

Survey of Tetrodotoxin Syntheses

Steve Miller

SuperGroup Meeting

06/02/04

Outline

- Introduction and trivia
- Kishi synthesis (1972, racemic)
- Isobe synthesis (2003, asymmetric)
- Du Bois synthesis (2003, asymmetric)
- References

“Certain varieties of puffer fish, especially the *tora fugu*, or tiger puffer (*S. rubripes*), and the closely related *ma fugu*, or common puffer (*S. porphyreus*), are highly prized as comestibles in Japan. The indulgence of the taste is fraught with some peril, since the livers and ovaries of the fish contain a powerful poison.”

R. B. Woodward [“The Structure of Tetrodotoxin” *Pure & Appl. Chem.* **1964**, 9, 49]

Introduction

- First isolated from the ovaries of the puffer fish (Fugu) in 1909.
- Named after the puffer fish family “Tetraodontidae.”
- Responsible for 10-200 deaths per year.



Tetrodotoxin Trivia

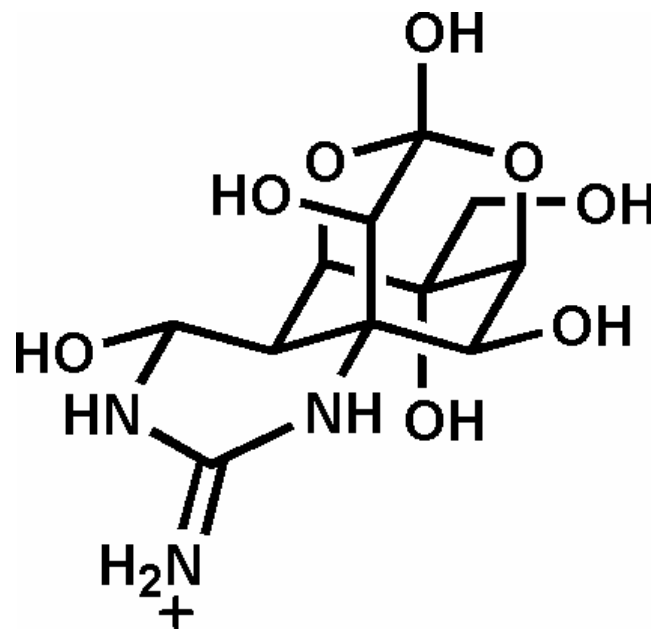
- 1200 times deadlier than cyanide. One fish contains enough poison to kill 30 adults.
- Must be licensed to prepare fugu. Typical meal is \$100-200. Some chefs add a small amount of toxin to the meal for a “tingly” effect.
- Supposedly the only delicacy not served the Japanese emperor.

Trivia Continued

- Toxin is actually generated by the bacteria *Pseudomonas* which the fish consumes
- A potent neurotoxin, it binds to pores of Na-channel proteins in nerve-cell membranes.
- It does not cross the blood-brain barrier, rendering the victim fully conscious but paralyzed. Death is typically from asphyxiation.
- No antidote. Treatment is emptying of the stomach, consumption of activated charcoal, and hoping for the best.

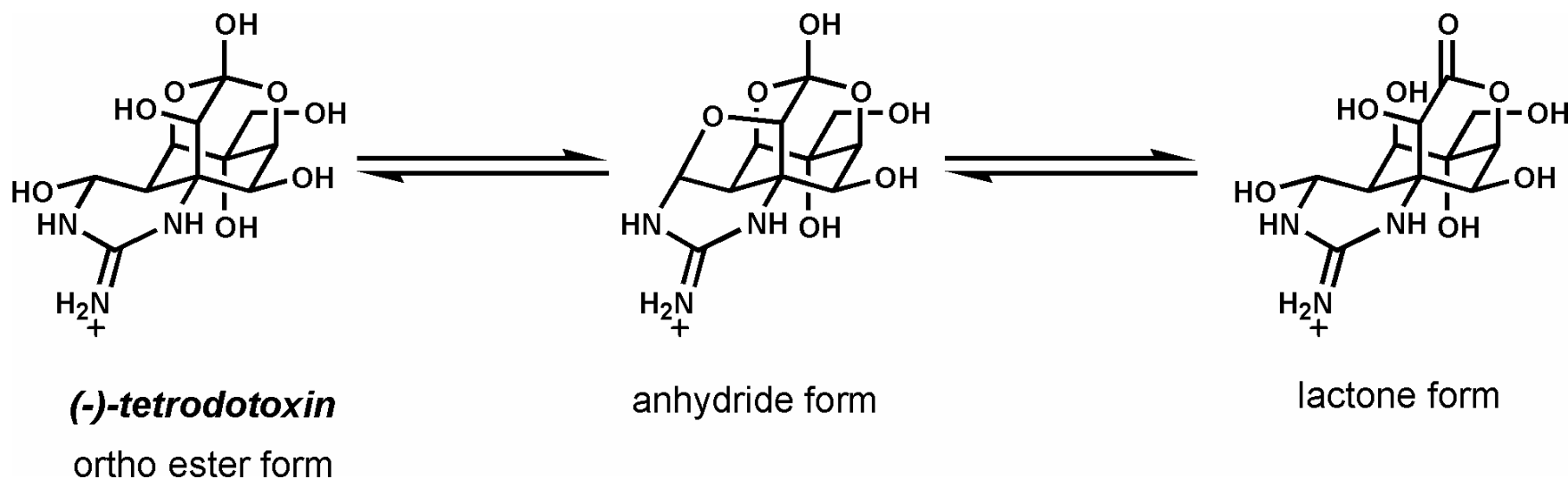
Tetrodotoxin

- Independent structure elucidation by the Hirata-Goto¹, Tsuda², and Woodward³ groups.
- Absolute stereochemistry confirmed by X-Ray in 1970.⁴
- Contains an unprecedented dioxadamantane skeleton functionalized by hydroxyl groups, an ortho ester, and a cyclic guanidine with hemiaminal.

