUOROETHYLENE
5. TOXICITY	ld 75-38-7
	<b>Date</b> 12.03.2002
	Clinical observations were made 2 X / day. Once per week close clinical evaluation was made on each animal.
	Body weight was measure at the start of the study and weekly thereafter.
	Hematology was evaluated in week 4 and 13
	Urinalysis was performed in the males in week 4 and 13.
	Vaginal smears were performed in control and high dose animals in weeks 3 and 4 and weeks 12 and 13.
	Pathology: Organ weights of all animals killed at week 5 and 13 of adrenals, brain, coagulating glands, epididymis, heart, kidney, liver, lung, ovaries, pituitary, prostate, simal vesicles, spleen, testes, uterus. Testes of animals killed after the 10 week recovery period were also weighted.
	Microscopic evaluation: Sperm morfology at week 14 Histological eximination of testis and epididymis(all animals), lungs and nasal turbinates (all animals), adrenals and pituitary (controls and high dose), ovaries and uterus (control and high dose)
	Statistical methods:
	Body weight: analysis of covariance foolowed by Dunnett's test organ weights, hematological, biochemical data: analysis of variance, followed by Dunnett's test Incidences of pathological changes: Fisher's exact test.
	Results:
	Compared to controls, no treatment related changes were observed in treated animals regarding body weight, heamatology, urinalysis, mortality, oestrus cycle, organ weights.
	Pathology: No treatment related effects were observed on marcoscopic examination.
	Microscopic examination revealed treatment related changes in the nose, in animals of both subsets. The changes occurred in the vomeronasal epithelium which cobvers partly the wall of the vomeronasal organ. The ciliated epithelium, which covers the other part of the wall, looked normal.
	Weanlings: At day 28 a varying number of large vacuoles was observed in the vomeronasal epithelium and, in the more severe cases, the vomeronasal epithelium showed a complete vacuolar change. The abnormality was noticed in 4 males and in all females exposed to 7000 ppm VF2. At day 91, similar degenerative lesions were observed, but they were distinctly less pronounced and were restricted to 3 females of the 7000 ppm group only

TOXICITY	ld 75-38-7
	Date 12.03.2002
	Young adults:
	At day 28, a varving number of large vacuoles was observed in the
	vomeronasal enithelium and in more severe cases the vomeronasal
	enithelium showed a complete vacuolar change. The abnormality was
	noticed in 6 males and 9 females exposed to 7000 ppm V/E2. At day 91
	such degenrative changes were not found: the vomeronasal organ looked
	normal in all male and all female rate
	In animals of both subsets signs of rhinitis, characterised by varying numbers of inflammatory cells and a skingt hyperplastic reaction of the local
	enithelium were observed in a number of male and female rats of all droups
	(including the controls). There was no concentration-response relationship.
	NOEC rat 13 weeks: 1000 ppm.
	Note:
	The findings in the vomeronasal organ of rats is considered
	to be of doubtfull biological significance.
	Reliability criteria: reliable without restrictions:
	comparable to guideline study (1b)
	Remark: Key study for SIDS evaluation
05.03.2002	(20)
Туре	:
Species	
Sex	: male
Strain	: other: Albino ChR-CD
Route of admin.	: inhalation
Exposure period	: 2 weeks
Frequency of treatm.	: 6 h/d, 5 d/wk
Post exposure period	:
Doses	: 25000 ppm
Control group	:
Method	:
Year	: 1977
GLP	:
Test substance	:
Remark	: Slight increase in total fluoride in urine after the last
	exposure; slightly higher erythrocyte count; tracheitis,
	mucosal hyperplasia and an increase in murine pneumonitis.
	Reliability criteria: not assignable: documentation
	insufficient for assessment (4e)
19.06.2001	(21)
Туре	:
Species	: rat
Sex	: male/female
Strain	: Fischer 344
Route of admin.	: inhalation
Exposure period	: 90 days
Exposure period	
Frequency of treatm.	. 01/d, 5 d/wk
Frequency of treatm. Post exposure period	: 01/d, 5 d/wk
Frequency of treatm. Post exposure period Doses	: 500, 1500, 5000, 15000, 50000 ppm
Frequency of treatm. Post exposure period Doses Control group	: 500, 1500, 5000, 15000, 50000 ppm :

DECD SIDS	1,1-DIFLUOROETHYLENE
. TOXICITY	ld 75-38-7
	Date 12.03.2002
Year GLP Test substance Remark	<ul> <li>1984</li> <li>no data</li> <li>10 animals per sex and per concentration were whole-body exposed.</li> <li>Results:</li> <li>Body weight gain.</li> <li>Statistically significant bodeweight changes were seen during the first half of the exposure period when compared to controls. There was none at study termination.</li> <li>There was an unexplained BW depression in the controls with an actual body weight loss in the controls of both sexes at week 5 of the study</li> <li>Clinical observations</li> <li>none that were considered related to treatment</li> </ul>
	Haematology
	Statistically significant decrease in RBC counts, hemoglobin levels and hematocrit levels for males of the 1500 and 50000 ppm dose groups. No changes in females.
	Males 0 ppm 1500 ppm 50000 ppm RBC 9310 8470 8483 HGB 16.32 15.12 15.17 HCT 47.34 42.94 43.50
	Clinical chemistry
	Significant increase of creatinine levels in the 500 (+ 24%) and 1500 ppm (+18%) femdes. Significant increase in SGOT levels in 50000 ppm males (+20%) and females (+56%). Significant increase in SDH levels in 15000 ppm males (+22%) and females (+28%) Significant increase in BUN levels in 1500ppm and 5000 ppm males (+ 16 and +21% resp) and in 5000 and 50000 ppm females (+21 and +15% rep.). Significant decrease in SGPT levels in 15000 ppm males (- 13%) and females (- 18%).
	Organ weights
	Absolute organ weights

Significant increase in liver weights in the 500, 5000 and 15000 ppm males (+13,+17, +14% resp) and in the 500 and 15000 ppm females.(+11, +16% resp) Significant increase in thymus weights in the 1500, 5000, and 50000 ppm females.(+22,+24, +22%) Significant increase in (right) kidney weights in all treated males (+16, +12, +17, +14, +12% resp) and in 15000 ppm females.(+13%) Significant decrease in the (right) testes weights in the 50000 ppm males. (-7%)

# **UNEP Publications**

TOXICITY	ld 75-38-7
	Date 12.03.2002
	Relative organ weights
	Significant increase in rel. liver weights in the 5000 and 15000 ppm females (+5,
	+7% resp). Significant decrease in rel. brain weight in 15000 ppm
	males (-7%). Significant decrease in rel. heart weight in 1500 and 5000
	ppm males (-9, -10% resp). Significant decrease in rel. testes weight in 5000 and 50000 ppm males (-8, -11%).
	Significant increase in rel. kidney weight in 500 and 5000 ppm males (+9, +7% resp) .
	Histopathology
	One rat in the 50000 ppm exhbited serous rhinitis, of which could not be excluded to be related to treatment. Other pathological findings were incidental and considered
	not to be related to treatment. Reliability criteria: reliable without restriction: Comparable to guideline study (1b)
21.11.2001	(15) (22)
Туре	:
Species	: mouse
Sex	: male/female
Strain	: B6C3F1
Route of admin.	: inhalation
Exposure period	: 13 weeks
Frequency of treatm.	: 6 h/d, 5 d/wk
Post exposure period	: 500 1500 5000 15000 50000 ppm
Control group	:
Method	:
Year	: 1984
GLP Tart substance	: no data
Remark	<ul> <li>10 animals per sex and per concentration (including control) were whole-body exposed to VF2.</li> </ul>
	Results
	Clnical observations
	Few mice at various concentrations showed body w eight loss and/or thinness.
	Mortality
	6 animals died during the study (two 5000 ppm males, one 500 ppm female, three 5000 ppm females) There were no treatment related deaths in the 15000 and 50000 ppm
	groups.
	Hematology
	Significant decrease in WBC counts in the 500 ppm females.

5 TOXICITY	ld 75-38-7
5. IUAICITT	Date 12.03.2002
	<b>54.0</b> (2.00.2002
	Significant decrease in lymphocytes in the 500, 5000 and
	50000 ppm males. (not considered biologically significant)
	Significant increase of the hematocrit in the 50000 ppm
	females.
	The latter may by caused by increased erythropoiesis that
	may be due to mild hypoxia.
	Body weight
	Statistically significant body weight depression was
	observed in 500 nnm females at week 9 and 10, in 1500 nnm
	females at weeks 9, 11 and 13, in 5000 nnm females at week 5
	and in 15000 nnm females at week 11
	During the second half of the study, the males had no
	statistically significant changes in their body weights
	It should be noted that several animals with marked RM/ loss
	during week 5 and 6 rapidly recovered in the next week
	This has partially been attributed to a malfunction in the
	watering system. Also 3 deaths are suspected to be caused by
	the same failure
	At study termination only the 1500 ppm females showed a
	statistically significant decrease in BW (-13%). Males of the 500, 15000 and
	50000 ppm and females of all VE2 dose groups
	showed a decreasing trend in BW.
	Organ weights
	Absolute organ weight
	Conferent increase in liver weight of the 1500 party males
	Significant increase in liver weight of the 1500 ppm males
	(+13%). Significant decrease in liver weights of the 15000 ppm
	females (-10%)
	Relative organ weight
	Statistically significant increase in rel. liver weight in
	500 and 1500 ppm males.(+10, +10% resp)
	Pathology
	Gross necropsy
	No findings that could be related to treatment.
	Histopathology
	A small number of of evenesed miss had renal changes
	A small number of of exposed mice had renal changes,
	kiuney/lubule regeneration. This was seen in a Very low
	incluence in all uosed males, except the 5000 ppm males and
	was also noted in the 5000 and the 50000 ppm temales.
	All resions were or minimial seventy. There was no evidence
	טו וכוומו עבשבווכומוטוו טו וופטוטאא.
	The NOFC is considered by the authors to be $< 500$ ppm
	Reliability criteria: reliable without restrictions: Comparable to guideline
	study (1h)
21.11.2001	(23)
	()
Туре	:
Spacios	• rat

5 TOXICITY	ld 75-38-7
5. 10/4/01/1	Date 12.03.2002
Sex	: male/female
Strain	: Sprague-Dawley
Route of admin.	: inhalation
Exposure period	: 13 weeks
Frequency of treatm.	: 6 h/d, 5 d/wk
Post exposure period	:
Doses	: 1000, 7000, 40000 ppm
Control group	: yes
Method	: other: comparable to OECD guideline 413
Year	: 1985
GLP	: yes
Test substance	: no data
Remark	: 30 animals per sex and per concentration were used in this study.
	Following parameters were evaluated:
	Condition and behaviour, ophatalmology, body weight, food
	consumption, heamatology, clinical chemistry, urinalysis,
	organ weights, gross and microscopic pathology.
	Interim kills were performed at 2 and 4 weeks.
	Decreased BW gain from 7000 ppm in males and females but not
	statistically significant at the end of the study.
	A transient (first weeks only) decreased food consumption in
	males and femals at 40000 ppm.
	Heamatology: $(7\%)$ statistically significant apomia in 40000 ppm males ( $7\%$ ) and females (
	statistically significant allernia in 40000 ppm males (-1 /0) and remaies (-
	1170). Statistically significant increase of mean corpuscillar
	Statistically significant increase of mean corpuscular volumes transform the study in the $40000$ ppm males ( $\pm 5\%$ ) and at study
	termination in the $40,000$ npm females (+6%)
	Statistically significant increase of mean corresponder
	baomoglobin in 40,000 nnm malos and fomalos at study
	termination ( $\pm 8$ and $\pm 8\%$ resp)
	Statistically significant degraps in WPC sounts, at study
	termination from 7000 ppm in males (34, 25% resp) and at 40000 ppm in
	females (-28%) (mainly decrease of lymphocytes)
	Clinical chemistry:
	In the first few weeks of the study there was a
	statistically significant decrease of serum alkaline
	phosphatase in the mid and high dose males and the high dose
	temales. This effect was transient.
	Alpha2 globulins were increased in mid and high dose males
	at study termination (+7,+10% resp).
	Several transient and not dose related changes were noted in
	some other serum enzymes (GPT, GOT, OCT) , in Ca, Na, Cl ,
	K, triglycerides and cholesterol were observed but were not
	considered to be related to the test compound.
	Urinalysis: gamma glutamyl tranferase activity was increased during the
	study and decreased at study termination in high dose males.
	A similar but transient effect was seen during the study in
	mid and high dose females.
	Organ weight changes:
	Absolute and relative beart weights were increased in all
	7 10001010 0110 110011 0 01 0 01 1 WEIGHTS WEIG HIN E0350 III all

TOXICITY	ld 75-38-7
	<b>Date</b> 12.03.2002
	females:abs:+19;rel:+13%), and in females of the mid dose group in the first
	4 weeks. Also the relative heart weight of the control group on the 4th
	group was higher as compared with that of the control group on the 4th
	The absolute spleen weight was decreased in males and
	females of the high dose group throughout the study (. When
	the spleen weight was expressed relative to body weight, the
	decrease was only observed in males of the high dose group
	on week 2 and in females of ther high dose group on week 2
	and week 13 and in females of the mid dose group on week 2.
	Relative lung weights were increased in males and females of
	the high dose group during the study but not at termination.
	Absolute and relative testes weights were decreased in the
	mid and high dose groups on week 4 and at study termination (end study:-
	23,65% resp) Relative testes weights were decreased in high dass group at study
	relative testes weights were decreased in high dose group at study termination (-64%)
	In addition, the absolute testes weight of the high dose males was
	decreased on week 2.
	Absolute epididymis weight was decreased in the high dose
	group on week 4 and in the mid and high dose groups at study
	termination (end study: -20,-42% resp).
	Relative weight of the epididymes was decreased in the high dose group at
	study termination (-42%).
	Pathology:
	Autopsy revealed soft testes and epididymes in the high dose
	group on week 4 and at study termination.
	Microscopic examination revealed treatment related changes
	in testes and epididymes of males of mid and high dose
	groups throughout the study.
	The changes were characterised by impairment of
	spermatogenesis.
	The spleen showed treament related changes, characterised by
	lymphocytic depletion of the marginal zone, in males and
	temales of the mid and high dose groups and in females of
	ule low dose group at week 2. Vacualar degeneration of the Vomeronasal organ in the nece
	was observed in all treatment aroups
	Changes appeared to decrease in severity with time.
	Mineralisation of the kidneys was increased in males of the
	high dose group after week4 and at study termination.
	NOEC 90 day , rat: < 1000 ppm
	Note: the summarised conclusions mentioned in the report do not coincide
	with the tabulated results.
	Reliability criteria: reliable with restrictions: Comparable to guideline study
00.44.0004	with acceptable restrictions. (inconsistent reporting)(2b)
22.11.2001	(24)
Туре	:
Type Species	: mouse
Type Species Sex	: : mouse : male/female
Type Species Sex Strain Bouto of odmin	: mouse male/female CD-1