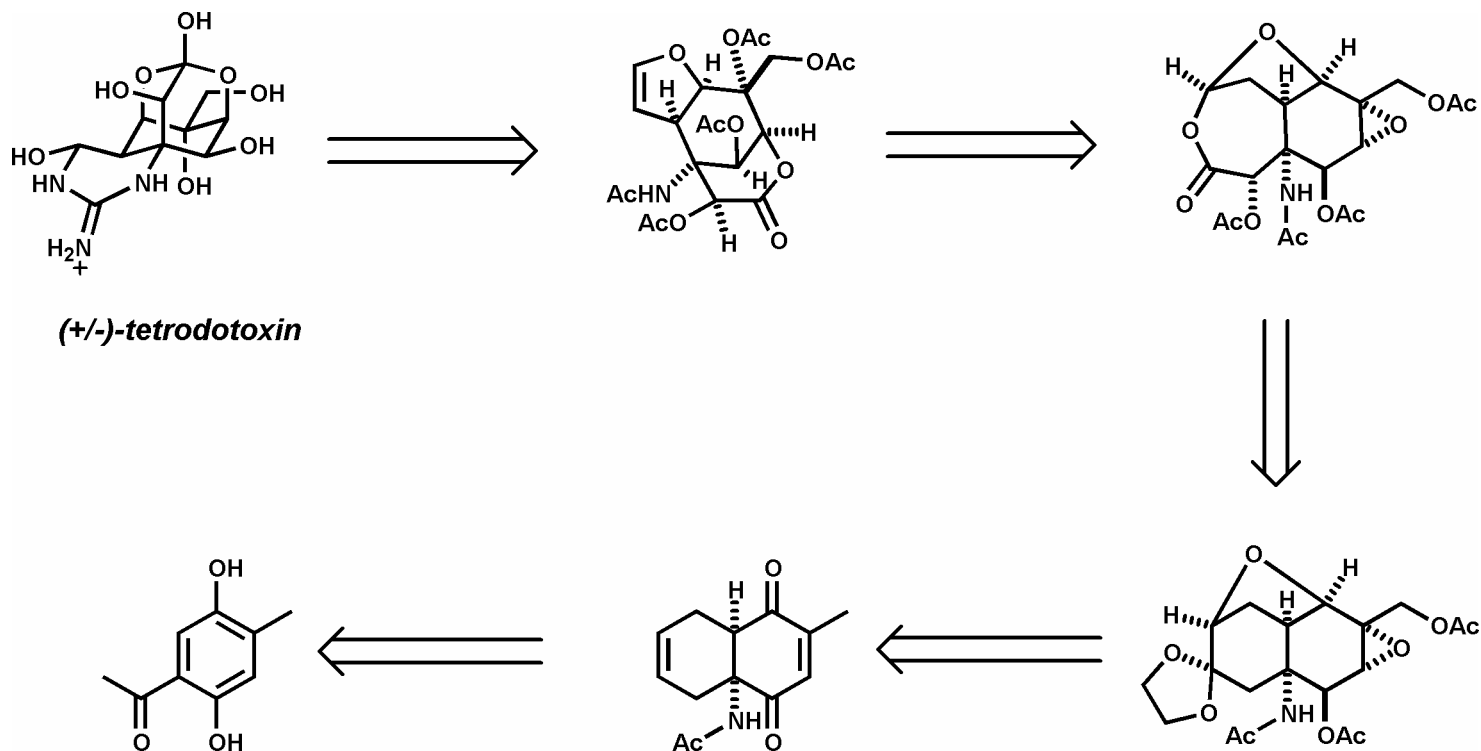


(-)-tetrodotoxin

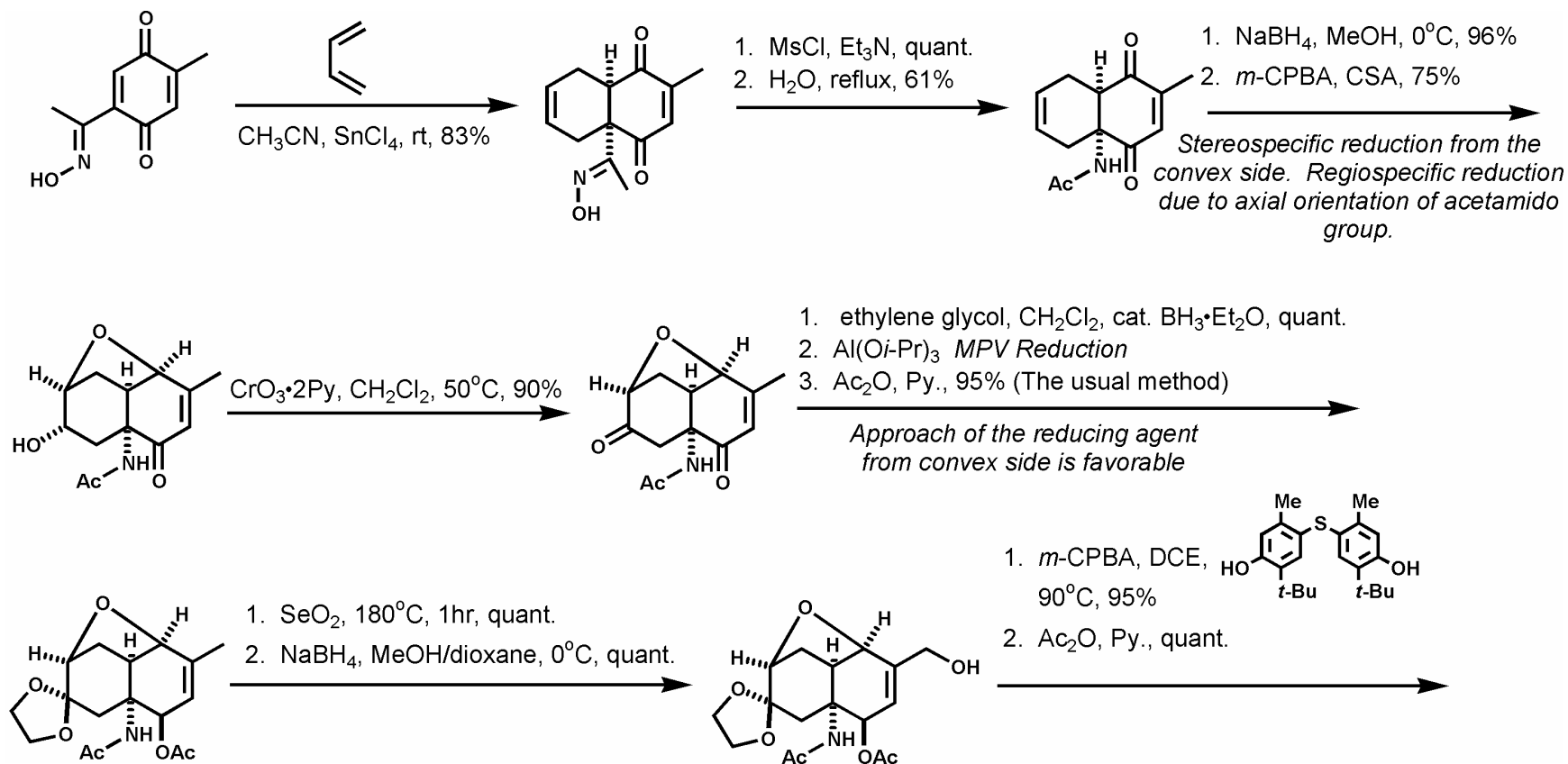
Tetrodotoxin Equilibria



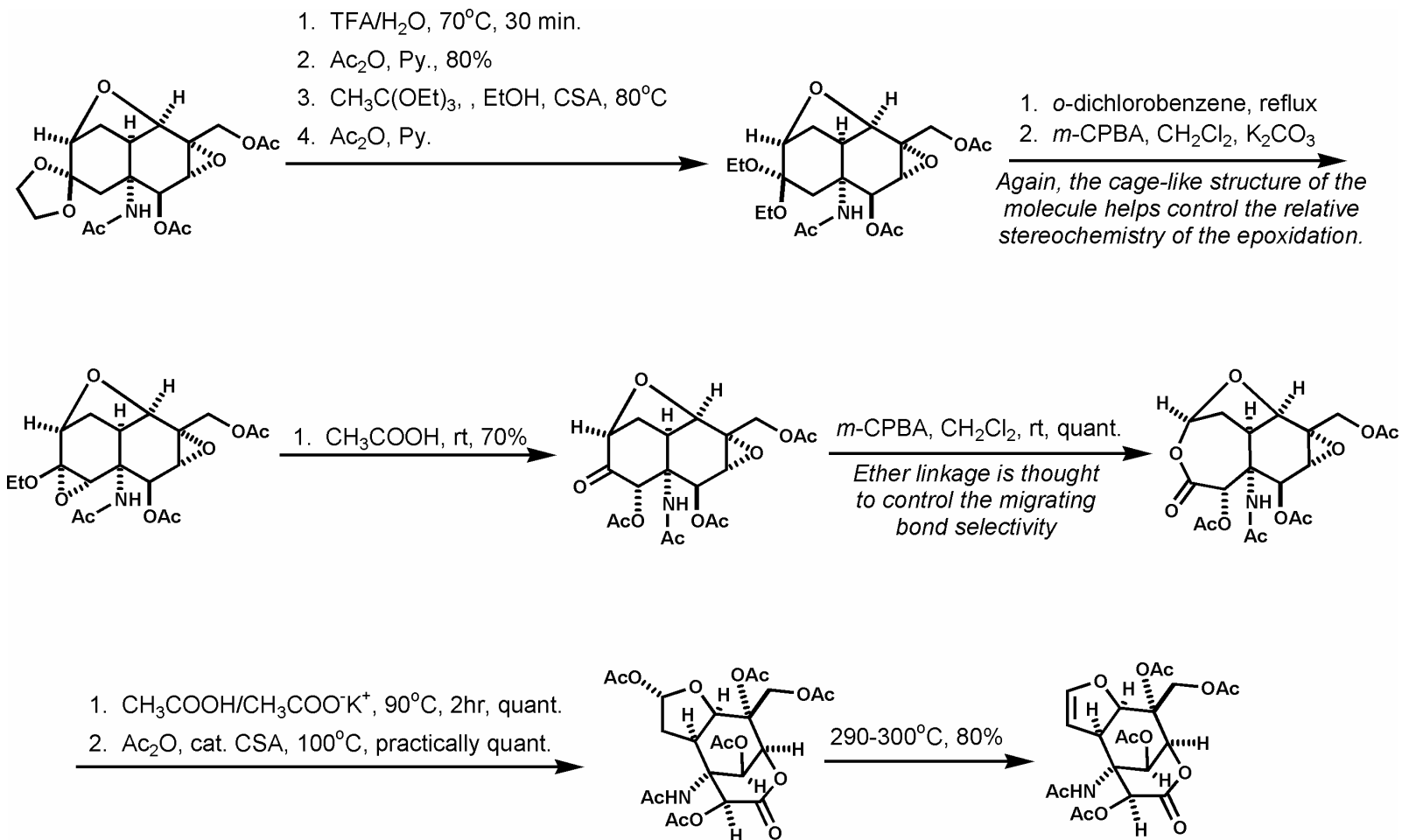
Kishi Retrosynthesis⁵



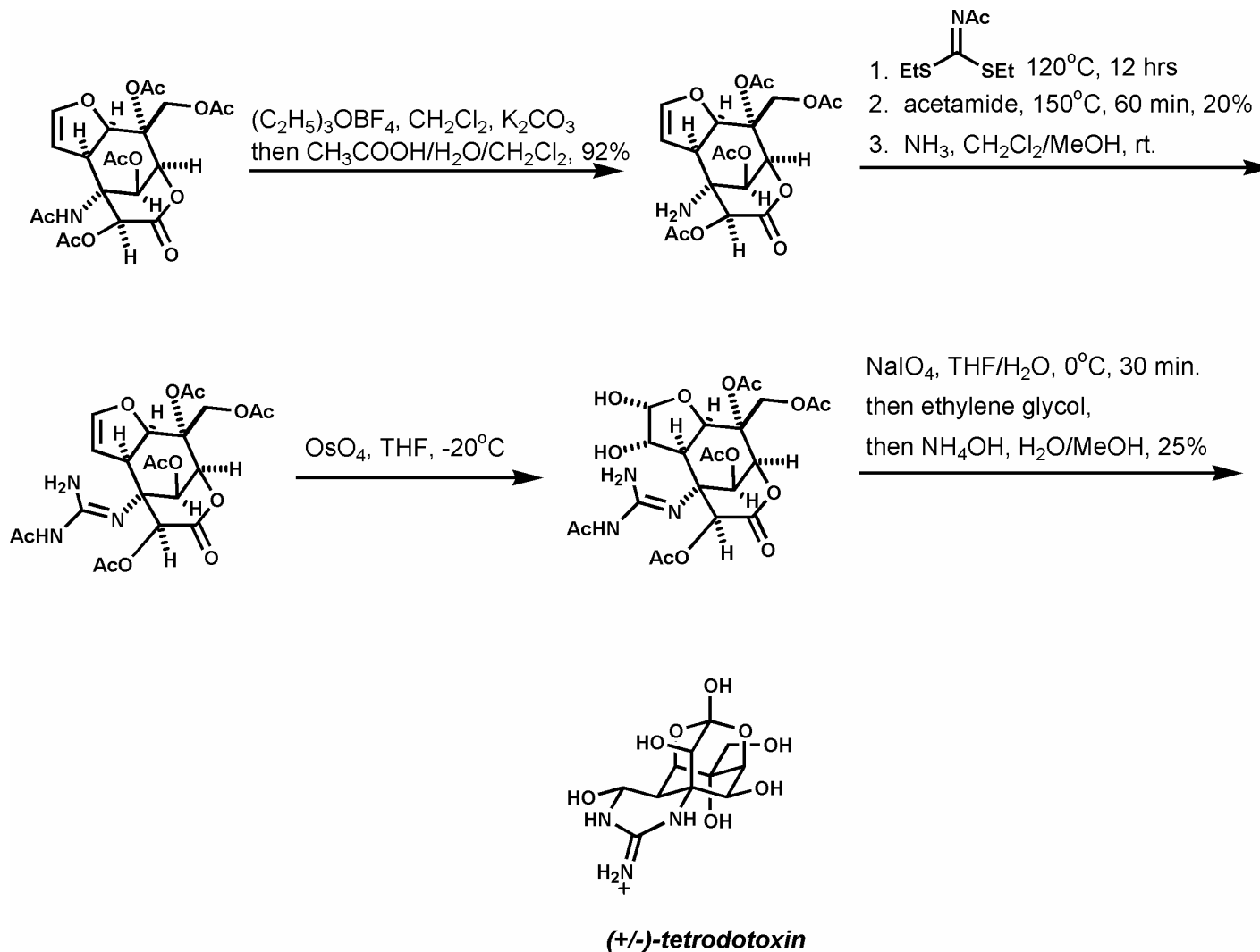
Kishi Synthesis⁵



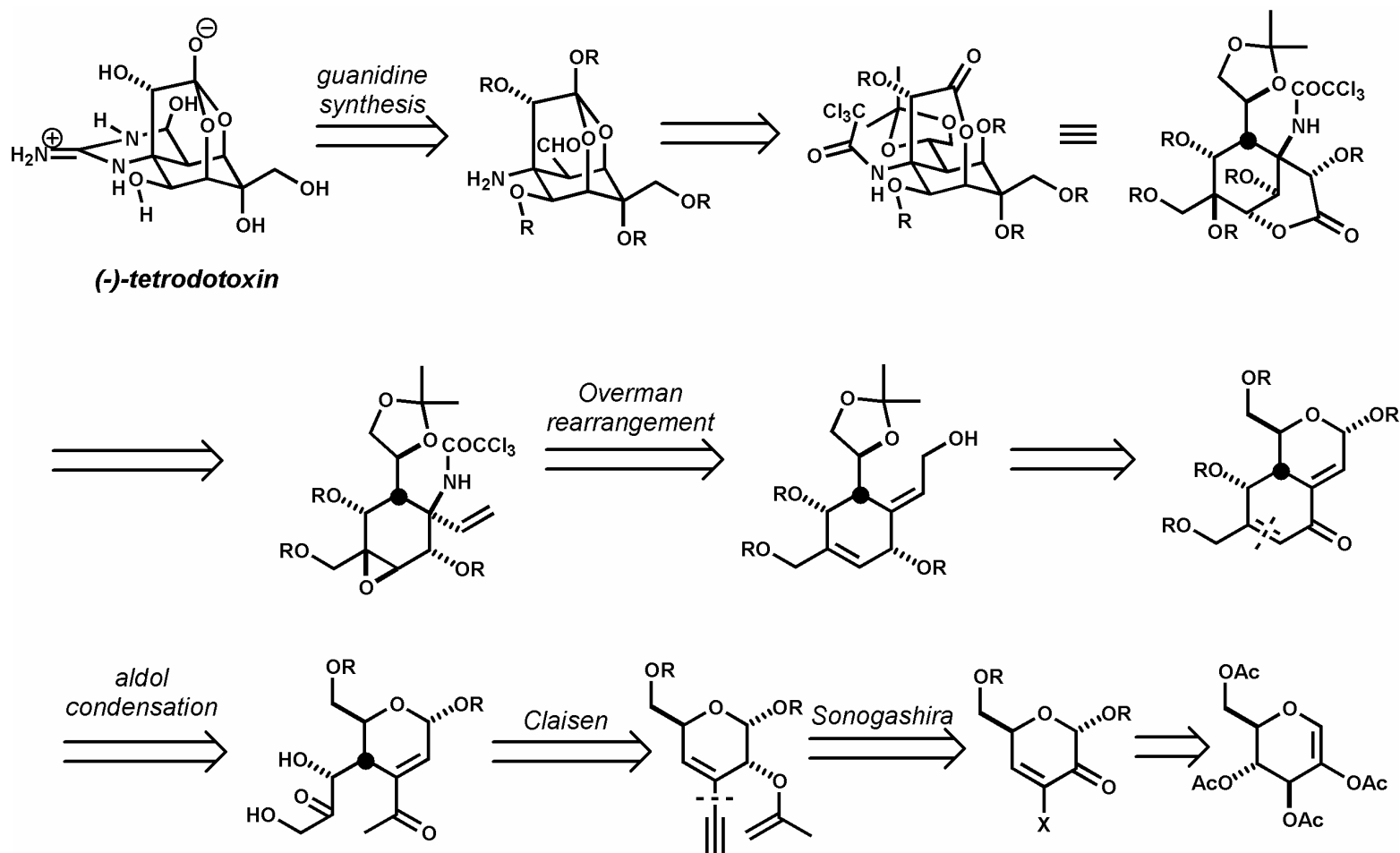