TOXICITY	ld 75-38-7
юлент	Date 12.03.2002
Frequency of treatm.	: 6 h/d. 5 d/wk
Post exposure period	
Doses	: 1000, 5000, 15000, 40000 ppm
Control group	
NOAEL	: = 40000 ppm
Method	: other: comparable to OECD guideline 412
Year	: 1988
GLP	: yes
Test substance	: as prescribed by 1.1 - 1.4
Remark	: 5 animals /sex/group were whole-body exposed to 0, 1000,
	5000, 15000 and 40000 ppm.
	The gas concentration in the chambers were monitored by GC
	four times per exposure day.
	There was no mortality.
	There were no clinical symptoms observed which were related
	to the treatment.
	There was no treatment related effect on body weight or BW
	gain.
	Absolute and relative organ weights were not affected by
	treatment.
	Macroscopic examination at necropsy revealed no treatment
	related changes.
	The NOEC in this study was considered to be 40000 ppm
	Reliability criteria: reliable without restrictions: Comparable to guideline
	study (1b)
22 08 2001	(25)
22.00.2001	(20)
Туре	:
Species	: mouse
Sex	: male/female
Strain	: CD-1
Route of admin.	: inhalation
Exposure period	: 13 weeks
Frequency of treatm.	: 6 h/d, 5 d/wk
Post exposure period	:
Doses	: 0, 1000, 7000, 40000 ppm
Control group	:
NUAEL	: = ppm
wethod Voor	: otner: comparable to UECD guideline 413
reaf CLP	. 1300
ULF Teat aubatanaa	. yes
rest substance Remark	. as prescribed by 1.1-1.4 10 animals per sex and per group wore whole body expected
Nellidik	. To animals per sex and per group were whole-body exposed.
	exposure day.
	Clinical observations:
	Treatment related, but not dose related increased locomotor
	activity. It was observed in both sexes and was prominent in
	mid study and declined afterwards.
	Increased sensitivity to touch (aggressivity) in 40000 ppm
	males during the last week of exposure, in 7000 and 40000
	pprintermates during the last 2 weeks of exposure.
	Dose related rough coat in males after eight weeks of
	exposure.

No treatment related effects

Hematology:

Significant increase in mean corpuscular hemoglobin content in the 40000 ppm males.

Absolute and relative organ weights:

No treatment related effects

Pathology:

No treatment related macro- and microscopic pathological changes observed.

The NOEC is 7000 ppm Reliability criteria: reliable without restrictions: Comparable to guideline study (1b) (26)

22.08.2001

#### 5.5 GENETIC TOXICITY 'IN VITRO'

Type System of testing Test concentration Cycotoxic concentr.	:	Escherichia coli reverse mutation assay exposure to maximally soluble concentration in aqueous medium. Unknown
Metabolic activation	÷	
Result		
Year	:	1968
GLP	:	No
Test substance	:	no data
Remark	:	E; Coli strain B and strain Sd-4 were cultured exposed to gaseous VF2 in their liquid medium by bubbling the gas trough the medium for 10 minutes (It is assumed that microorganisms were exposed to the max disolved VF2 concentration in the medium which is unknown). E Coli strains were plated 24 hrs after exposure on agar. Viability decreased to 62% of control for strain B and to 94 % for Sd-4.
		Mutation rate was 1000 higher than control for strain B and 100 times higher for strain Sd-4
		This test procedure is not standard and results are therfore difficult to interpret.
		Since this study is not considered as a key study, further details of the study procedure are not described.
		Reliability criteria: Not reliable: Unsuitable test system
26.04.2001		(27)
Type System of testing	:	Ames test exposure to the gas
58		UNEP Publications

OECD SIDS 1,1-DIFLUOROETHYLENE ld 75-38-7 5. TOXICITY Date 12.03.2002 Test concentration Unknown Cvcotoxic concentr. Metabolic activation with and without Result Method Year 1977 GLP no data Test substance no data Remark Histidine dependent Salmonella typhimurium strains TA 1535, TA 1537, TA 1538, TA 98 and TA 100 were used Tests were carried out in the presence and absence of rat liver derived S -9 mix activation system. Positive controls were tested with different mutagens. The cultures were exposed to unknown gaseous concentrations for1, 24, 48 and 72 hrs. The revertant colonies were counted in all concentration groups and compared to the number of revertant colonies in the control group. Statistical methods: not mentioned Results: Positive in TA 1535 only in the in absence (after 72 hrs)and in presence of S-9 activation (from 24 hrs exposure on) Negative in TA 1537, 1538 TA 98, TA 100 Reliability criteria: Not reliable: documentation insufficient for assessment (3a) 26.04.2001 (28)Type Other: cell transformation System of testing Test concentration unknown Cycotoxic concentr. Metabolic activation without ÷ Result negative Method Year 1978 GLP no data Test substance no data Remark BALB/3T3 cells were cultured in bottles with our without culture medium. The test with culture medium was run in duplo. The test without culture medium was run once. Bottles were gassed (concentrations not described) and cells were exposure for 0, 0.5, 1, 2, 4, 6, 24, and 48 hrs. A positive control was run with3-methylcholanthrene. The test without the culture medium did not show any transformation of cells. There was a slight increase in incidence of transformed cells in one of two replicates of the test with culture

**UNEP Publications** 

medium.

DECD SIDS	1,1-DIFLUOROETHYLENE
. TOXICITY	ld 75-38-7 Date 12.03.2002
	Date 12.03.2002
	In this test VF2 was considered not mutagenic to BALB/3T3 cells under the conditions of the test.
	The concentrations used are not mentioned in the test
	Therefore the observed effects cannot be interpreted.
	A lack of oxygen may have influenced the outcome of the test.
	Reliability criteria: Not reliable: documentation
26.04.2001	insufficient for evaluation (3a) (29)
	()
Туре	: Ames test
System of testing	: exposure to gas
Cycotoxic concentration	. up to 50 %
Metabolic activation	: with and without
Result	:
Method	
Year	: 1979 : no data
Test substance	: no data
Remark	: Histidine dependent Salmonella typhimurium strains TA 1535
	and TA 100 were used to detect base-pair substitution
	frame-shift mutations
	Tests were carried out in the presence and absence of rat
	liver derived S - 9 mix activation system.
	The cultures were exposed to gaseous concentrations from 0
	to 50% VF2 for 48 hrs.
	the end of the exposure by GC.
	The revertant colonies were counted in all concentration
	groups and compared to the number of revertant colonies in
	the control group.
	Statistical methods:
	number of revertants for each dose group: z-test
	Dose dependency: Spearman rank test.
	Results:
	Positive in TA 1535 only in the presence of S-9 activation,
	trom 10 %,
	without S9
	malout co.
	Reliability criteria: reliable without restriction:
26.04.2004	comparable to guideline study (1b)
20.04.2001	(3U)
Туре	: Amestest
System of testing	: exposure to the gas
Test concentration	: up to 50%
Cycotoxic concentr.	: with and without

1,1-DIFLUOROETHYLENE ld 75-38-7

> Data 12 03 2002

	<b>Date</b> 12.03.200	)2
Result	: negative	
Method	:	
Year	: 1979	
GLP	: no data	
Test substance	: no data	
Remark	: Histidine dependent Salmonella typhimurium strain TA 100 was	
	exposed up to 24 hrs to gaseous concentrations of 0, 20, and	
	50 % VF2.	
	Exposure was made with and without rat liver derived S9	
	activation.	
	Exposure was measured by GC.	
	There was a marginal but not significant increase in	
	revertant colonies at 24hrs exposure at 50% and only in the	
	presence of S9.	
	Reliability criteria: Not reliable: significant	
~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	methodological deficiencies (3b)	
26.04.2001	(31)	
Turno		
iype System of testing	· exposure to the gas phase	
Jystem of testing	. exposure to the gas phase	
Cycotoxic concentr		
Motobolio activation		
Docult	: menative	
Method	. liegauve	
Voar	. 1086	
GIP		
Test substance	as prescribed by 11-14	
Romark	: Cell cultures of the BH4 clone of Chinese hamster overv	
(cinan)	cells were exposed to daseous concentrations of VE2	
	Concentrations tested were 0(nitrogen) 20 40 60 80 and	
	100% VF2 Actual test concentration werez measured at the	
	beginning and at the end of the exposure by GC	
	beginning and at the end of the exposure by CO.	
	Each concentration was tested on cell cultures in the	
	presence or absence of rat liver derived S9 activation	
	system	
	Exposure period:	
	18 - 19 hrs for non-activated cultures	
	5 hrs for activated cultures	
	Evaluated parameters:	
	Cytotoxicity: after exposure by survival of subcultured	
	cells for seven days.	
	Mutagenicity: incidence transformation of the HGPRT gene	
	locus by measuring incidence of cells with resistance to	
	6-TG.	
	Statistical analysis:	
	for effect of or each concentration in comparison with	
	control: mutant frequency: E test of significance	
	ior concentration-responce relationship: ANOVA	
	ior linear, quadratic or higher order effects: F-test of significance.	
	Posulte:	
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OECD SIDS 5. TOXICITY

5. TOXICITY	ld 75-38-7
5. 10/10/11	Date 12.03.2002
	The mutant frequency of none of the sample concentrations
	tested was significantly greater than that of the negative
	controls (p<0.05)
	and
	The correlation between mutant frequency and the
	concentration of the test sample was not significantly
	greater than 0 (p< 0.05)
	Thus under the conditions of the test VF2 was considered not
	a mutagen in this test.
	This is considered as a key study for SIDS evaluation.
	Reliability criteria: reliable without restriction:
	comparable to guideline study (1b)
26.04.2001	(32)
Туре	: Cytogenetic assay
System of testing	: exposure to gaseous VF2
Test concentration	: 0, 25, 50, 75 and 100%
Cycotoxic concentr.	
Metabolic activation	: with and without
Result	:
Method	:
Year	: 1986
GLP	: yes
Test substance	: as prescribed by 1.1 - 1.4
Remark	: This test was performed according to accepted EPA and OECD
	guidelines.
	Cell cultures of the BH4 clone of the CHO cell line were
	exposed to different concentration of VF2 in the gas phase.
	The exposure was done in presence and absence of rat liver
	derived S-9 activation system.
	Exposure periods were 5 hrs (without activation) and 2 hrs
	(with activation).
	negative control consis of nitrogen exposure only.
	Positive control was done with exposure to ethylmethane
	sulphonate.
	Evaluated parameters:
	Cytotoxicity by measurement of delay of cell replication
	after end of exposure.
	Genotoxic effect by measurement of incidence of chromosomal aberrations.
	Statistical evaluation:
	Aberrations per cell (dose vs control): Mann-Withney U test
	Dose responce relation: Jonckheere test
	Percent abnormal cells and cells with more than one
	aberration: dose vs control: Fisher exact test
	dose response relation: Cochran-Armitage for trend.
	Results:
	No cell replication delay in any of the tested
	concentrations.
	No treatment related chromosomal aberrations in the trials
	without S9 activation.

5. TOXICITY	ld 75-38-7	
	<b>Date</b> 12.03.2002	
	No treatment related chromosomal aberrations in the trials	
	with S9 activation.	
	VF2 was considered not to cause chromosomal aberrations	
	under the conditions of the test.	
	Reliability criteria: reliable without restriction: GLP	
26.04.2001	(33)	
5.6 GENETIC TOXICI		
<b>T</b>		
Type Species	: Micronucieus assay	
Sex	: male/female	
Strain	: CD-1	
Route of admin.	: inhalation	
Exposure period	: 6h	
Doses	: 5000, 15000, 40000 ppm	
Method		
Year	1988	
GLP	: yes	
Test substance	: as prescribed by 1.1 - 1.4	
Remark	5 male and 5 lemale CDT mice were exposed (nose-only) to 0, 5000, 15000 and 40000 ppm VE2 for 6 brs	
	A positive control treated with chlorambucyl was	
	concurrently evaluated.	
	Test concentrations were monitored by GC every 1/2 hr during	
	the exposure.	
	Bone marrow smears of treated and control animals were made	
	at 24, 48 and 72 hrs after initiation of the exposure.	
	Parameters evaluated:	
	ervthrocytes vs mature cells.	
	Mutagenicity: incidence of micronuclei in polychromatic	
	erythrocyte.	
	Statistical evaluation:	
	Incidence of micronuclei in enthrocytes of treated animals	
	vs control: Mann-Withney U test	
	Results:	
	No ovidence of toxicity of VE2 for the murine have merry of	
	animals exposed to up to 40000 ppm V/F2	
	No difference in frequency of micronuclei in polychromatic	
	erythrocytes between treated and control animals (both	
	sexes).	
	It was considered that we deaths are different fit - to -t)/CO	
	n was considered that under the conditions of the test VF2 did not cause chromosomal or other damage leading to	
	micronucleus formation in polychromatic erythrocytes of	
	exposed mice.	
		_
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DECD SIDS	1,1-DIFLUOROETHYLENE
5. TOXICITY	ld 75-38-7 Date 12.03.2002
	Reliability criteria: Reliable without restriction:
40.05.0004	comparable to guideline study (1b)
18.05.2001	(34)
Туре	: Drosophila SLRL test
Species	: Drosophila melanogaster
Sex	:
Strain Route of admin	: inhalation
Exposure period	: 24 h
Doses	: 4.95, 22.8 and 43 %
Result	: negative
Method Year	: • 1088
GLP	: Ves
Test substance	: as prescribed by 1.1 - 1.4
Remark	: VF2 was evaluated for mutagenic activity using the Oregon-R
	strain of Drosophyla melanogaster.
	affected the fertility of the treated males compared to the
	negative controlat concentrations up to 50%
	the main test was conducted using 3 mixtures of 4.95, 22.8
	and 43% of VF2 with 20% oxygen and the balance made by N2.
	Negative control was 20% O2 and 80% N2
	The positive control was ethyl methane sulphonate in the
	feed.
	Actual test atmosphere was monitored by GC.
	24 hrs after end exposure treated males were mated to virgin Base females
	Statistical evaluation:
	Lethality per dose vs control: ? Jethality-concentration relation: Cochran Artmitage trend
	analysis.
	Results:
	The differences in lethality in progeny of males between control and VF2 treated groups were not statistically
	different.
	VF2 was considered not to be mutagenic to the X chromosome
	of Drosophyla melanogaster under the conditions of the test.
	Reliability criteria: Reliable without restriction:
	comparable to guideline study (1b)
26.04.2001	(35) (36)
5.7 CARCINOGENICI	ſY
Species	: rat
Sex	: male/female
Strain	: Sprague-Dawley
Route of admin.	: Inhalation
Exposure period Frequency of treatm	: 6 h/d. 5 d/wk