Kishi Synthesis⁵



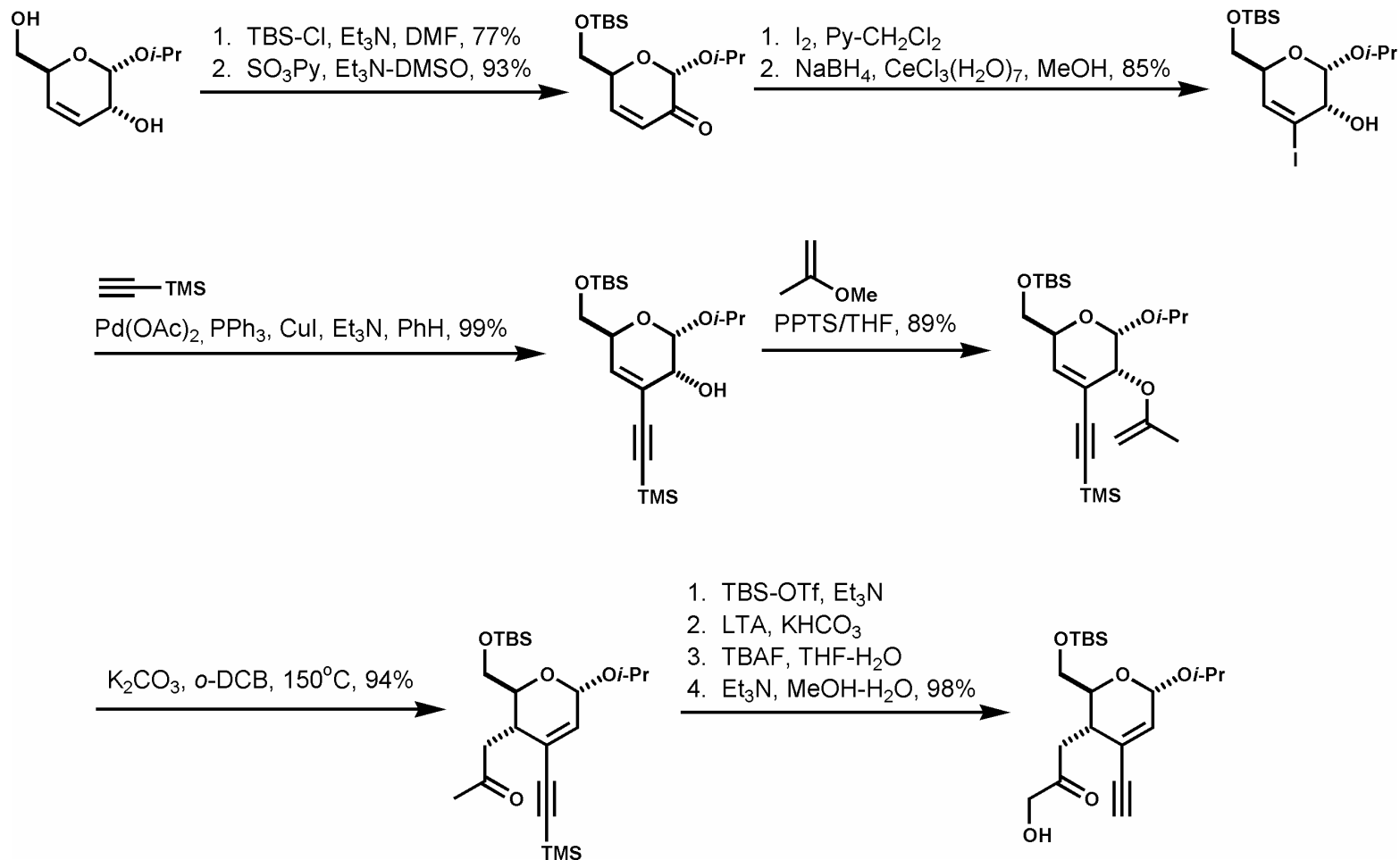
Kishi Synthesis⁵



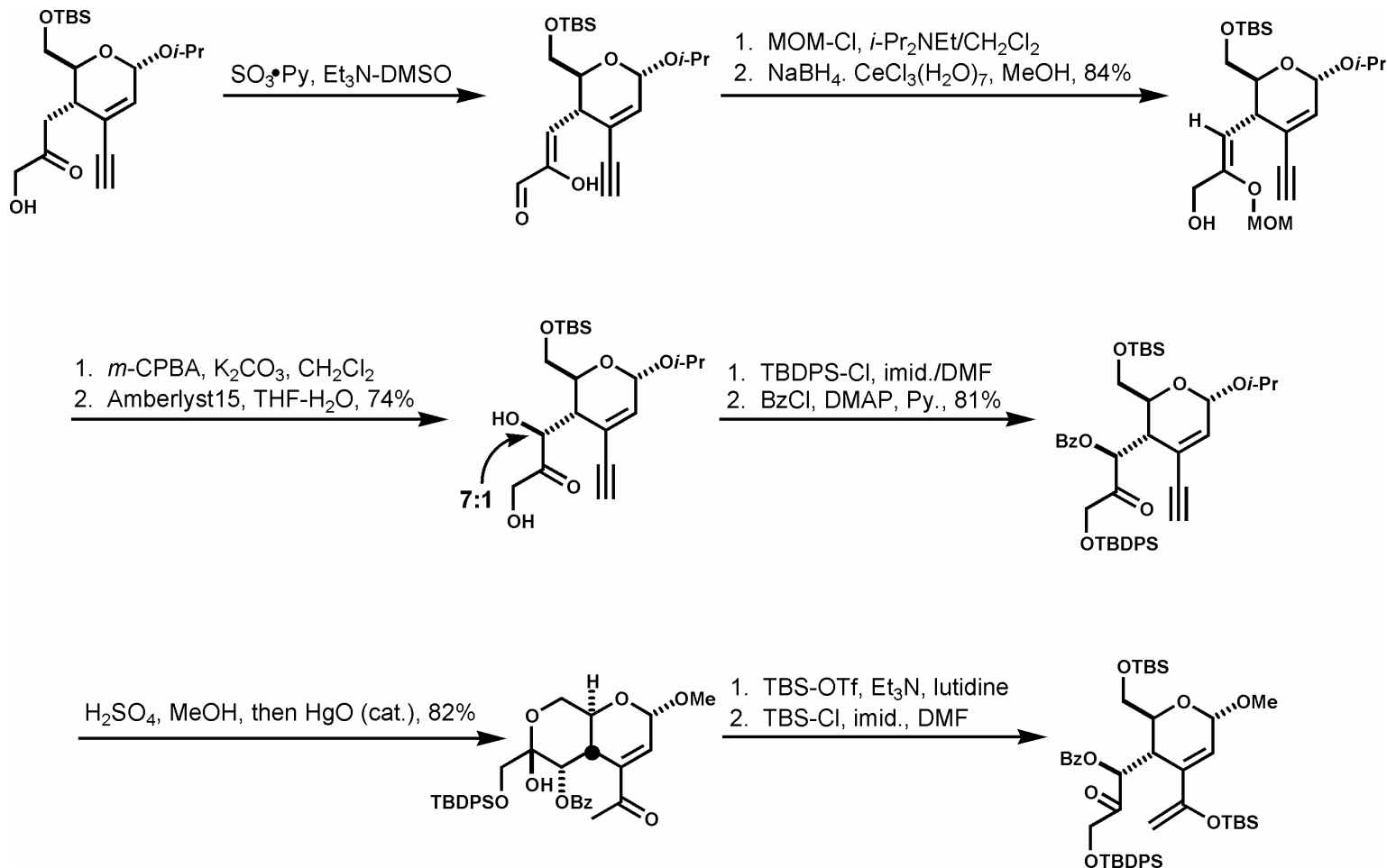
Isobe Retrosynthesis⁶



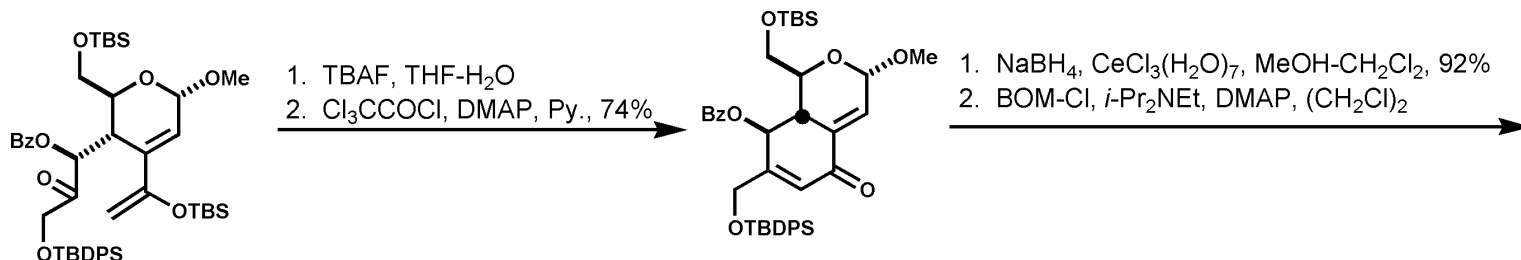
Isobe Synthesis⁶



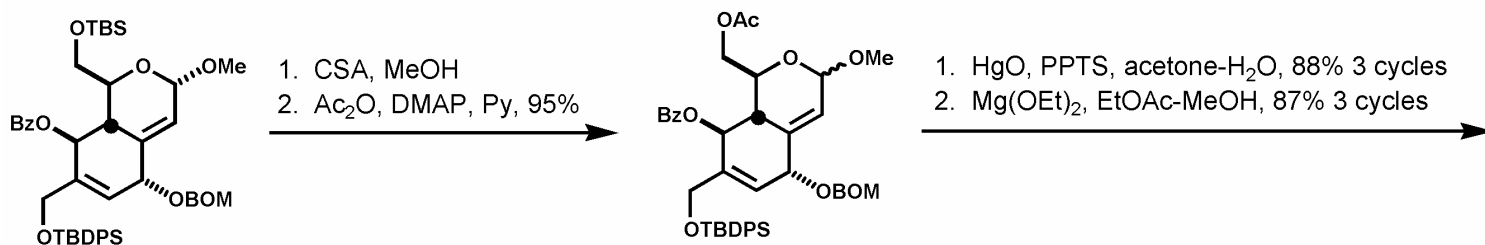
Isobe Synthesis⁶



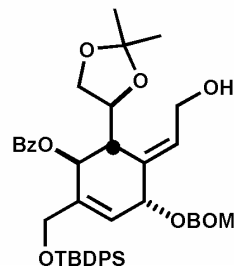
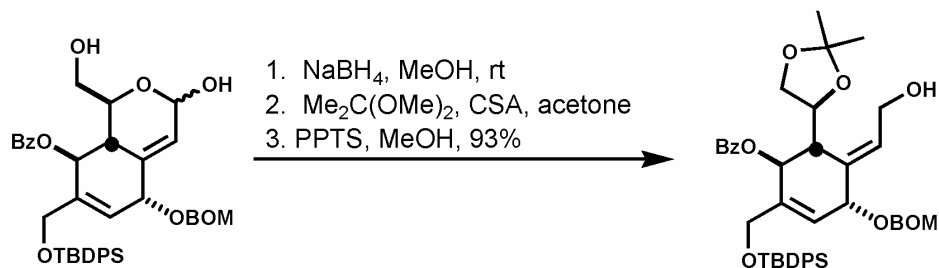
Isobe Synthesis⁶



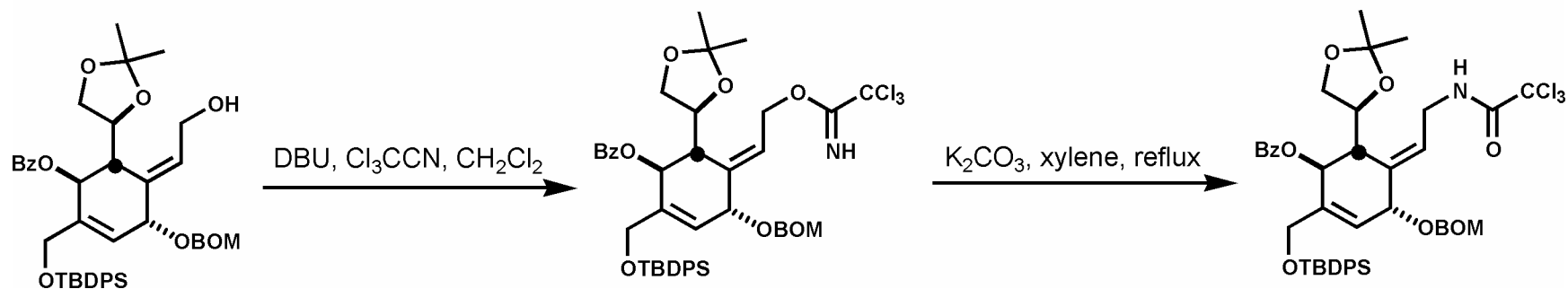
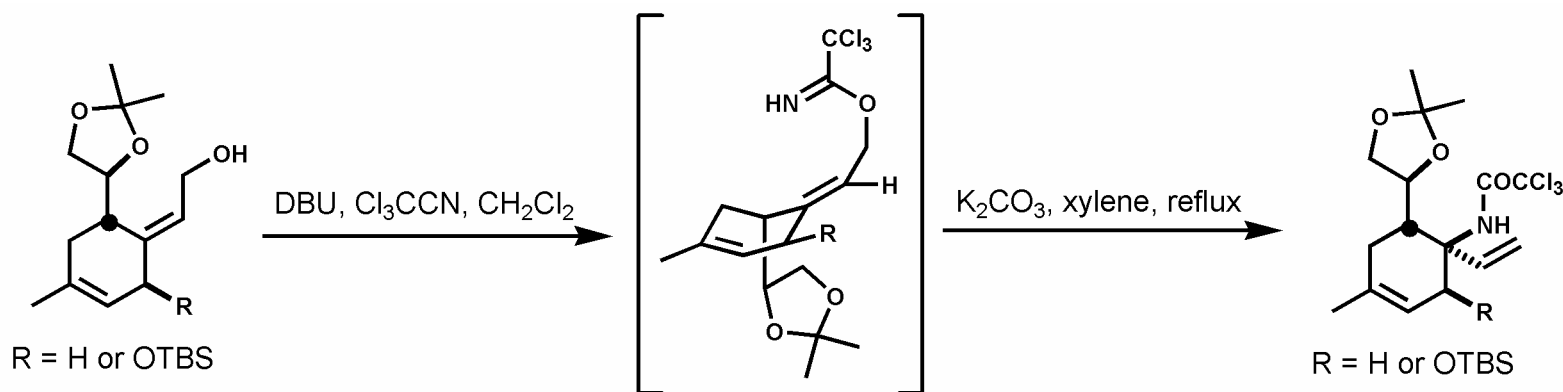
1. NaBH₄, CeCl₃(H₂O)₇, MeOH-CH₂Cl₂, 92%
2. BOM-Cl, *i*-Pr₂NEt, DMAP, (CH₂Cl)₂



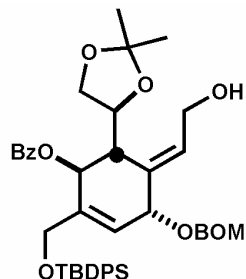
1. HgO, PPTS, acetone-H₂O, 88% 3 cycles