5 TOYICITV	I,I-DII LOOKOLIIII LEINI Id. 75-38-7
5. TOXICITY	Date 12.03.2002
	Date 12.03.2002
Doses	: 150,600,2500,10000 ppm
Result	:
Control group	· · Ves
Mathad	- other: OECD quideline 452
Veer	
rear	. 1991
GLP	: yes
Test substance	: as prescribed by 1.1 - 1.4
Remark	: Control group consisted of 140 animals of each sex.
	In the exposed groups, each group consisted of 80 animals of
	each sex.(including 20 animals/sex /group as satellites)
	All animals were whole-body exposed.
	Exposure concentrations were monitored by GC analysis
	Interim kill of 20 animals per sex and per group was
	notermed at 12 mention
	penomed at 12 months.
	Parameters observed:
	Clinical observation was made twice per day.
	Ophtalmoscopy was perform prior to study on all animals
	killed at 12 months (satellite), on 10 animals/sex of
	control and high dose group at six months
	control and high dose group at six months.
	Serological examination was made on receipt of animals, at
	start of exposure at week 22 at week 26 one year after
	start at 18 months and at 2 years
	start, at to montho and at 2 yours .
	Body weight was recorded once every week until week 13 and
	every four week thereafter.
	Food consumption was determined during the first 13 weeks in
	all groups (measured per cage)
	Hematology was performed on 20 animals/sex/group on weeks
	13 26 52
	Clinical chemistry was performed on blood drawn from 20
	animals/sex/group on weeks 13 and 26 and on weeks 53 and 54
	on the animals killed at 12 months.
	Urinalysis was performed on satellite animals on weeks 13,
	26, and 52
	Pathology:
	Macroscopic examination was performed on all animals.
	Organ weights were measured on all satellite animals killed
	at 12 months and on 10 animals/sex/group at study
	termination.
	Weighted organs included: adrenals, brain, epididvmis,
	heart, kidneys, liver, lungs with mediastinal lymphnodes.
	trachea and larvnx ovaries nituitary spleen testes
	thumid
	u iyi viu. Histological examination was performed on following ticcups
	of controls and high does animale.
	or controls and high dose animals:
	adrenais, aorta, axiliary lymphnodes, brain, ceacum,
	coagulating glands, colon, epididymis, eye, femur with joint
	and the second second second distribution of the second distribution is a second second second second second se
	and bone marrow, neart, kidney, larynx, liver, lung, mammary

TOXICITY	ld 75-38-7
5. TOXICIT I	Date 12.03.2002
	oesofagus, ovaries, pancreas, parathyroid, parotid salvary
	gland, pharynx , pituitary.
	In the 600 ppm animals following tissues were examined:
	adrenals, brain, kidney, liver, larynx, lung, nose, ovary,
	partathyroids, pituitary.
	Statistical methods used:
	Body weights: analysis of covariance, Dunnetts multiple
	comparison test,
	Body weight and BW gain: Bartlet's test, Dunnet's test.
	Food intake: Bartlett's test, LSD
	Absolute and relative organ weights, hematological data,
	urine data: Bartiett's test, Dunnett's test
	WBC counts, Reticulocyt counts, blood globulins, urine pH:
	Kruskal-Wallis method
	Incidences in histopathological changes, mortality: Fishers
	test
	All pairwise comparisons were two-tailed.
	Results:
	No treatment-related mortality was observed.
	Clinical observation:
	No treatment-related changes were noted.
	Ophtalmology:
	No treatment-related effects were observed.
	Body weight gain and food consumption:
	No treament-related effects on BW gain was observed, but
	food consumption tended to be lower in treated animals than
	in controls
	Hematology:
	No consistent and clear changes were observed which could be
	related to treatment
	Urinalysis:
	No treatment-related changes were observed.
	Organ weights:
	Absolute organ weights after 12 months were similar in all
	groups. Relative organ weights (brain, beart, anidudimia) of the
	neialive organ weignis (Drain, nearl, epidydiinis) of the
	recompared to compared to control at study termination.
	Pathology:
	Macroscopic and microscopic examination at necropsy did not
	reveal treatment related effects.
	Some differences in incidence of miner need changes were

OECD SIDS	1,1-DIFLUOROETHYLENE
5. TOXICITY	ld 75-38-7
	Date 12.03.2002
	considered as of negligible toxicological concern.
	Evaluation of the observed neoplasms did not reveal treatment related shifts in benign or malignant tumor incidence, total number of tumors, or total number of tumor bearing animals.
	Key study for SIDS evaluation.
	Reliability criteria: reliable without restriction: GLP Guideline study (1a)
22.08.2001	(37)
Species	: mouse
Sex	: male/female
Strain	: CD-1
Route of admin.	: inhalation
Exposure period	: 18 months
Frequency of treatm.	: 6 h/d, 5 d/wk
Post exposure period	:
Doses	: 0, 600, 2500, 10000 ppm
Result	
Control group	:
Method	: other: OECD guideline 451
Year	: 1991
GLP	: ves
Test substance	: as prescribed by 1.1 - 1.4
Remark	: 82 animals per sex and per group were whole- body exposed.
Homan	Actual exposure concentrations were measured by GC. 6 times
	per exposure day.
	Parameters observed:
	Gross clinical observations were made twice daily. Detailed physical eximinations were performed pretest and weekly thereafter.
	Ophtalmology was performed pretest and on moths 6, 12 and at study termination
	Body weight was measured twice pretest, weekly til week 13, monthly til week 52, then weekly til end study.
	Food consumption was measured weekly til week 13 the, monthly til week 52, then weekly thereafter.
	Hematology and dinical chemistry were evaluated at 12 and 18 months
	Pathology: Macroscopic examination on all animals. Histology on respiratory tract, and tissue masses in all animals of all dose groups. In controls and high doses: abdominal aorta, adrenals, bone and bone marrow, brain, oesofagus, eyes with optic nerve, gall bladder, heart, cecum, colon, rectum ileum, jejunum, duodenum, kidney, larynx, liver, lungs, lymph nodes, nasopharynx, sciatic nerve, ovaries, pancreas pharynx, pituitary, prostate, salivary gland, s eminal vesicles, skin, spinal cord, skin, mammary gland, spleen, stomach, testes,

	Date 12.03.2002
	epididymis, thymus, thyroid, parathyroid, trachea, urinary bladder, uterus.
	Statistical methods:
	Chi-square: proportion of incidences between groups, fishers exact test: comparison of treatment group with control bonferroni inequality: overall test of stated significance level armitage test: for linearity trend in dose groups
	Results:
	Survival was similar amongst groups at study termination
	Clinical observations:
	No treatment related effects observed.
	Ophtalmology:
	No treament related effects observed.
	Body weight gain and foor consumption:
	No treatment related effects observed.
	Hematology:
	No treatment related effects observed.
	Pathology:
	No treatment related m acroscopic or microscopic lesions observed.
	When observed benign and malignant neoplasms in VF2 exposed animals were compared to respective controls, none were considered to be related to exposure to the test substance.
	Overall the morphological findings in the animals from this study were not considered to be of any toxicological or oncological significance with respect to VF2.
	VF2 is not considered to be a carcinogen for the mouse.
	Key study for SIDS evaluation
	Reliability criteria: reliable without restriction:
21.11.2001	(38) (38)
Species	: rat
Sex	: male/female
Strain	: Sprague-Dawley
Route of admin.	: gavage
Exposure period	: 52 weeks
Frequency of treatm.	: 4 to 5 d/wk