2. Mg(OEt)₂, EtOAc-MeOH, 87% 3 cycles



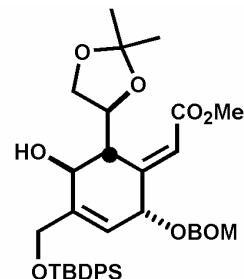
Isobe Synthesis⁶



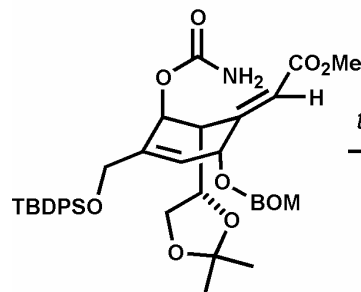
Isobe Synthesis⁶



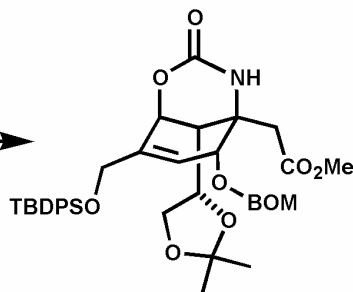
1. DIBAL-H
2. TEMPO, NCS
3. NaClO₂, NaH₂PO₄, Me₂C=CHMe-H₂O
4. TMS-CHN₂, 78%



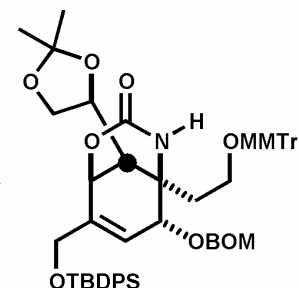
1. Cl₃CCONCO, CH₂Cl₂
2. Et₃N, MeOH, 99%



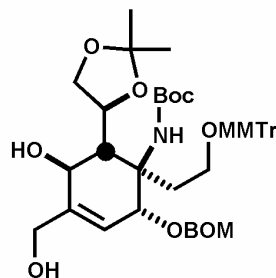
t-BuOK, THF, 90%



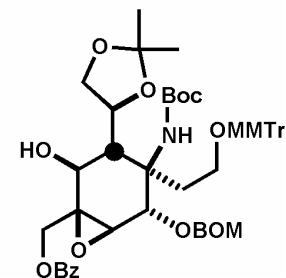
1. LiBH₄, THF
2. MMTr-Cl, Py., 98%



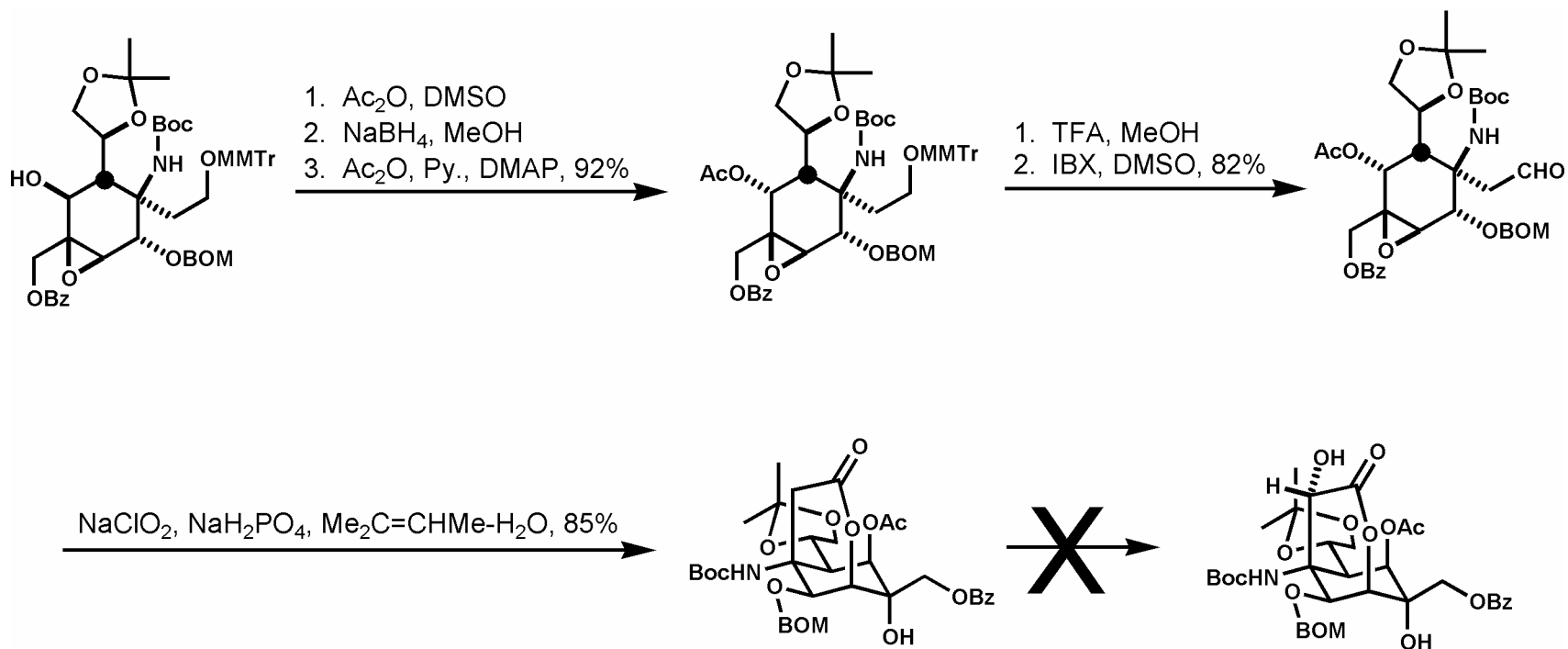
1. Boc₂O, Et₃N, DMAP, THF
2. LiOH, MeOH, (CH₂Cl)₂-H₂O, 84%



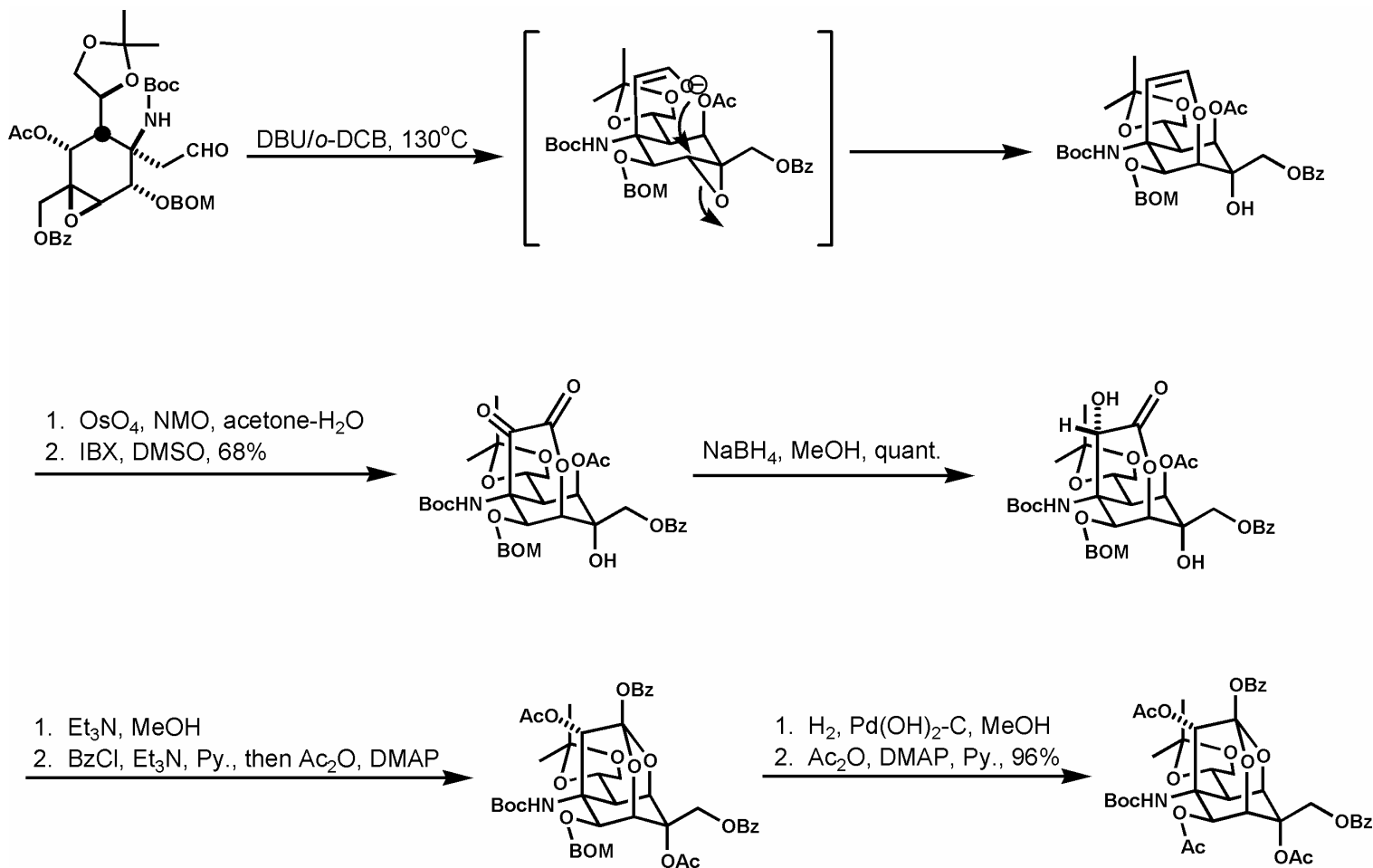
1. *m*-CPBA, Na₂HPO₄, (CHCl₂)₂
2. BzCl, Et₃N, CH₂Cl₂, 93%