VEUD SIDS	
5. TOXICITY	IC 75-38-7 Date 12.03.2002
Doses	: 4.12, 8.25 mg/kg (dissolved in olive oil)
Result	:
Control group	: yes, concurrent vehicle
Method	: other: is not comparable to accepted study guidelines
Year	: 1979
GLP	: no
Test substance	: no data
Remark	: In the high dose group 35 animals per sex were used.
	In the low dose and in control 30 animaLs of each sex were
	used per group.
	Parameters ovaluated:
	Falameters evaluated.
	Body weight: eve ry 2 weeks during treament, every eight
	weeks after treatment.
	Clinical evaluation: every 2 weeks
	Pathology: macroscopic and histological examination on all
	animals. Following tissues were evaluated: Zymbal glands,
	intrascapular brown fat, salivary glands, tongue, lungs,
	liver, kidney, adrenals, spleen stomach, intestines,
	bladder, brain, every organ with pathological lesions.
	No statistical evaluation was performed.
	Results:
	Lipomas and liposarcomas were observed:
	number of lipomas were increased in the high dose group
	compared to the number found in controls but no lipomas were
	found in the low dose group.
	There was a dose related trend in the numbers of
	liposarcomas.(0% in controls versus 1.7% in low dose versus
	4.3% in high dose.
	ů – Elektrik
	No other pathological effects were mentioned
	Reliability criteria: Not reliable: documentation
	insufficient for assessment (3a)
22.08.2001	. (39)
22.00.2001	
5.8.1 TOXICITY TO FERT	nLITY
Туре	: Fertility
Species	: rat
Sex	: male/female
UUX	: Sprague-Dawley
Strain	
Strain Route of admin.	: inhalation
Strain Route of admin. Exposure period	: inhalation : 13 weeks
Strain Route of admin. Exposure period Frequency of treatm.	: inhalation : 13 weeks : 6 hrs/d 5 d/wk
Strain Route of admin. Exposure period Frequency of treatm. Premating exposure per	: inhalation : 13 weeks : 6 hrs/d 5 d/wk iod

: 250, 1000, 7000 ppm UNEP Publications

: 13 weeks

:

Duration of test

No. of generation studies Doses

5 TOXICITY	H 75.38.7
5. TOXICITY	Data 12.03.2002
Control aroup	: ves
NOAEL parental	: 7000 ppm
Method	: other: comparable to modified OECD guideline 422
Year	: 1985
GLP	: Ves
Test substance	: as prescribed by 1.1 - 1.4
Remark	: The male fertility part of a combined subchronic toxicity/reproduction/fertility
	study is described here.
	weanling rats (180 per sex and per concentration)and young adult rats (180
	per sex and per concentration) were whole-body exposed .
	An interim kill was performed at week 4.
	A group of animals was retained for a recovery period of 10
	weeks.
	Parameters evaluated:
	Clinical observations were made 2 X / day
	Once per week close clinical evaluation was made on each
	animal.
	Body weight was measure at the start of the study and weekly
	thereafter.
	Hematology was evaluated in week 4 and 13
	Urinalysis was performed in the males in week 4 and 13.
	Vaginal smears were performed in control and high dose animals in weeks 3 and 4 and weeks 12 and 13.
	Pathology:
	Organ weights of all animals killed at week 5 and 13 of
	adrenals, brain, coagulating glands, epididymis, heart.
	kidney, liver, lung, ovaries, pituitary, prostate, simal
	vesicles, spleen, testes, uterus.
	Testes of animals killed after the 10 week recovery period
	were also weighted.
	Microscopic evaluation:
	Sperm morfology at week 14
	Histological eximination of testis and epididymis(all
	animals), lungs and nasal turbinates (all animals), adrenals
	and pituitary (controls and high dose), ovaries and uterus
	(control and high dose)
	Statistical methods:
	Body weight: analysis of covariance foolowed by Dunnett's
	ltsi organ weights hematological biochemical data: analysis of
	vigan weignis, nematological, biochemical data. allalysis ol variance followed by Dunnett's test
	Incidences of pathological changes: Fisher's exact test.
	Results:
	Compared to controls, no treatment related changes were
	observed in treated animals regarding body weight.

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. 10/10/11/1	<b>Data</b> 12.02.2002
	Date 12.03.2002
	Pathology:
	No treatment related effects were observed on marcoscopic
	examination.
	sperm concentration, number of sperm cells with deformed
	heads/tails or on numbers of isolated heads/tails either in
	weanlings or in young adults.
	NOEC rat 13 weeks, fertility end point: 7000 ppm.
	Reliability criteria: reliable without restrictions:
	comparable to guideline study (1b)
	Demarky Key study for SIDC such star
21 11 2001	Remark. Rey sludy for SIDS evaluation (20)
21.11.2001	(20)
Туре	: Fertility
Species	: rat
Sex	: male/female
Strain	: Sprague-Dawley
Route of admin.	: inhalation
Exposure period	: 0-5 d (pregnancy) 2 wk (mating period)
Promoting exposure period	. 6 fi/d, 7 d/wk (maing period) of 5 d/wk (premaing period)
Male	· 15 wooks
Female	15 weeks
Duration of test	
No. of generation	
studies	
Doses	: 250, 1000, 7000 ppm
Control group	:
NOAEL parental	: 7000 ppm
NOAEL F1 offspring	: 7000 ppm
Method	: other: comparable to modified OECD guideline 422
Year	: 1985
GLP	: yes
lest substance	: as prescribed by 1.1 - 1.4
Remark	The female fertility part of a combined subchronic
	toxicity/fertility/reproduction toxicity study is described here.
	20 animals per sex and per dose group were whole body
	exposed as described above. Monitoring of test atmosphere
	was done by GC.
	Exposed males and females were mated for 2 weeks with
	untreated partners. During mating exposed animals were also
	exposed during 6hrs/d and 7d/wk.
	Pregnant females were exposed during first 5 days of
	Pregnant females were exposed during first 5 days of gestation.
	Pregnant females were exposed during first 5 days of gestation. Pregnant females were sacrificed at day 15 of gestation.
	Pregnant females were exposed during first 5 days of gestation. Pregnant females were sacrificed at day 15 of gestation. Evaluated parameters:
	Pregnant females were exposed during first 5 days of gestation. Pregnant females were sacrificed at day 15 of gestation. Evaluated parameters:
	Pregnant females were exposed during first 5 days of gestation. Pregnant females were sacrificed at day 15 of gestation. Evaluated parameters: Clinical symptoms: Cago side inspection, twice daily:
	Pregnant females were exposed during first 5 days of gestation. Pregnant females were sacrificed at day 15 of gestation. Evaluated parameters: Clinical symptoms: Cage-side inspection, twice daily