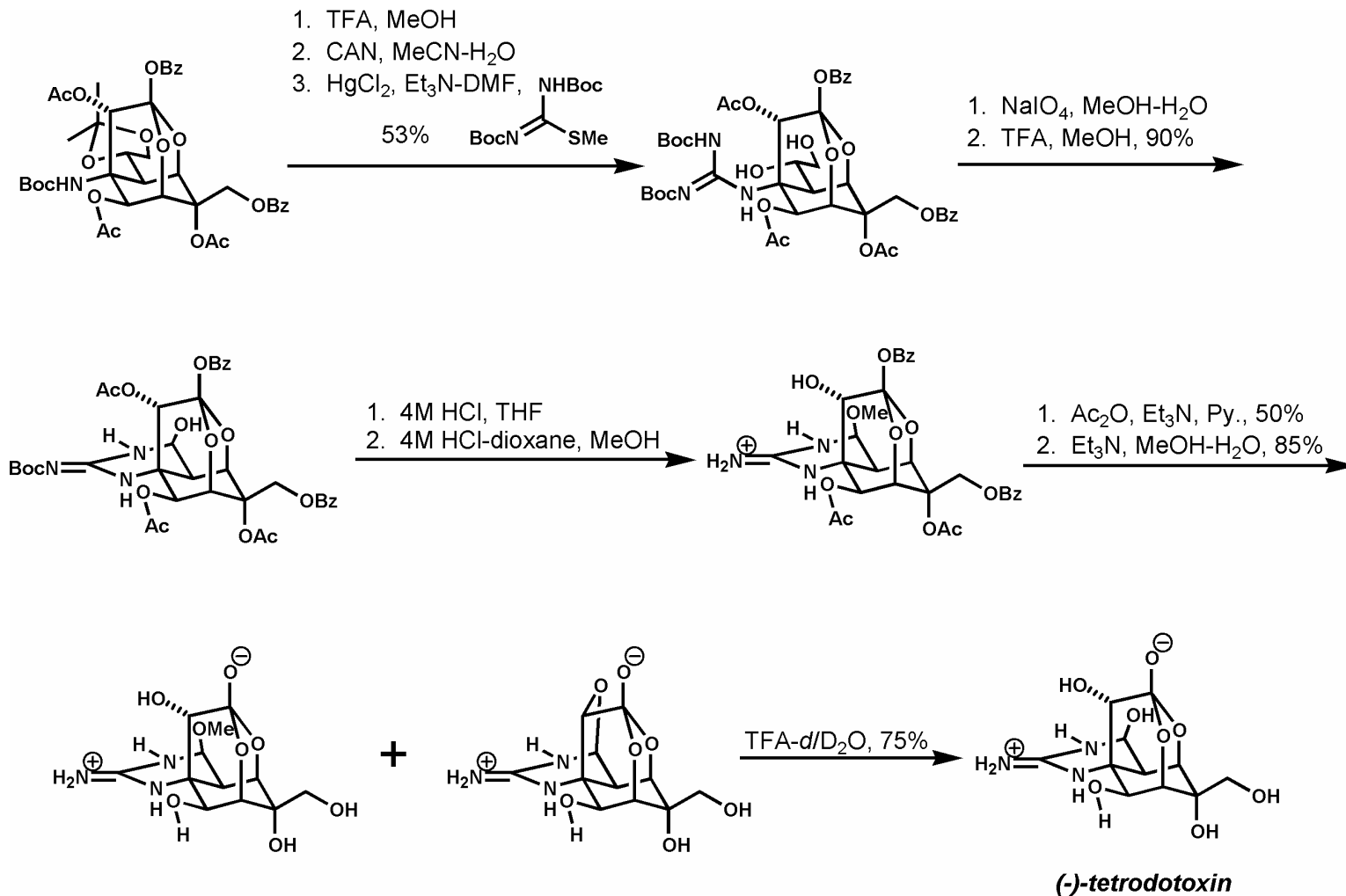
Isobe Synthesis⁶



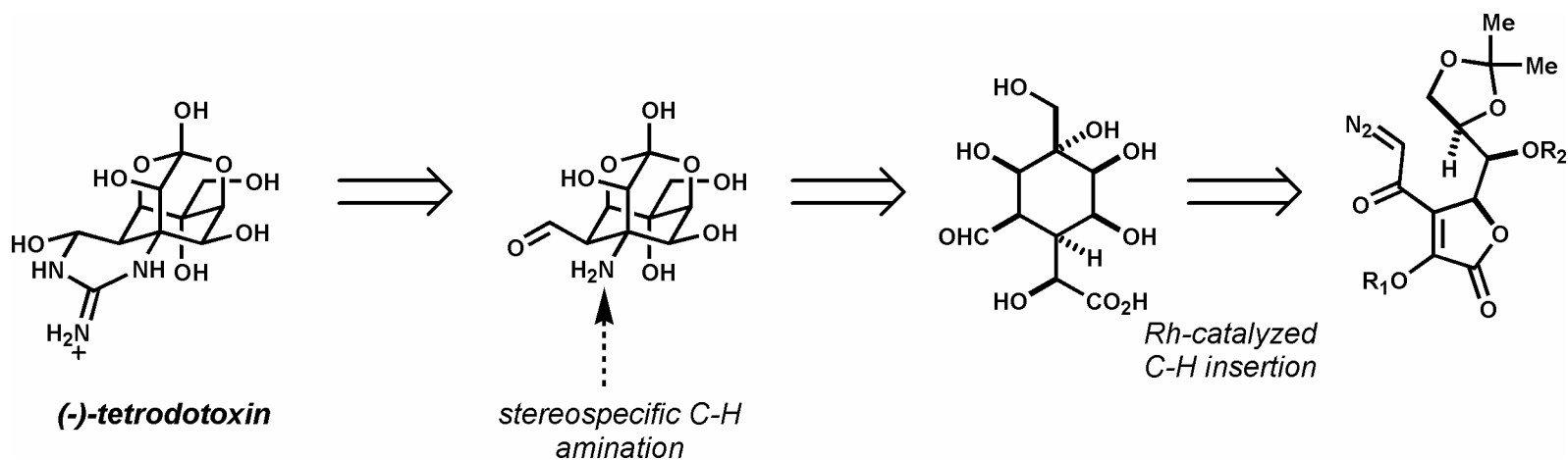
Isobe Synthesis⁶



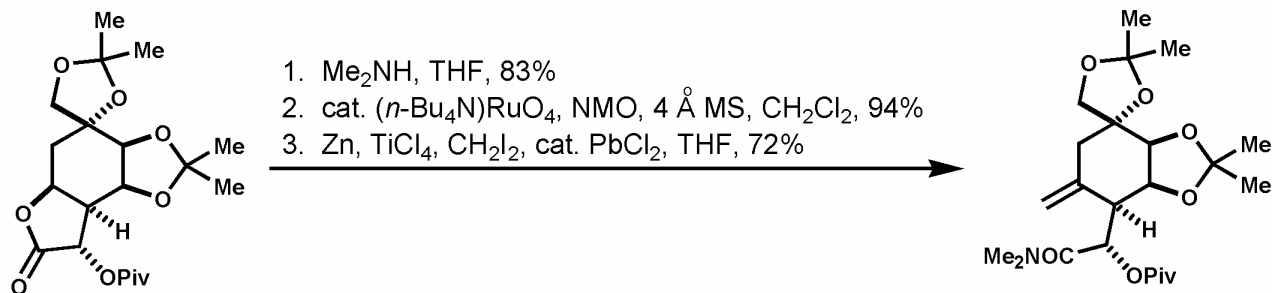
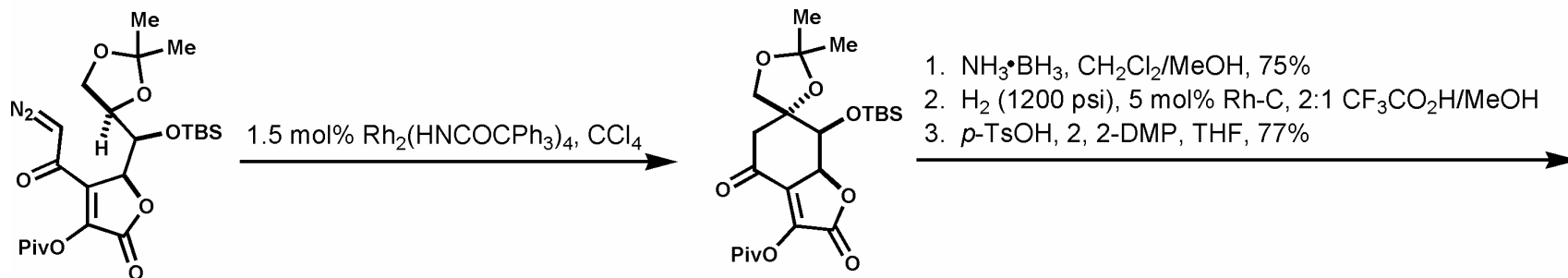
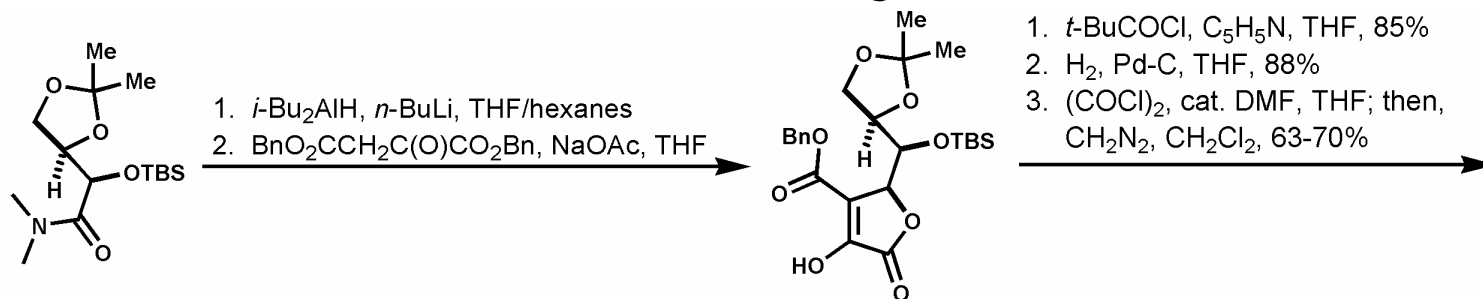
Isobe Synthesis⁶