1,1-DIFLUOROETHYLENE

OECD SIDS 5. TOXICITY

**Id** 75-38-7 Date 12.03.2002

	Body weights: Body weights were recorded prestudy,once per week and at sacrifice.
	Organ weights: after kill, in treated and untreated animals following organs were weighted: testes epididymes, seminal vesicles, prostate, ovaries.
	Fertility evaluation and reproductive performance: in treated animals and in untreated males and females which were mated with de former following parameters were recorded: mating index, male fertility index, female fertility index, fecundity index, pre-coital time. In treated and untreated females: number of corpora lutea, number of implantation sites, number of early and late resorptions.
	Statistical methods: Body weights of parent rats: analysis of variance Number of females mated, number of pregnant females, number of females bearing live fetusses: Chi-square test Difference in median pre-coital times: Mann-Withney U test mean pre-coital time including only females mated: Student t test organ weights, litter data, pre-implantation loss: analysis of variance followed by Dunnett's Multiple comparison test.
	Results:
	No mortality nor clinical symptoms in any group.
	Body weight of parents similar in all groups
	No treatment related differences in fertility indices or in reproductive performance
	No treatment related autopsy findings, difference in ovary weights and in litter data
	Reproductive organ weights of treated males were similar in all groups.
	NOEC : = or > 7000 ppm
	This is a key study for SIDS evaluation.
	Reliability criteria: Reliable without restriction:
22.08.2001	(40)
8.2 DEVELOPMENTAL TOX	CITY/TERATOGENIC ITY

#### 5.8.2 1 1/11 UG IC.

Species : rat Sex : female Strain : other: CRL:COBS CD (SD) BR
OECD SIDS	1,1-DIFLUOROETHYLENE
5. TOXICITY	ld 75-38-7
	Date 12.03.2002
Route of admin. Exposure period Frequency of treatm. Duration of test Doses Control group NOAEL maternal tox. NOAEL teratogen. Method Year GLP	<ul> <li>inhalation</li> <li>6-15 days of gestation</li> <li>6 h/d</li> <li>2000, 10000 ppm</li> <li>yes</li> <li>= 10000 ppm</li> <li>= 10000 ppm</li> <li>ino</li> <li>no</li> <li>ino</li> <li>ino</li> </ul>
rest substance Remark	<ul> <li>no cata</li> <li>20 females per dose group were mated with males of the same strain.</li> <li>From days 6 through 15 females were whole-body exposed for 6 hrs per day.</li> <li>Test atmosphere concentrations were monitored by GC. At day 20 of gestation feti were delivered by cesarean section.</li> <li>Parameters evaluated:</li> <li>Clinical observations: daily</li> <li>Body weight measured at days 0, 6, 15, 20 of gestation.</li> <li>Food consumption: recorded periods: days 0-6,6-15, 15-20.</li> <li>At day 20: number of implantation sites, corpora lutea, live and dead fetuses, resorption sites</li> <li>Feti: weight, external examination and externally visible abnormalities. Changes in soft tissues of head, thoracic and visceral organs, skeletal abnormalities.</li> <li>Statistical methods:</li> <li>differences in means and near-normal distribution of body weights, food consumption, mean pup weight: Dunnett's t test sex and pregnancy ratio: 2X2 contingency tables with Yates' correction</li> <li>Number of abnormal fetusses within a litter: Wilcoxon Rank Sum.</li> <li>Results:</li> <li>No mortality.</li> <li>Mothers did not show treatment related external or macroscopic features.</li> <li>No changes in body weight or food consumption.</li> <li>No treatment related effects on number of implantation sites, corpora lutea, live and dead fetuses, resorption sites.</li> <li>No treatment related effects on the soft tissues of head or on thoracic and visceral organs.</li> <li>No treatment related effects on the soft tissues of head or on thoracic and visceral organs.</li> <li>No treatment related effects on the fetal skeleton.</li> <li>It was concluded that exposure to 10000 pm VF2 of pregnant rats during gestation days 6 through 15 did not cause an adverse effect on dams nor their progeny.</li> </ul>

**UNEP** Publications

OECD SIDS	1,1-DIFLUOROETHYLENI
5. TOXICITY	ld 75-38-7 Date 12.03.2002
	Reliability criteria:reliable with restrictions: Study well documented, meets generally accepted scientific principles, acceptable for assessment.(2e)
	This is a key study for SIDS evaluation.
22.08.2001	(41)
5.8.3 TOXICITY TO	REPRODUCTION, OTHER STUDIES
5.9 SPECIFIC IN	/ESTIGATIONS
5.10 EXPOSURE I	EXPERIENCE
Remark	<ul> <li>Vinylidene fluoride measurements were made in working atmosphere using a MIRAN apparatus (infra-red). The sensitivity is about 1 ppm and precision is about 0.5 ppm. Background values represent 0.1 ppm. Mean value measured in 1992-1993 in a vinylidene fluoride synthesis pilot plant was 0.9 ppm. During 0.7 % of working time an alarm level of 22.7 ppm was reached. In a laboratory, the general mean was 0.5 ppm with an alarm level mean of 20 ppm representing 0.5 % of total working time. In a polymerization unit the monthly mean ranged from 0.038 ppm to 2.09 ppm. The number of alarm levels above 50 ppm was about 2 per month.</li> </ul>
10.05.1995	(42)
Remark	<ul> <li>Within the frame of an industrial hygiene monitoring programme (organised by the Fluoroalkene industry Group on request of US EPA), personal sampling were performed for various jobs within tw o VF2 polymerization plants in the US.</li> <li>All sampling was performed with a 100 ml "Critical Orifice Personell Sampler" (personal sampling).(flow rate range: 0.06 to 0.08 ml/min ). COPS</li> </ul>
	were worn by different persons with specific jobs for a full shift. VF2 concentrations in the samples were measured with GC. In one plant (Dupont), no exposure data were noted above the detection limit of 1.2 to 1.9 ppm. Jobs included supervisor,
	polymerizing operator, finishing operator, laboratory technician, mechanical jobs. In the second plant (Pennwalt), personal sampling during a workshift (LOD:
	0.01 to 0.03 ppm) indicated the following values (range) - finished product operator: < 0.03 - 30 ppm
	- polymer operator : < 0.01-38.4 ppm - maintenance jobs : < 0.03-0.71 ppm - plant supervisor : < 0.03 ppm - laboratory technician : < 0.01-1.21 ppm
	- field service operator : < 0.01 - 10.3 ppm

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## **UNEP** Publications

OECD SIDS	1.1-DIFLUOROETHYLENE
5. TOXICITY	ld 75-38-7
	Date 12.03.2002
	All values measured above 10 ppm were incidental and related to specific situations: valve relief, plugging of a line.
27.08.2001	Further details on this exposure survey are described in, a confidential report that was submitted to EPA in 1985. (43)

5.11 ADDITIONAL REMARKS

OECD SIDS	1.1-DIFLU	JOROETHYLENE
6. ANALYT. METH. FOR DETECTION AND IDENTIFICATION	ld	75-38-7
	Date	12.03.2002

## 6.1 ANALYTICAL METHODS

## 6.2 DETECTION AND IDENTIFICATION

OEC	D SIDS	1.1-DIFLU	JOROETHYLENE
7. EF	F. AGAINST TARGET ORG. AND INTENDED USES	ld Date	75-38-7 12.03.2002
7.1	FUNCTION		
7.2	EFFECTS ON ORGANISMS TO BE CONTROLLED		
7.3	ORGANISMS TO BE PROTECTED		
7.4	USER		
75	RESISTANCE		

OEC	D SIDS	1,1-DIFLUOROETHYLENE
8. M	EAS. NEC. TO PROT. MAN, ANIMALS, ENVIRONMENT	ld 75-38-7
8.1	METHODS HANDLING AND STORING	Date 12.03.2002
8.2	FIRE GUIDANCE	
8.3	EMERGENCY MEASURES	
8.4	POSSIB. OF RENDERING SUBST. HARMLESS	
8.5	WASTE MANAGEMENT	
8.6	SIDE-EFFECTS DETECTION	
8.7	SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER	2
8.8	REACTIVITY TOWARDS CONTAINER MATERIAL	

OECD	SIDS 1,1-DIFLUOROETHYLENE
9. KEFI	Date 12.03.2002
(4)	http://townat.pml.nih.cov
(1)	nup://toxnet.nmi.nin.gov
(2)	Solvay (1993).
(3)	Dohany, J.E., Robb, L.E. (1980), Fluorine compounds, organic, Kirk-Othmer Encycl. Chem. Tech. 3rd ed., NY, Wiley INTRSCI Publ., 11, 65.
(4)	Yaws, C.L., Yang, H.C., Hopper, J.R., Hansen, K.C. (July 1990), Organic chemicals: water solubility data, Chemical Engineering, 115-118.
(5)	Matheson Gas Data Book, 1966
(6)	Elf Atochem, Safety Data Sheet, 15.02.1995. Reliability criteria: not assignable: documentation insufficient for assessment (4e)
(7)	Elf Atochem, Safety Data Sheet, 15.02.1995.
(8)	Mears, W.H., et al.,Ind. Eng. Chem, vol 47 p 1449-1454, 1955
(9)	Solvay SDS.
(10)	Chou, J.T., Jurs, P.C. (1979), "Computer Assisted Computation of Partition Coefficients from Molecular Structures Using Fragment Constants", J. Chem. Inf. Comput. Sci., 19, 172-78.
(11)	Solvay MSDS
(12)	Syracuse Research Corporation Calculated Values (1988)
(13)	Internal data Solvay, J. Franklin, 2001
(14)	EPIWIN model (EPA, 2001) part: Level III fugacity model.
(15)	EPI WINN, (SRC model), EPA, 2001
(16)	Pennwalt corporation, Unpublished data, Acute inhalation toxicity of vinylidene fluoride, 1982
(17)	Carpenter, C.P., Smyth, H.F., Pozzani, U.C. (1949), The assay of acute vapor toxicity, and the grading and interpretation of results on 96 chemical compounds, J. Ind. Hyg. Toxicol., 31(6), 343-346.
(18)	Lester, D., Greenberg, L.A. (1950), A.M.A. Arch. Ind. Hyg. Occupational Med., 2, 335.
(19)	Burgison, R.M. et al J. Pharmacol.ExpTher., vol 114, p 470-472, 1955
(20)	Reuzel, P.G., Beems, R.B., Dreef-van der Meulen, H.C., Willems, M.I., (1986), Sub-chronic (13- week) inhalation toxicity study of vinylidene fluoride in weanling and young adult rats, CIVO/TNO report no V86.321/250956
(21)	Dupont (1977), US EPA Doc. no 878220582, Fiche no OTS0215306.
(22)	Manus, A.G., Maloney, B.A., Craig, D.K., Keller, J.G. (1984), Thirteen-week subchronic study in F344 rats - Vinylidene fluoride - Final report, LBI Project no 12199-02 (NTP program).
(23)	Manus, A.G., Maloney, B.A., Craig, D.K., Keller, J.G. (1984), Thirteen-week subchronic study in B6C3F1 mice - Vinylidene fluoride - Final report, LBI Project no 12199-03 (NTP program).
(24)	Appelman, L.M., Beems, R.B., Falke, H.E., Dreef-van der Meulen, H.C., Reuzel, P.G. (1985), Sub-chronic (13-week) inhalation toxicity study of vinylidene fluoride in rats, CIVO/TNO report no V85 449/241407

OECD SIDS	1,1-DIFLUOROETHYLENE
9. REFEREN	CES ld 75-38-7 Date 12.03.2002
(25)	Newton, P.E. (1988), A two week inhalation toxicity study of vinylidene fluoride in the mouse, Bio/dynamics report project no 87-8035.
(26)	Newton, P.E. (1989), A thirteen week inhalation toxicity study of vinylidene fluoride in the mouse, Bio/dynamics report project 87-8021.
(27)	Landry, M.M., Fuerst, R. (1968), Gas ecology of bacteria, Chapter 34, 370-80.
(28)	Jagannath, D.R. (1977), Mutagenicity evaluation of Isotron 1132a, Litton Bionetics, LBI project no 20838.
(29)	Matheson, D.W. (1978), Mutagenicity evaluation of Isotron 1132a in the in vitro transformation of BALB/3T3 cells assay, Litton Bionetics, LBI project no 20840.
(30)	Russell, J.F. (1979), Mutagenic activity of ethylene, 1,1-difluoro- in the Salmonella/microsome assay, Haskell Lab. report no 729-78.
(31)	Bartsch, H., Malaveille, C., Barbin, A., Planche, G. (1979), Mutagenic and alkylating metabolites of halo-ethylenes, chlorobutadienes and dichlorobutenes produced by rodent or human liver tissues, Arch. Toxicol., 41, 249-77.
(32)	Rickard, L.B. (1986), Mutagenicity evaluation of vinylidene fluoride in the CHO/HPRT assay, Haskell Lab. report no 601-86.
(33)	Rickard, L.B., Vlachos, D.A. (1986), Evaluation of vinylidene fluoride in the in vitro assay for chromosome aberrations in chinese hamster ovary (CHO) cells, Haskell Lab. report no 606-86.
(34)	Hodson-Walker, G., Mackay, J.M., Cracknell, S., Cowlyn, T., (1988), Vinylidene fluoride: assessment of clastogenic action on bone marrow erythrocytes in the micronucleus test, LSR report 88/0655.
(35)	Sernau, R.C. (1988), Mutagenicity test on vinylidene fluoride (1,1-difluoroethylene) Drosophila Melanogaster sex-linked recessive lethal test, Hazleton HLA study no 10214-0-461.
(36)	Sernau, R.C. (1989), Mutagenicity test on vinylidene fluoride (1,1-difluoroethylene) Drosophila Melanogaster sex-linked recessive lethal test, Hazleton HLA study no 10214-0-461. (Second Revision).
(37)	Arts, J.H., Bos-Kuijpers, M.H., Woutersen, R.A. (1991), Chronic toxicity/carcinogenicity inhalation study of vinylidene fluoride vapour in rats, CIVO/TNO report V91.039.
(38)	Newton, P.E. (1991), An inhalation oncogenicity study of vinylidene fluoride in the mouse, Bio/dynamics report project no 87-8022.
(39)	Maltoni, C., Tovoli, D. (1979), First experimental evidence of the carcinogenic effects of vinylidene fluoride, Med. Lavoro, 5, 363-68.
(40)	Koeter, H.B., van Marwijk, M.W., Reuzel, P.G. (1986), Fertility inhalation toxicity study with
(41)	Mecler, F.J., Beliles, R.P. (1978), Teratology study in rats Isotron 1132a - 1,1-difluoroethylene, Litton Bionetics, LBI project no 20891.
(42)	Solvay data.
(43)	Fluoroalkene Industry Group (1986).