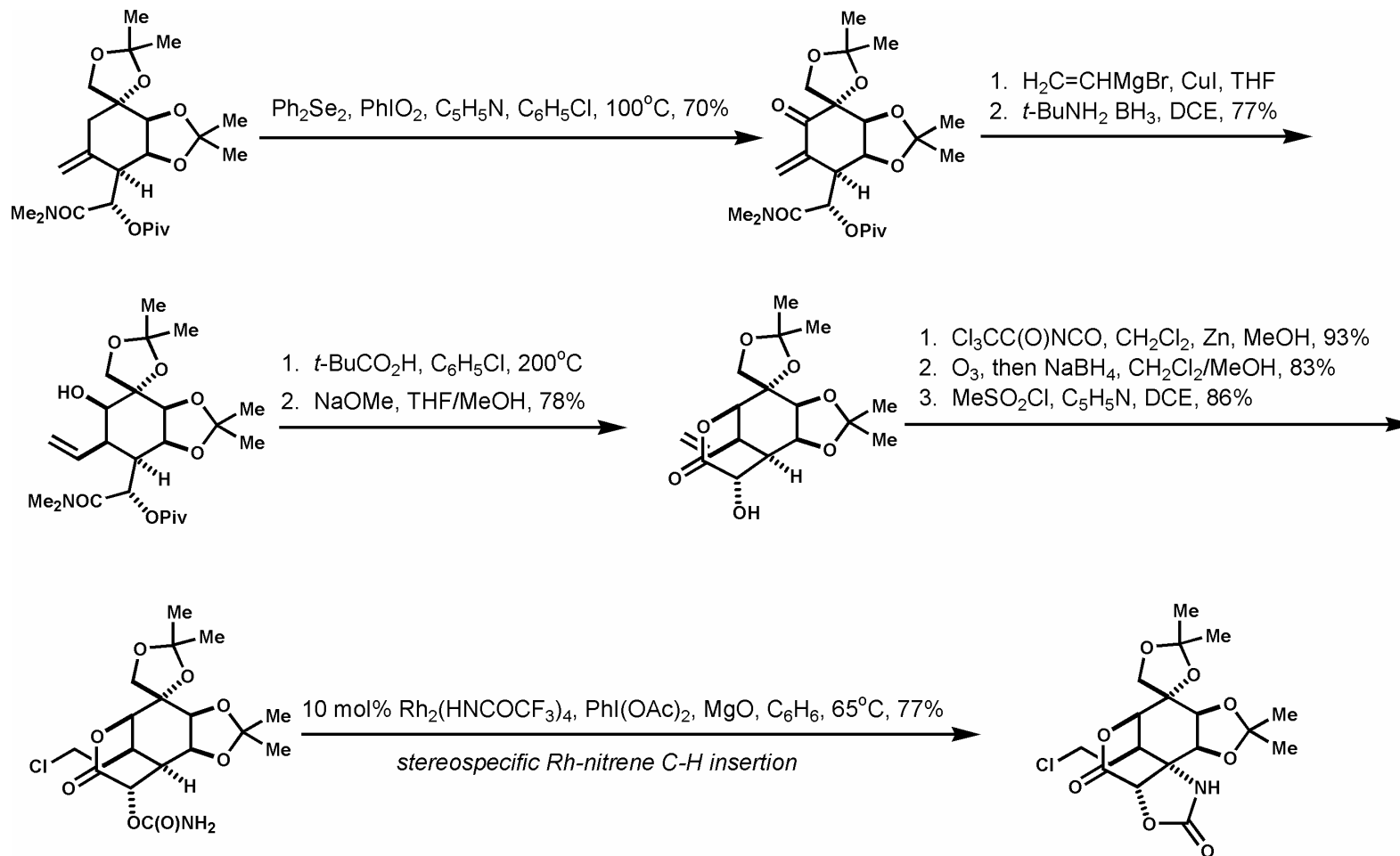
Du Bois Retrosynthesis⁷



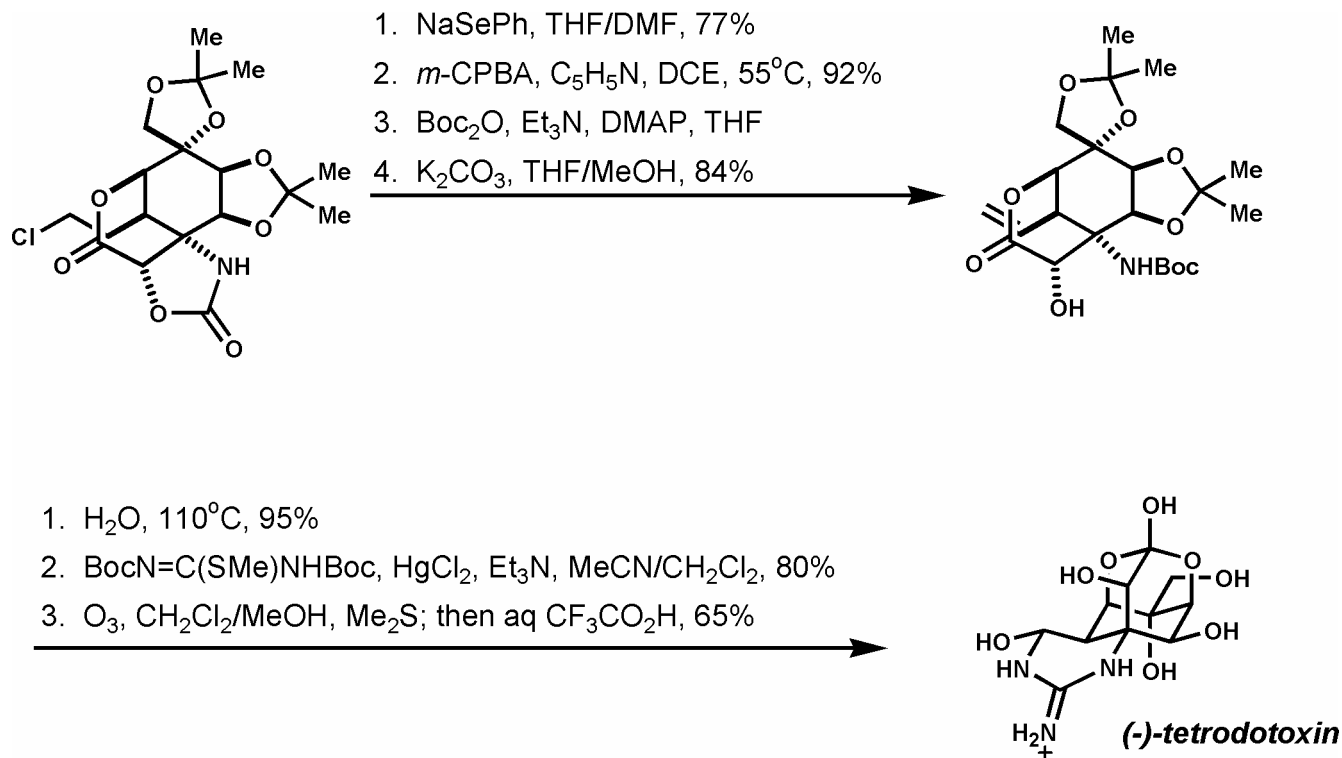
Du Bois Synthesis⁷



Du Bois Synthesis⁷



Du Bois Synthesis⁷



References

1. (a) Goto, T.; Kishi, Y.; Takahashi, S.; Hirata, Y. *Tetrahedron* **1965**, 21, 2059-2088.
2. Tsuda, K.; Ikuma, S.; Kawamura, M.; Tachikawa, R.; Sakai, K.; Tamura, C.; Amakasu, O. *Chem. Pharm. Bull.* **1964**, 12, 1357-1374.
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5. (a) Kishi, Y.; Aratani, M.; Fukuyama, T.; Nakatsubo, F.; Goto, T.; Inoue, S.; Tanino, H.; Sugiura, S.; Kakoi, H. *J. Am. Chem. Soc.* **1972**, 94, 9217-9219.
(b) Kishi, Y.; Fukuyama, T.; Aratani, M.; Nakatsubo, F.; Goto, T.; Inoue, S.; Tanino, H.; Sugiura, S.; Kakoi, H. *J. Am. Chem. Soc.* **1972**, 94, 9219-9221.
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